

# Cesium Tungsten Bronze Encyclopedia

中钨智造科技有限公司

CTIA GROUP LTD

CTIA GROUP LTD

Global Leader in Intelligent Manufacturing for Tungsten, Molybdenum, and Rare Earth Industries

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## INTRODUCTION TO CTIA GROUP

CTIA GROUP LTD, a wholly-owned subsidiary with independent legal personality established by CHINATUNGSTEN ONLINE, is dedicated to promoting the intelligent, integrated, and flexible design and manufacturing of tungsten and molybdenum materials in the Industrial Internet era. CHINATUNGSTEN ONLINE, founded in 1997 with [www.chinatungsten.com](http://www.chinatungsten.com) as its starting point—China's first top-tier tungsten products website—is the country's pioneering e-commerce company focusing on the tungsten, molybdenum, and rare earth industries. Leveraging nearly three decades of deep experience in the tungsten and molybdenum fields, CTIA GROUP inherits its parent company's exceptional design and manufacturing capabilities, superior services, and global business reputation, becoming a comprehensive application solution provider in the fields of tungsten chemicals, tungsten metals, cemented carbides, high-density alloys, molybdenum, and molybdenum alloys.

Over the past 30 years, CHINATUNGSTEN ONLINE has established more than 200 multilingual tungsten and molybdenum professional websites covering more than 20 languages, with over one million pages of news, prices, and market analysis related to tungsten, molybdenum, and rare earths. Since 2013, its WeChat official account "CHINATUNGSTEN ONLINE" has published over 40,000 pieces of information, serving nearly 100,000 followers and providing free information daily to hundreds of thousands of industry professionals worldwide. With cumulative visits to its website cluster and official account reaching billions of times, it has become a recognized global and authoritative information hub for the tungsten, molybdenum, and rare earth industries, providing 24/7 multilingual news, product performance, market prices, and market trend services.

Building on the technology and experience of CHINATUNGSTEN ONLINE, CTIA GROUP focuses on meeting the personalized needs of customers. Utilizing AI technology, it collaboratively designs and produces tungsten and molybdenum products with specific chemical compositions and physical properties (such as particle size, density, hardness, strength, dimensions, and tolerances) with customers. It offers full-process integrated services ranging from mold opening, trial production, to finishing, packaging, and logistics. Over the past 30 years, CHINATUNGSTEN ONLINE has provided R&D, design, and production services for over 500,000 types of tungsten and molybdenum products to more than 130,000 customers worldwide, laying the foundation for customized, flexible, and intelligent manufacturing. Relying on this foundation, CTIA GROUP further deepens the intelligent manufacturing and integrated innovation of tungsten and molybdenum materials in the Industrial Internet era.

Dr. Hanns and his team at CTIA GROUP, based on their more than 30 years of industry experience, have also written and publicly released knowledge, technology, tungsten price and market trend analysis related to tungsten, molybdenum, and rare earths, freely sharing it with the tungsten industry. Dr. Han, with over 30 years of experience since the 1990s in the e-commerce and international trade of tungsten and molybdenum products, as well as the design and manufacturing of cemented carbides and high-density alloys, is a renowned expert in tungsten and molybdenum products both domestically and internationally. Adhering to the principle of providing professional and high-quality information to the industry, CTIA GROUP's team continuously writes technical research papers, articles, and industry reports based on production practice and market customer needs, winning widespread praise in the industry. These achievements provide solid support for CTIA GROUP's technological innovation, product promotion, and industry exchanges, propelling it to become a leader in global tungsten and molybdenum product manufacturing and information services.



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Cesium Tungsten Bronze Product Introduction

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1. Cesium Tungsten Bronze Overview

Cesium Tungsten Bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) produced by CTIA GROUP is manufactured using advanced solvothermal and chemical vapor deposition processes, ensuring high purity and excellent optoelectronic performance. Cesium tungsten bronze is a nano-functional material widely applied in energy-saving glass, optoelectronics, catalysis, batteries, and biomedical fields due to its outstanding near-infrared (NIR) shielding performance, high visible light transmittance, and electrochemical stability. Its unique tungsten-oxygen-cesium crystal structure makes it the preferred material for smart materials and new energy applications.

2. Cesium Tungsten Bronze Features

- Chemical composition:  $\text{Cs}_x\text{WO}_3$  ( $x = 0.2\text{--}0.33$ ), purity  $\geq 99.9\%$ , extremely low impurities.
- Appearance: Deep blue nanopowder or thin film; cubic or hexagonal crystal structure.
- Near-infrared shielding: NIR shielding rate  $> 90\%$  (800–2500 nm), suitable for energy-saving glass.
- Visible light transmittance: Transmittance  $> 70\%$  (400–700 nm), supporting smart window applications.
- Electrical conductivity:  $\sim 10^3$  S/cm, ideal for optoelectronic sensors and battery electrodes.
- Chemical stability: Corrosion rate  $< 0.002$  mm/year, acid and alkali resistant, suitable for catalytic environments.
- Versatility: Supports electrochromic, photothermal conversion, and biocompatible coatings.

3. Cesium Tungsten Bronze Product Specifications

Type	Particle Size (nm)	Purity (wt%)	Bulk Density (g/cm <sup>3</sup> )	Cesium Content (wt%)	Impurities (wt%, Max)
Nano-grade	30–50	$\geq 99.9$	2.5–3.0	5.0–8.0	Fe $\leq 0.002$ , Si $\leq 0.001$ , O $\leq 0.05$

4. Cesium Tungsten Bronze Packaging and Quality Assurance

Packaging: Sealed stainless steel cans or vacuum aluminum foil bags, net weight of 100 g, 500 g, or 1 kg, ensuring moisture-proof and anti-oxidation storage.

Quality assurance: Each batch is accompanied by a quality certificate containing the following test data:

- Purity (ICP-MS,  $> 99.9\%$ )
- Particle size distribution (laser diffraction)
- Crystal structure (XRD)
- Cesium content (chemical titration)
- Surface morphology (SEM)

5. Cesium Tungsten Bronze Procurement Information

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Website: For more information about cesium tungsten bronze, please visit China Tungsten Online (<http://www.cesium-tungsten-bronze.com/>).

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## Table of contents

### Chapter 1: Introduction and History of Cesium Tungsten Bronze

- 1.1 Definition and chemical composition of cesium tungsten bronze
- 1.2 Discovery and development of cesium tungsten bronze
- 1.3 Status of cesium tungsten bronze in materials science
- 1.4 Global research status and market overview of cesium tungsten bronze
- 1.5 Key application areas of cesium tungsten bronze

### Chapter 2: Crystal structure and properties of cesium tungsten bronze

- 2.1 Crystal structure and chemical bond characteristics of cesium tungsten bronze
- 2.2 Optical properties of cesium tungsten bronze: near-infrared absorption and light transmittance
- 2.3 Electrical properties of cesium tungsten bronze: conductivity and carrier migration
- 2.4 Thermal properties of cesium tungsten bronze: thermal conductivity and stability
- 2.5 Theoretical calculation and performance prediction of cesium tungsten bronze

### Chapter 3: Synthesis Method of Cesium Tungsten Bronze

- 3.1 Solid-state reaction method of cesium tungsten bronze
- 3.2 Solvothermal and hydrothermal methods of cesium tungsten bronze
- 3.3 Chemical vapor deposition (CVD) of cesium tungsten bronze
- 3.4 Sol-gel method of cesium tungsten bronze
- 3.5 Green synthesis and nanoparticle control of cesium tungsten bronze

### Chapter 4: Characterization Technology of Cesium Tungsten Bronze

- 4.1 X-ray diffraction (XRD) and crystal analysis of cesium tungsten bronze
- 4.2 Scanning electron microscopy (SEM) and transmission electron microscopy (TEM) of cesium tungsten bronze
- 4.3 X-ray photoelectron spectroscopy (XPS) and chemical state of cesium tungsten bronze
- 4.4 UV-Vis-NIR spectroscopy of cesium tungsten bronze
- 4.5 Electrical and thermal test methods of cesium tungsten bronze

### Chapter 5: Optical and Thermal Applications of Cesium Tungsten Bronze

- 5.1 Cesium tungsten bronze smart window film and energy-saving glass
- 5.2 Cesium tungsten bronze near-infrared shielding coating
- 5.3 Cesium tungsten bronze light-heat conversion and solar energy utilization
- 5.4 Cesium tungsten bronze optical sensors and detectors
- 5.5 Cesium tungsten bronze thermal management materials

### Chapter 6: Energy and Environmental Applications of Cesium Tungsten Bronze

- 6.1 Lithium-ion batteries and supercapacitors of cesium tungsten bronze
- 6.2 Photocatalysis and water decomposition of cesium tungsten bronze
- 6.3 Air purification and pollutant adsorption of cesium tungsten bronze

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- 6.4 Electrode materials for fuel cells of cesium tungsten bronze
- 6.5 Hydrogen storage and energy storage of cesium tungsten bronze

## **Chapter 7: Industrial Production of Cesium Tungsten Bronze**

- 7.1 Production process and equipment of cesium tungsten bronze
- 7.2 Raw material supply chain and cost analysis of cesium tungsten bronze
- 7.3 Large-scale production technology of cesium tungsten bronze
- 7.4 Quality control and testing of cesium tungsten bronze
- 7.5 Market application cases of cesium tungsten bronze

## **Chapter 8: Standards and Regulations for Cesium Tungsten Bronze**

- 8.1 International and national standards for cesium tungsten bronze (ISO, GB/T)
- 8.2 Environmental and safety regulations for cesium tungsten bronze (REACH, RoHS)
- 8.3 Nanomaterial risk assessment for cesium tungsten bronze
- 8.4 Occupational health and safety requirements for cesium tungsten bronze
- 8.5 Product certification and compliance for cesium tungsten bronze
- 8.6 CTIA GROUP LTD Cesium Tungsten Bronze MSDS

## **Chapter 9: Sustainability and Environmental Impact of Cesium Tungsten Bronze**

- 9.1 Environmental impact assessment of cesium tungsten bronze production process
- 9.2 Green manufacturing technology of cesium tungsten bronze
- 9.3 Waste treatment and recycling of cesium tungsten bronze
- 9.4 Carbon footprint and emission reduction strategy of cesium tungsten bronze
- 9.5 Policy drivers for sustainable development of cesium tungsten bronze

## **Chapter 10: Future Research and Prospects of Cesium Tungsten Bronze**

- 10.1 Exploration of new synthesis methods for cesium tungsten bronze
- 10.2 Potential for next-generation applications of cesium tungsten bronze
- 10.3 Integration of intelligent and digital technologies for cesium tungsten bronze
- 10.4 Global cooperation and technical challenges for cesium tungsten bronze
- 10.5 Future development trends and suggestions for cesium tungsten bronze

## **Appendix**

- Appendix 1: Cesium Tungsten Bronze Terms and Abbreviations
- Appendix 2: Cesium Tungsten Bronze References
- Appendix 3: Cesium Tungsten Bronze Data Sheet

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## Chapter 1: Introduction and History of Cesium Tungsten Bronze

Cesium Tungsten Bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) is a functional nanomaterial that has great potential in energy conservation, environmental protection, electronics and energy due to its excellent near-infrared absorption ( $\sim 70\%$  at  $1000\text{ nm}$ ), high conductivity ( $\sim 10^3\text{ S/cm}$ ) and chemical stability. This chapter introduces the definition and chemical composition of cesium tungsten bronze, its discovery and development history, its position in materials science, global research status and market overview, and key application areas, providing background for subsequent chapters (Chapter 2 to Chapter 10). This encyclopedia aims to systematically explain the theoretical basis, preparation technology, performance characterization, application scenarios, industrialization, regulatory requirements, sustainability and future directions of cesium tungsten bronze.

### 1.1 Cesium Tungsten Bronze

Cesium tungsten bronze is a tungsten-based oxide with the chemical formula  $\text{Cs}_x\text{WO}_3$ , where  $x$  represents the doping ratio of cesium (Cs), usually varying between 0 and 1.  $\text{Cs}_x\text{WO}_3$  belongs to the tungsten bronze family, and its structure is composed of  $\text{WO}_6$  octahedrons, with cesium ions inserted into the octahedral gaps to form a hexagonal or cubic crystal structure (Chapter 2.1). The change in the  $x$  value significantly affects the performance of the material. For example, when  $x \sim 0.32$ ,  $\text{Cs}_{0.32}\text{WO}_3$  exhibits the best near-infrared absorption and conductivity.

- **Chemical composition :**
  - **Main elements :** cesium (Cs), tungsten (W), oxygen (O).
  - **Molar ratio :**  $\text{Cs}_x\text{W}_1\text{O}_3$ ,  $x \leq 1$ , oxygen content is fixed at 3.
  - **Molecular weight :** Taking  $\text{Cs}_{0.32}\text{WO}_3$  as an example,  $\sim 287.3\text{ g/mol}$ .
  - **Purity requirements :** industrial grade  $\geq 99.5\%$ , research grade  $\geq 99.9\%$  (Chapter

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7.4).

- **Physical properties :**

- **Appearance :** Dark blue or green nanopowder, particle size ~20–50 nm (Chapter 3.5).
- **Density :** ~7.2 g/cm<sup>3</sup>.
- **Solubility :** Insoluble in water, resistant to acid and alkali (Chapter 4.3).

Cesium tungsten bronze determines its unique optical and electrical properties, making it widely used in smart window films (Chapter 5.1), photocatalysis (Chapter 6.2) and batteries (Chapter 6.1). Compared with other tungsten bronzes (such as Na<sub>x</sub>WO<sub>3</sub> and K<sub>x</sub>WO<sub>3</sub>), Cs<sub>x</sub>WO<sub>3</sub> exhibits stronger NIR shielding performance (~70% vs. ~50% for Na<sub>x</sub>WO<sub>3</sub>) due to the larger ionic radius of cesium ions (~1.88 Å).

## 1.2 Discovery and Development of Cesium Tungsten Bronze

Cesium tungsten bronze originated from the study of tungsten bronze in the 19th century. In 1823, German chemist Wöhler first synthesized tungsten bronze and observed dark compounds formed by alkali metal doped WO<sub>3</sub>. In the 1950s, Japanese scientist Kihlberg confirmed the hexagonal crystal structure of Cs<sub>x</sub>WO<sub>3</sub> through X-ray diffraction (XRD), laying the foundation for the structure (Chapter 4, 4.1). In the 1970s, Cs<sub>x</sub>WO<sub>3</sub> was used in display research due to its electrochromic properties (~60% transmittance change).

- **Key Milestones :**

- **1980s :** American researchers discovered the NIR absorption properties of Cs<sub>x</sub>WO<sub>3</sub> (~1000–2500 nm), which promoted its exploration in the field of optical coatings (Chapter 5.2).
- **1990s :** Japan developed the solvothermal method (Chapter 3.2), which enabled large-scale synthesis of Cs<sub>x</sub>WO<sub>3</sub> nanoparticles (<50 nm), reducing the cost to ~1000 USD/kg.
- **2000s :** Chinese research teams optimized the photocatalytic performance of Cs<sub>x</sub>WO<sub>3</sub> (Chapter 6.2), with a hydrogen production efficiency of ~200 μmol / (g · h).
- **2010s :** The EU promotes the application of Cs<sub>x</sub>WO<sub>3</sub> in smart window films (Chapter 5.1), with energy saving efficiency of ~50% and market growth to ~US\$50 million.
- **2020s :** Global focus on green synthesis (Chapter 3.5), carbon footprint reduced to ~0.5 tons CO<sub>2</sub>/ton (Chapter 9.4).

In recent years, the research on cesium tungsten bronze has shifted from basic performance to industrialization (Chapter 7) and sustainability (Chapter 9), especially in the Asia-Pacific region, where China supports the energy-saving application of Cs<sub>x</sub>WO<sub>3</sub> through its "dual carbon" policy (Chapter 9 9.5).

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### 1.3 The Status of Cesium Tungsten Bronze in Materials Science

Cesium tungsten bronze occupies an important position in materials science because it combines the properties of nanomaterials, semiconductors and optical materials, filling the gaps in traditional materials in the fields of NIR regulation and energy conversion.

- **Scientific value :**
  - **Nano properties :** Cs<sub>x</sub>WO<sub>3</sub> nanoparticles (~20 nm) have a high specific surface area (~80 m<sup>2</sup>/g, Chapter 4.2), which improves catalytic efficiency (Chapter 6.2).
  - **Semiconductor properties :** band gap ~2.5–3.0 eV (Chapter 2.2), supporting photoelectric conversion (Chapter 5.3).
  - **Plasmon effect :** Localized surface plasmon resonance (LSPR) enhances NIR absorption (~70%), which is better than traditional ITO (~40%, Chapter 5.2).
- **Comparison with other materials :**
  - **Compared with ITO :** Cs<sub>x</sub>WO<sub>3</sub> has advantages in NIR shielding (~70% vs. ~40%) and cost (~500 USD/kg vs. ~1000 USD/kg).
  - **Compared with VO<sub>2</sub> :** The thermal stability of Cs<sub>x</sub>WO<sub>3</sub> (>500°C vs. ~68°C phase transition) is more suitable for high temperature environments (Chapter 5, 5.5).
  - **Compared to graphene :** Cs<sub>x</sub>WO<sub>3</sub> is more specific in NIR absorption, but has slightly lower conductivity (~10<sup>3</sup> vs. ~10<sup>6</sup> S/cm, Chapter 2, 2.3).
- **Interdisciplinary impact :**
  - Promote the development of photonics (Chapter 5.4), energy storage (Chapter 6.1) and environmental science (Chapter 6.3).
  - It provides a research paradigm for functional nanomaterials (such as MXenes and MoS<sub>2</sub>) (Chapter 10, 10.2).

Cesium tungsten bronze has put it at the forefront of materials science, especially in the fields of energy conservation and environmental protection (Chapter 9.1).

### 1.4 Global Research Status and Market Overview of Cesium Tungsten Bronze

The global research and market of Cesium Tungsten Bronze shows rapid growth by 2025, especially in Asia Pacific, Europe, and North America.

- **Research status :**
  - **China :** Tsinghua University and other institutions focus on green synthesis (Chapter 3, 3.5) and photocatalysis (Chapter 6, 6.2), with an average of ~150 patent applications per year.
  - **Japan :** The University of Tokyo optimized Cs<sub>x</sub>WO<sub>3</sub> thin film (Chapter 5.1), with a NIR shielding rate of ~80%.
  - **EU :** Germany's Fraunhofer Institute has developed Cs<sub>x</sub>WO<sub>3</sub> battery materials (Chapter 6.1) with a cycle life of >1000 times.
  - **USA :** MIT explores the quantum effects of Cs<sub>x</sub>WO<sub>3</sub> (Chapter 2.5), increasing conductivity by ~20%.

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- **Market Overview :**
  - **Size :** The global market is expected to reach US\$120 million in 2025 and increase to US\$250 million in 2030 (average annual growth of ~15%).
  - **Major regions :** Asia Pacific ~50% (China ~30%), Europe ~30%, North America ~15%.
  - **Price :** Nano-grade CsxWO<sub>3</sub> ~ 500 USD/kg, thin film grade ~ 1000 USD/kg (Chapter 7.2).
  - **Driving factors :** energy-saving demand (smart window film, Chapter 5, 5.1), new energy (batteries, Chapter 6, 6.1) and environmental protection policies (Chapter 9, 9.5).
- **challenge :**
  - High synthesis cost (~500 USD/kg vs. ITO ~100 USD/kg).
  - The toxicity of nanoparticles needs to be evaluated (Chapter 8, 8.3).
  - The consistency of large-scale production is low (Chapter 7.3, error ~10%).

Global research is shifting towards low-cost synthesis (Chapter 3.5) and intelligent application (Chapter 10.3) to meet market demand.

## 1.5 Key Application Fields of Cesium Tungsten Bronze

Cesium tungsten bronze is widely used in the following fields due to its versatility, see Chapter 5 to Chapter 6 for details.

- **Optics and Thermal Engineering (Chapter 5) :**
  - **Smart window film :** CsxWO<sub>3</sub> coating reduces building energy consumption by ~50% (Chapter 5.1).
  - **Photothermal conversion :** Solar energy absorption efficiency ~60% (Chapter 5.3).
  - **NIR shielding :** Automotive glass coating, shielding rate ~70% (Chapter 5.2).
- **Energy (Chapter 6) :**
  - **Battery :** CsxWO<sub>3</sub> electrode, energy density ~200 Wh /kg (Chapter 6, 6.1).
  - **Photocatalysis :** hydrogen production efficiency ~200 μmol / ( g·h ) (Chapter 6, 6.2).
  - **Hydrogen storage :** Hydrogen storage capacity ~1.5 wt % (Chapter 6, 6.5).
- **Environment (Chapter 6) :**
  - **Air purification :** adsorption of VOCs, efficiency ~90% (Chapter 6, 6.3).
  - **Water treatment :** Photocatalytic degradation of dyes, efficiency ~85% (Chapter 6, 6.2).
- **Electronics (Chapter 5) :**
  - **Sensor :** CsxWO<sub>3</sub> thin film, sensitivity ~10 ppm (NO<sub>2</sub>, Chapter 5.4).
  - **Display :** Electrochromic, response time <1 s (Chapter 5.4).
- **Case :** In 2024, CTIA GROUP LTD developed CsxWO<sub>3</sub> smart window film, which was applied to a green building in Shanghai, saving energy by ~40% and with a market value

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of ~US\$10 million (Chapter 7.5).

These application fields demonstrate the strategic value of cesium tungsten bronze in energy conservation, environmental protection and new energy, and will be further expanded in intelligent and green manufacturing in the future (Chapter 10, 10.1–10.5).

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## Chapter 2: Crystal structure and properties of cesium tungsten bronze

Cesium tungsten bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) is a functional nanomaterial whose unique crystal structure and excellent physical and chemical properties make it widely used in the fields of optics, electronics, energy and thermal management. The performance of  $\text{Cs}_x\text{WO}_3$  comes from its hexagonal or cubic crystal structure, near-infrared (NIR) absorption ability ( $\sim 70\%$  at  $1000\text{ nm}$ ), high conductivity ( $\sim 10^3\text{ S/cm}$ ) and thermal stability ( $>500^\circ\text{C}$ ). This chapter discusses the crystal structure and chemical bond characteristics, optical properties (NIR absorption and transmittance), electrical properties (conductivity and carrier migration), thermal properties (thermal conductivity and stability), as well as theoretical calculations and performance predictions of cesium tungsten bronze, providing a basis for subsequent application and preparation research.

### 2.1 Crystal structure and chemical bond characteristics of cesium tungsten bronze

cesium tungsten bronze is the core of its functional performance.  $\text{Cs}_x\text{WO}_3$  is composed of  $\text{WO}_6$  octahedrons, and cesium ions ( $\text{Cs}^+$ ) are inserted into the octahedral gaps to form a hexagonal ( $x \sim 0.32$ ) or cubic ( $x \sim 1$ ) crystal system. The hexagonal structure (space group  $\text{P6}_3/\text{mcm}$ ) has a one-dimensional tunnel, which is conducive to the migration of cesium ions, while the cubic structure ( $\text{Pm}\bar{3}\text{m}$ ) is more compact, affecting the electrical properties.

- **Structural features :**

- **WO<sub>6</sub> octahedron :** Tungsten ( $\text{W}^{6+}$ ) forms an octahedron with six oxygens ( $\text{O}^{2-}$ ), and the WO bond length is  $\sim 1.9\text{ \AA}$ .
- **Cesium doping :**  $\text{Cs}^+$  (radius  $\sim 1.88\text{ \AA}$ ) occupies hexagonal tunnels or cubic interstitials, with the  $x$  value determining the structure and properties.
- **Lattice parameters :** hexagonal  $\text{Cs}_{0.32}\text{WO}_3$ ,  $a \sim 7.4\text{ \AA}$ ,  $c \sim 7.6\text{ \AA}$ ; cubic  $\text{CsWO}_3$ ,

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$a \sim 3.8 \text{ \AA}$ .

- **Effect of particle size** : Nanoparticles ( $\sim 20 \text{ nm}$ ) are slightly distorted due to surface effects, with a lattice strain of  $\sim 0.5\%$ .
- **Chemical bond characteristics** :
  - **WO bond** : Covalent-ionic mixed bond, imparting high chemical stability (corrosion rate  $< 0.001 \text{ mm/year}$ ).
  - **Cs-O bond** : Weak ionic bond,  $\text{Cs}^+$  can partially migrate, affecting conductivity ( $\sim 10^3 \text{ S/cm}$ ).
  - **Localized electrons** :  $\text{W}^{5+} / \text{W}^{6+}$  mixed valence states generate free electrons and enhance NIR absorption.

X-ray diffraction (XRD) analysis shows that the hexagonal structure of  $\text{Cs}_{0.32}\text{WO}_3$  is superior to the cubic structure in NIR shielding and conductivity because its tunneling is conducive to the dynamic behavior of electrons and ions. Adjusting the doping ratio ( $x$ ) can optimize the performance. For example, when  $x=0.32$ , the NIR absorption rate reaches  $\sim 70\%$ , while when  $x=0.5$ , the conductivity is improved by  $\sim 15\%$ .

## 2.2 Optical properties of cesium tungsten bronze: near-infrared absorption and transmittance

cesium tungsten bronze are known for its NIR absorption and visible light transmittance, which gives it unique advantages in smart window films and light-to-heat conversion.

- **Near infrared absorption** :
  - **Mechanism** : Localized surface plasmon resonance (LSPR) and electronic transitions between  $\text{W}^{5+} / \text{W}^{6+}$  lead to strong NIR absorption ( $800\text{--}2500 \text{ nm}$ ).
  - **Performance** :  $\text{Cs}_{0.32}\text{WO}_3$  nanoparticles ( $\sim 20 \text{ nm}$ ) have an absorption rate of  $\sim 70\%$  at  $1000 \text{ nm}$ , which is better than ITO ( $\sim 40\%$ ).
  - **Influencing factors** : As the  $x$  value increases ( $0.1 \rightarrow 0.32$ ), the absorption peak red-shifts by  $\sim 200 \text{ nm}$ ; as the particle size decreases ( $50 \rightarrow 20 \text{ nm}$ ), the absorption increases by  $\sim 10\%$ .
- **Visible light transmittance** :
  - **Transmittance** : In the range of  $400\text{--}700 \text{ nm}$ , the transmittance of  $\text{Cs}_x\text{WO}_3$  film (thickness  $\sim 100 \text{ nm}$ ) is  $\sim 80\%$ , which is suitable for energy-saving glass.
  - **Electrochromic** : With applied voltage ( $\sim 2 \text{ V}$ ), transmittance can be adjusted to  $\sim 60\%$  for dynamic dimming.
  - **Stability** : After UV aging ( $1000 \text{ h}$ ), the transmittance decay is  $< 5\%$ .
- **Application potential** :
  - Smart window film: NIR shielding rate  $\sim 70\%$ , reducing building energy consumption  $\sim 50\%$ .
  - Photothermal conversion: Solar energy absorption efficiency  $\sim 60\%$  for thermal management.

Ultraviolet-visible-near-infrared spectroscopy (UV-Vis-NIR) tests show that the band gap of

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CsxWO<sub>3</sub> is ~2.5–3.0 eV, which is lower than that of WO<sub>3</sub> (~3.2 eV), which enhances the NIR response. Compared with VO<sub>2</sub> (phase transition temperature ~68°C), CsxWO<sub>3</sub> is more stable at high temperatures and is suitable for a wide range of climate conditions.

### 2.3 Electrical properties of cesium tungsten bronze: conductivity and carrier migration

cesium tungsten bronze come from its semiconductor properties and free electrons, making it suitable for sensors and battery electrodes.

- **Conductivity :**
  - **Value :** The conductivity of Cs<sub>0.32</sub>WO<sub>3</sub> thin film (thickness ~100 nm) is ~10<sup>3</sup> S/cm, which is lower than graphene (~10<sup>6</sup> S/cm) but higher than WO<sub>3</sub> (~10 S/cm).
  - **Mechanism :** W<sup>5+</sup> /W<sup>6+</sup> mixed valence state and Cs<sup>+</sup> doping introduce free electrons with carrier concentration of ~10<sup>20</sup> cm<sup>-3</sup>.
  - **Influencing factors :** As the x value increases (0.1→0.5), the conductivity increases by ~20%; when the temperature rises to 300°C, the conductivity decreases by ~10%.
- **Carrier migration :**
  - **Mobility :** Electron mobility ~10 cm<sup>2</sup>/(V·s), affected by grain boundary scattering.
  - **Type :** n-type semiconductor, electrons are the dominant carriers.
  - **Test :** Hall effect measurements show that the carrier lifetime of Cs<sub>0.32</sub>WO<sub>3</sub> is ~1 ps.
- **Application potential :**
  - **Sensor :** CsxWO<sub>3</sub> thin film detects NO<sub>2</sub> (~10 ppm) with a response time of <5 s.
  - **Battery Electrode :** Li-ion battery, conductivity supports cycle life ~1000 times.

Four-probe measurements show that the conductivity of nanoparticles (~20 nm) is lower than that of thin films (~100 nm) due to a ~30% increase in grain boundary resistance. Optimizing doping (x~0.32) and annealing (~400°C) can improve performance.

### 2.4 Thermal properties of cesium tungsten bronze: thermal conductivity and stability

Cesium Tungsten Bronze include low thermal conductivity and high thermal stability, which is suitable for thermal management and high temperature applications.

- **Thermal Conductivity :**
  - **Value :** Thermal conductivity of Cs<sub>0.32</sub>WO<sub>3</sub> (~20 nm) is ~1.5 W/(m·K), which is lower than that of WO<sub>3</sub> (~3 W/(m·K)).
  - **Mechanism :** Phonon scattering and nanosize effect reduce thermal conductivity.
  - **Influencing factors :** As the x value increases (0.1→0.5), the thermal conductivity increases by ~10%; as the particle size increases (20→50 nm), the thermal conductivity increases by ~15%.
- **Thermal stability :**
  - **Temperature range :** CsxWO<sub>3</sub> is stable at <500°C, oxygen loss is ~5% at >600°C.
  - **Oxidation resistance :** 500°C in air, oxidation rate <0.01 mg/cm<sup>2</sup>·h.

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- **Thermal expansion** : The thermal expansion coefficient is  $\sim 8 \times 10^{-6} \text{ K}^{-1}$ , which is lower than that of ITO ( $\sim 10 \times 10^{-6} \text{ K}^{-1}$ ).
- **Application potential** :
  - **Thermal Management** : Low thermal conductivity for thermal barrier coatings, reducing temperatures by  $\sim 10^\circ\text{C}$ .
  - **High temperature electrodes** : Battery electrodes are stable at  $400^\circ\text{C}$  with a cycle performance degradation of  $< 5\%$ .

Differential scanning calorimetry (DSC) showed that the thermal decomposition temperature of  $\text{CsxWO}_3$  was  $\sim 650^\circ\text{C}$ , which is better than  $\text{VO}_2$  ( $\sim 68^\circ\text{C}$  phase transition). Thermogravimetric analysis (TGA) confirmed its high temperature stability and suitability for harsh environments.

## 2.5 Theoretical calculation and performance prediction of cesium tungsten bronze

Theoretical calculations provide guidance for the performance optimization of  $\text{CsxWO}_3$ , using density functional theory (DFT) and molecular dynamics (MD) methods.

- **DFT calculations** :
  - **Band gap** :  $\text{Cs}_{0.32}\text{WO}_3$  band gap  $\sim 2.5 \text{ eV}$ ,  $\text{W}^{5+}/\text{W}^{6+}$  state enhances NIR absorption.
  - **Electronic structure** :  $\text{Cs}^+$  doping lowers the Fermi level by  $\sim 0.5 \text{ eV}$  and increases the carrier concentration by  $\sim 10^{20} \text{ cm}^{-3}$ .
  - **Optical properties** : The simulated NIR absorption peak is  $\sim 1000 \text{ nm}$ , which is consistent with the experiment (error  $< 5\%$ ).
- **MD simulation** :
  - **Thermal conductivity** : predicted  $\sim 1.5 \text{ W/(m}\cdot\text{K)}$  due to phonon scattering.
  - **Stability** :  $\text{Cs}^+$  has a mobility of  $\sim 10^{-8} \text{ cm}^2/\text{s}$  at  $500^\circ\text{C}$  and maintains structural integrity.
- **Performance prediction** :
  - **Doping optimization** :  $x=0.4$  can increase conductivity by  $\sim 25\%$  and NIR absorption by  $\sim 10\%$ .
  - **Nano-design** : particle size  $< 10 \text{ nm}$ , LSPR enhanced absorption  $\sim 15\%$ .
  - **Composite material** :  $\text{CsxWO}_3/\text{TiO}_2$  composite, photocatalytic efficiency increased by  $\sim 30\%$ .

Case: CTIA GROUP LTD optimized  $\text{Cs}_{0.32}\text{WO}_3$  thin film through DFT, with NIR shielding rate reaching  $\sim 75\%$  and cost reduction of  $\sim 20\%$  ( $\sim 400 \text{ USD/kg}$ ). The calculation results guided the development of green synthesis and intelligent applications.

The accuracy of the theoretical model (error  $< 5\%$ ) provides a basis for the performance regulation of  $\text{CsxWO}_3$ . In the future, AI and big data will be combined to further improve the prediction accuracy.

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### Chapter 3: Synthesis Method of Cesium Tungsten Bronze

Cesium tungsten bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) is a high-performance nanomaterial whose near-infrared absorption ( $\sim 70\%$  at  $1000\text{ nm}$ ), conductivity ( $\sim 10^3\text{ S/cm}$ ), and chemical stability depend on the precise control of the synthesis method. The synthesis method determines the crystal structure (hexagonal or cubic), particle size ( $\sim 10\text{--}50\text{ nm}$ ), doping ratio ( $x \sim 0.1\text{--}1$ ), and performance (such as NIR shielding rate  $\sim 70\%$ ) of  $\text{Cs}_x\text{WO}_3$ . This chapter discusses five main synthesis methods in detail: solid-phase reaction method, solvothermal method and hydrothermal method, chemical vapor deposition (CVD), sol-gel method, and green synthesis and nanoparticle control, analyzes their reaction mechanism, process parameters, equipment requirements, yield, cost, advantages and disadvantages, and application scenarios, and provides technical references for research and industrialization.

#### 3.1 Solid-state reaction method of cesium tungsten bronze

The solid phase reaction method is a high-temperature synthesis method that prepares  $\text{Cs}_x\text{WO}_3$  through direct reaction of solid raw materials. It is suitable for bulk material and powder production. Its advantages are simple equipment and high yield, but the particle size is large ( $\sim 1\text{--}10\text{ }\mu\text{m}$ ), making it difficult to prepare nanoparticles.

- **Reaction mechanism :**
  - **Chemical equation :**  $\text{Cs}_2\text{CO}_3 + \text{WO}_3 \rightarrow \text{Cs}_x\text{WO}_3 + \text{CO}_2\uparrow$  ( $x \sim 0.1\text{--}0.32$ )
  - **Process :**  $\text{Cs}_2\text{CO}_3$  decomposes to release  $\text{Cs}^+$ , which combines with the  $\text{WO}_6$  octahedron in  $\text{WO}_3$ , and  $\text{Cs}^+$  is inserted into the hexagonal tunnel to form  $\text{Cs}_x\text{WO}_3$ . High temperature ( $\sim 800^\circ\text{C}$ ) promotes diffusion, and  $\text{W}^{6+}$  is partially reduced to  $\text{W}^{5+}$ , generating free electrons.
  - **Side reaction :** Excessively high temperature ( $>1000^\circ\text{C}$ ) causes Cs to volatilize and the x value decreases by  $\sim 10\%$ .
- **Process parameters :**
  - **Raw materials :**  $\text{Cs}_2\text{CO}_3$  (purity  $>99.5\%$ ),  $\text{WO}_3$  ( $>99.9\%$ ), molar ratio  $1:3\text{--}1:10$ .
  - **Temperature :**  $700\text{--}900^\circ\text{C}$ , ramp rate  $\sim 10^\circ\text{C/min}$ .
  - **Atmosphere :** Inert (Ar) or reducing ( $\text{H}_2/\text{Ar}$ ,  $5\%\text{ H}_2$ ), prevent oxidation.

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- **Time** : 4–8 h, cooling rate  $\sim 5^{\circ}\text{C}/\text{min}$ .
- **Equipment** : Tube furnace (temperature resistance  $> 1200^{\circ}\text{C}$ ), alumina crucible.
- **Yield and cost** :
  - **Yield** :  $\sim 90\%$  (mass basis),  $\sim 1\text{--}10$  kg per batch.
  - **Cost** :  $\sim 200$  USD/kg,  $\sim 60\%$  raw materials,  $\sim 30\%$  energy.
- **performance** :
  - **Particle size** :  $\sim 1\text{--}10\ \mu\text{m}$ , surface area  $\sim 5\ \text{m}^2/\text{g}$ .
  - **Purity** :  $> 99.5\%$ , impurities (Fe, Si)  $< 0.01\ \text{wt}\%$ .
  - **NIR absorption** :  $\sim 50\%$  at  $1000\ \text{nm}$ , below the nanometer scale ( $\sim 70\%$ ).
  - **Electrical conductivity** :  $\sim 500\ \text{S}/\text{cm}$ , affected by grain boundaries.
- **Pros and Cons** :
  - **Advantages** : The process is mature, suitable for industrialization, and the equipment cost is low ( $\sim 10,000$  USD).
  - **Disadvantages** : large particle size, weak NIR performance, Cs volatilization leads to uneven x value (error  $\sim 5\%$ ).
  - **Improvement** : Adding additives (such as  $\text{Li}_2\text{CO}_3$ ) reduces the reaction temperature by  $\sim 100^{\circ}\text{C}$ , and ball milling treatment reduces the particle size by  $\sim 50\%$ .
- **application** :
  - Bulk electrode: battery electrode, cycle life  $\sim 800$  times.
  - Coarse powder : Thermal management coating, cooling  $\sim 5^{\circ}\text{C}$ .

The solid-phase reaction method is suitable for low-cost, large-scale production, but needs to be optimized to meet nanoscale requirements.

### 3.2 Solvothermal and hydrothermal methods of cesium tungsten bronze

Solvothermal and hydrothermal methods synthesize  $\text{Cs}_x\text{WO}_3$  nanoparticles ( $\sim 10\text{--}50\ \text{nm}$ ) through high temperature and high pressure liquid phase reactions, which are known for their high specific surface area ( $\sim 80\ \text{m}^2/\text{g}$ ) and excellent NIR performance ( $\sim 70\%$ ). The hydrothermal method uses water as the solvent, while the solvothermal method uses organic solvents (such as ethanol).

- **Reaction mechanism** :
  - **Chemical equation** :  $\text{CsOH} + \text{WCl}_6 + \text{H}_2\text{O}/\text{ROH} \rightarrow \text{Cs}_x\text{WO}_3 + \text{HCl}\uparrow$  ( $x \sim 0.2\text{--}0.5$ )
  - **Hydrothermal method** : CsOH provides  $\text{Cs}^+$ ,  $\text{WCl}_6$  hydrolyzes to form  $\text{WO}_6$  units, and  $\text{Cs}^+$  is inserted into the hexagonal structure at  $180\text{--}220^{\circ}\text{C}$ .
  - **Solvothermal method** : Ethanol reduces surface tension, controls particle size  $< 20\ \text{nm}$ , and increases  $\text{W}^{5+}$  ratio by  $\sim 15\%$ .
  - **Key** : pH  $\sim 8\text{--}10$ , regulates  $\text{Cs}^+$  insertion efficiency  $\sim 90\%$ .
- **Process parameters** :
  - **Raw materials** : CsOH ( $> 99.5\%$ ),  $\text{WCl}_6$  ( $> 99.9\%$ ), solvent (deionized water or ethanol,  $> 99.8\%$ ).
  - **Temperature** : hydrothermal  $180\text{--}220^{\circ}\text{C}$ , solvothermal  $150\text{--}200^{\circ}\text{C}$ .

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- **Pressure** : 1–5 MPa, closed reactor.
- **Time** : 12–24 h, stirring rate ~200 rpm.
- **Equipment** : Stainless steel reactor (volume ~1–100 L), pressure resistance >10 MPa.
- **Yield and cost** :
  - **Yield** : ~85% (hydrothermal), ~80% (solvothermal), ~0.1–1 kg per batch.
  - **Cost** : hydrothermal ~400 USD/kg, solvent thermal ~500 USD/kg, solvent accounts for ~40%.
- **performance** :
  - **Particle size** : hydrothermal ~20–50 nm, solvothermal ~10–20 nm.
  - **Purity** : >99.8%, Cl<sup>-</sup> residual < 0.005 wt %.
  - **NIR absorption** : ~70% at 1000 nm, transmittance ~80% (400–700 nm).
  - **Conductivity** : ~800 S/cm, better than solid phase method.
- **Pros and Cons** :
  - **Advantages** : Nano-scale particle size, high NIR performance, controllable morphology (rods, spheres).
  - **Disadvantages** : Complex equipment (~50,000 USD), high waste treatment cost (~10%).
  - **Improvements** : Adding surfactants (such as PVP) to control particle size distribution (<5% error), recycling solvents to reduce costs by ~20%.
- **application** :
  - Smart window film: NIR shielding rate ~70%, energy saving ~50%.
  - Photocatalysis: hydrogen production efficiency ~200 μmol / ( g · h ).

The hydrothermal method has lower cost, while the solvothermal method has smaller particle size, which is suitable for high performance requirements.

### 3.3 Chemical Vapor Deposition (CVD) of Cesium Tungsten Bronze

Chemical vapor deposition (CVD) deposits Cs<sub>x</sub>WO<sub>3</sub> thin films (thickness ~10–100 nm) through vapor phase precursor reactions, which are suitable for electronic and optical devices. It is characterized by film uniformity and high purity (>99.9%).

- **Reaction mechanism** :
  - **Chemical equation** :  $\text{CsCl (g)} + \text{WOCl}_4\text{(g)} + \text{H}_2\text{O(g)} \rightarrow \text{Cs}_x\text{WO}_3\text{(s)} + \text{HCl}\uparrow$  (x~0.1–0.32)
  - **Process** : CsCl and WOCl<sub>4</sub> are vaporized at 400–600°C, H<sub>2</sub>O provides oxygen, Cs<sup>+</sup> and WO<sub>6</sub> are nucleated on a substrate (such as glass), and the W<sup>5+</sup> ratio is ~10%.
  - **Side reaction** : Excessively high temperature (>700°C) causes WO<sub>3</sub> precipitation and the x value decreases by ~5%.
- **Process parameters** :
  - **Precursors** : CsCl (>99.9%), WOCl<sub>4</sub> (>99.8%), H<sub>2</sub>O (ultrapure).
  - **Temperature** : 400–600°C for substrate, ~800°C for reaction chamber.

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- **Atmosphere** : Ar /H<sub>2</sub> (5% H<sub>2</sub>), flow rate ~100 sccm .
- **Pressure** : 0.1–10 Torr.
- **Equipment** : CVD reactor ( hot wall or cold wall), vacuum pump (<0.01 Torr).
- **Yield and cost** :
  - **Yield** : ~95% (deposition efficiency), ~0.01–0.1 kg per batch.
  - **Cost** : ~1000 USD/kg, ~50% for equipment.
- **performance** :
  - **Thickness** : 10–100 nm, uniformity <5% error.
  - **Purity** : >99.9%, impurity (Cl) <0.001 wt %.
  - **NIR absorption** : ~75% at 1000 nm, transmittance ~85% (400–700 nm).
  - **Conductivity** : ~1200 S/cm, better than powder.
- **Pros and Cons** :
  - **Advantages** : High quality film, suitable for precision devices, excellent NIR performance.
  - **Disadvantages** : high cost, low yield, substrate limitation (needs temperature resistance > 400°C).
  - **Improvement** : Plasma-enhanced CVD (PECVD) reduces temperature by ~100°C and increases efficiency by ~15%.
- **application** :
  - 6G antenna: transmission rate ~10 Gbps.
  - Electrochromic device: response time <1 s.

CVD is suitable for high value-added films, but costs need to be reduced for its promotion.

### 3.4 Sol-gel method of cesium tungsten bronze

The sol-gel method prepares Cs<sub>x</sub>WO<sub>3</sub> nanoparticles or films through liquid phase chemical reactions, which has the advantages of low temperature and morphology control and is suitable for laboratory and small- and medium-scale production.

- **Reaction mechanism** :
  - **Chemical equation** :  $\text{CsNO}_3 + \text{W}(\text{OC}_2\text{H}_5)_6 + \text{H}_2\text{O} \rightarrow \text{Cs}_x\text{WO}_3 + \text{C}_2\text{H}_5\text{OH}\uparrow$  (x ~ 0.2–0.4)
  - **Process** : W(OC<sub>2</sub>H<sub>5</sub>)<sub>6</sub> is hydrolyzed to form WO<sub>6</sub> gel, CsNO<sub>3</sub> provides Cs<sup>+</sup>, and calcined at 300–500°C to form Cs<sub>x</sub>WO<sub>3</sub> nanoparticles or films.
  - **Key** : Control the hydrolysis rate (pH ~7–9) to avoid WO<sub>3</sub> precipitation.
- **Process parameters** :
  - **Raw materials** : CsNO<sub>3</sub> (>99.5%), W(OC<sub>2</sub>H<sub>5</sub>)<sub>6</sub> (>99.8%), ethanol/water mixed solvent.
  - **Temperature** : 25–60°C for hydrolysis, 300–500°C for calcination.
  - **Time** : 2–4 h for gelation and 4–6 h for calcination.
  - **Equipment** : Constant temperature water bath, box furnace, spin coater (thin film).
- **Yield and cost** :
  - **Yield** : ~90%, ~0.1–0.5 kg per batch.

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- **Cost** : ~600 USD/kg, ~70% of which is raw materials.
- **Performance** :
  - **Particle size** : ~15–30 nm (powder), thickness ~50–200 nm (thin film).
  - **Purity** : >99.7%, C residue <0.01 wt %.
  - **NIR absorption** : ~65% at 1000 nm, transmittance ~80%.
  - **Conductivity** : ~700 S/cm.
- **Pros and Cons** :
  - **Advantages** : low temperature process, simple equipment (~5,000 USD), various forms (powder, film).
  - **Disadvantages** : organic residue, slightly lower purity, and high calcination energy consumption.
  - **Improvement** : Microwave-assisted calcination, time reduced by ~50%, purity increased by ~0.2%.
- **Application** :
  - Photocatalyst: Degrades dyes with an efficiency of ~85%.
  - Energy-saving coating: NIR shielding rate ~65%.

The sol-gel method is highly flexible and suitable for research and development and customized production.

### 3.5 Green Synthesis and Nanoparticle Control of Cesium Tungsten Bronze

Green synthesis prepares Cs<sub>x</sub>WO<sub>3</sub> nanoparticles (<20 nm) through environmentally friendly raw materials and low-energy consumption processes, emphasizing sustainability and nanoscale control to meet the 2030 carbon neutrality goal.

- **Reaction mechanism** :
  - **Chemical equation** :  $\text{Cs}_2\text{CO}_3 + (\text{NH}_4)_{10}\text{W}_{12}\text{O}_{41} + \text{H}_2\text{O} \rightarrow \text{Cs}_x\text{WO}_3 + \text{NH}_3\uparrow$  (x ~ 0.2–0.32)
  - **Process** : (NH<sub>4</sub>)<sub>10</sub>W<sub>12</sub>O<sub>41</sub> is used as an environmentally friendly tungsten source, Cs<sub>2</sub>CO<sub>3</sub> provides Cs<sup>+</sup>, and low-temperature (<200°C) hydrothermal or microwave-assisted reaction is used to control the particle size to ~10–20 nm.
  - **Key** : Bio-template (e.g. cellulose) or ultrasound-assisted, particle size distribution <5%.
- **Process parameters** :
  - **Raw materials** : Cs<sub>2</sub>CO<sub>3</sub> (>99.5%), (NH<sub>4</sub>)<sub>10</sub>W<sub>12</sub>O<sub>41</sub> (>99.8%), deionized water.
  - **Temperature** : 100–200°C (hydrothermal), 80–120°C (microwave).
  - **Time** : 6–12 h (hydrothermal), 0.5–2 h (microwave).
  - **Equipment** : microwave reactor (~1000 W), ultrasonic cleaner (~500 W).
- **Yield and cost** :
  - **Yield** : ~80%, ~0.05–0.5 kg per batch.
  - **Cost** : ~450 USD/kg, ~20% energy.
- **performance** :
  - **Particle size** : ~10–20 nm, surface area ~100 m<sup>2</sup>/g.

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- **Purity** : >99.8%,  $\text{NH}_4^+$  residual <0.005 wt %.
- **NIR absorption** : ~72% at 1000 nm, transmittance ~82%.
- **Conductivity** : ~900 S/cm.
- **Green Features** :
  - **Carbon footprint** : ~0.3 tons  $\text{CO}_2$ /ton, lower than traditional hydrothermal (~0.5 tons).
  - **Waste liquid** :  $\text{NH}_3$  recovery rate >95%, water recycling ~80%.
  - **Energy consumption** : Microwave method is ~50% lower than hydrothermal (~100 kWh/kg).
- **Pros and Cons** :
  - **Advantages** : environmental protection, low energy consumption, small particle size and excellent performance.
  - **Disadvantages** : Slightly lower yield, high equipment investment (~20,000 USD).
  - **Improvement** : AI optimizes reaction parameters, increasing yield by ~10% and reducing costs by ~15%.
- **Case** : In 2024, CTIA GROUP LTD used microwave-assisted green synthesis to prepare  $\text{Cs}_{0.32}\text{WO}_3$  (~15 nm), with NIR shielding rate of ~72% and carbon footprint of ~0.3 tons  $\text{CO}_2$ /ton, which was used in green building window films.
- **application** :
  - Air purification: VOCs adsorption, efficiency ~90%.
  - Hydrogen storage: Capacity ~1.5 wt %.

Green synthesis represents the future trend and combined with nano-control technology will promote the sustainable industrialization of  $\text{CsxWO}_3$ .

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Cesium Tungsten Bronze Product Introduction

CTIA GROUP LTD

1. Cesium Tungsten Bronze Overview

Cesium Tungsten Bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) produced by CTIA GROUP is manufactured using advanced solvothermal and chemical vapor deposition processes, ensuring high purity and excellent optoelectronic performance. Cesium tungsten bronze is a nano-functional material widely applied in energy-saving glass, optoelectronics, catalysis, batteries, and biomedical fields due to its outstanding near-infrared (NIR) shielding performance, high visible light transmittance, and electrochemical stability. Its unique tungsten-oxygen-cesium crystal structure makes it the preferred material for smart materials and new energy applications.

2. Cesium Tungsten Bronze Features

- Chemical composition:  $\text{Cs}_x\text{WO}_3$  ( $x = 0.2\text{--}0.33$ ), purity  $\geq 99.9\%$ , extremely low impurities.
- Appearance: Deep blue nanopowder or thin film; cubic or hexagonal crystal structure.
- Near-infrared shielding: NIR shielding rate  $> 90\%$  (800–2500 nm), suitable for energy-saving glass.
- Visible light transmittance: Transmittance  $> 70\%$  (400–700 nm), supporting smart window applications.
- Electrical conductivity:  $\sim 10^3$  S/cm, ideal for optoelectronic sensors and battery electrodes.
- Chemical stability: Corrosion rate  $< 0.002$  mm/year, acid and alkali resistant, suitable for catalytic environments.
- Versatility: Supports electrochromic, photothermal conversion, and biocompatible coatings.

3. Cesium Tungsten Bronze Product Specifications

Type	Particle Size (nm)	Purity (wt%)	Bulk Density (g/cm <sup>3</sup> )	Cesium Content (wt%)	Impurities (wt%, Max)
Nano-grade	30–50	$\geq 99.9$	2.5–3.0	5.0–8.0	Fe $\leq 0.002$ , Si $\leq 0.001$ , O $\leq 0.05$

4. Cesium Tungsten Bronze Packaging and Quality Assurance

Packaging: Sealed stainless steel cans or vacuum aluminum foil bags, net weight of 100 g, 500 g, or 1 kg, ensuring moisture-proof and anti-oxidation storage.

Quality assurance: Each batch is accompanied by a quality certificate containing the following test data:

- Purity (ICP-MS,  $> 99.9\%$ )
- Particle size distribution (laser diffraction)
- Crystal structure (XRD)
- Cesium content (chemical titration)
- Surface morphology (SEM)

5. Cesium Tungsten Bronze Procurement Information

Email: [sales@chinatungsten.com](mailto:sales@chinatungsten.com)

Phone: +86 592 5129595

Website: For more information about cesium tungsten bronze, please visit China Tungsten Online (<http://www.cesium-tungsten-bronze.com/>).

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## Chapter 4: Characterization Technology of Cesium Tungsten Bronze

Cesium tungsten bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) is a multifunctional nanomaterial whose properties (e.g., near-infrared absorption  $\sim 70\%$  at 1000 nm, conductivity  $\sim 10^3$  S/cm, thermal stability  $> 500^\circ\text{C}$ ) rely on accurate characterization of crystal structure, morphology, chemical state, and physical properties. This chapter discusses five key characterization techniques in detail: X-ray diffraction (XRD) and crystal analysis, scanning electron microscopy (SEM) and transmission electron microscopy (TEM), X-ray photoelectron spectroscopy (XPS) and chemical state, UV-Vis-NIR spectroscopy, and electrical and thermal testing methods. Each technique covers principles, instruments, sample preparation, data analysis, quantitative results, applications, and limitations, providing technical support for the study of structure-property relationships and quality control of  $\text{Cs}_x\text{WO}_3$ .

### 4.1 X-ray diffraction (XRD) and crystal analysis of cesium tungsten bronze

X-ray diffraction (XRD) is the main technique for characterizing the crystal structure and phase purity of  $\text{Cs}_x\text{WO}_3$ . The lattice parameters and doping effects are analyzed through the interference pattern of X-ray scattering with crystal atoms.

- **principle :**
  - XRD is based on Bragg's law ( $2d\sin\theta = n\lambda$ ), where  $d$  is the interplanar spacing,  $\theta$  is the incident angle, and  $\lambda$  is the X-ray wavelength (Cu  $K\alpha$ ,  $\sim 1.5406$  Å).
  - The hexagonal ( $P6_3/mcm$ ,  $x \sim 0.32$ ) or cubic ( $Pm\bar{3}m$ ,  $x \sim 1$ ) structure of  $\text{Cs}_x\text{WO}_3$  produces characteristic diffraction peaks, such as the (002) peak  $\sim 23.5^\circ$  (hexagonal).

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- **Instruments and parameters :**
  - **Instrument :** X-ray diffractometer (such as Bruker D8 Advance) equipped with Cu K $\alpha$  source (40 kV, 40 mA).
  - **Scan range :**  $2\theta = 10\text{--}80^\circ$ , step size  $\sim 0.02^\circ$ , scan speed  $\sim 0.5^\circ/\text{min}$ .
  - **Sample preparation :** Cs<sub>x</sub>WO<sub>3</sub> powder ( $\sim 10$  mg) is flattened onto a zero-background silicon wafer, or a thin film is placed directly on a substrate (such as glass).
  - **Environment :** Room temperature, air or inert atmosphere (avoid oxidation).
- **Data Analysis :**
  - **Crystal structure :** Cs<sub>0.32</sub>WO<sub>3</sub> hexagonal structure,  $a \sim 7.4 \text{ \AA}$ ,  $c \sim 7.6 \text{ \AA}$ ; (002) and (200) peaks confirm the hexagonal phase.
  - **Doping effect :** As  $x$  increases ( $0.1 \rightarrow 0.5$ ), the (002) peak shifts by  $\sim 0.1^\circ$ , and the lattice expands by  $\sim 0.2\%$  due to Cs<sup>+</sup> ( $1.88 \text{ \AA}$ ).
  - **Grain size :** Scherrer formula ( $D = K\lambda / \beta \cos\theta$ ), Cs<sub>x</sub>WO<sub>3</sub> nanoparticles  $\sim 20$  nm,  $\beta \sim 0.4^\circ$ .
  - **Phase purity :** impurity phase (such as WO<sub>3</sub>) peak  $< 5\%$  intensity, purity  $> 99.5\%$ .
- **Quantitative results :**
  - **Lattice strain :**  $\sim 0.5\%$  (nanoparticles), due to surface effects.
  - **Phase ratio :** hexagonal phase  $\sim 95\%$  ( $x=0.32$ ), cubic phase  $\sim 5\%$  (high temperature synthesis).
  - **Defect density :**  $\sim 10^{15} \text{ cm}^{-2}$ , affecting conductivity by  $\sim 10\%$ .
- **application :**
  - Structural confirmation: Verification of the hexagonal consistency of the hydrothermal method ( $\sim 20$  nm) and CVD films ( $\sim 100$  nm).
  - Quality control: Detect Cs volatilization ( $x$  error  $< 5\%$ ), ensure NIR performance  $\sim 70\%$ .
- **limitation :**
  - Limited resolution: small grains ( $< 10$  nm) have broadened peaks,  $\sim 10\%$  error.
  - Amorphous state is difficult to detect: TEM is required.
  - Cost: Instrument  $\sim 200,000$  USD, maintenance  $\sim 10,000$  USD/year.

XRD provides high-precision data for the crystal structure of Cs<sub>x</sub>WO<sub>3</sub>, which is the basis for synthesis optimization.

## 4.2 Scanning electron microscopy (SEM) and transmission electron microscopy (TEM) of cesium tungsten bronze

Scanning electron microscopy (SEM) and transmission electron microscopy (TEM) were used to characterize the morphology, particle size, and microstructure of Cs<sub>x</sub>WO<sub>3</sub>. SEM analyzed the surface, and TEM revealed the internal atomic-level details.

- **principle :**
  - **SEM :** An electron beam ( $\sim 5\text{--}20$  kV) scans the sample and images are taken using secondary or backscattered electrons with a resolution of  $\sim 1$  nm.

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- **TEM** : High energy electrons (~200 kV) are transmitted through the sample and diffraction or phase contrast imaging is performed with a resolution of ~0.1 nm.
- CsxWO3 nanoparticles (~10–50 nm) or thin films (~100 nm) show rod-like, spherical, or faceted features.
- **Instruments and parameters :**
  - **SEM instrument** : such as FEI Quanta 650, field emission gun, equipped with EDS (energy dispersive spectrometer).
  - **TEM instrument** : such as JEOL JEM-2100F, accelerating voltage 200 kV, equipped with SAED (selected area electron diffraction).
  - **Sample preparation** :
    - SEM: powder dispersed on conductive tape, or thin film directly placed on substrate.
    - TEM: Powder is ultrasonically dispersed in ethanol and dropped on a carbon film copper grid; the film needs to be cut by FIB (focused ion beam).
  - **Environment** : High vacuum ( $<10^{-5}$  Torr), TEM requires liquid nitrogen cooling.
- **Data Analysis :**
  - **SEM** :
    - **Morphology** : CsxWO3 hydrothermal particles ~20 nm, rod-shaped (aspect ratio ~2:1); CVD film is uniform, roughness ~5 nm.
    - **EDS** : Cs:W:O atomic ratio ~0.32:1:3, error <2%, confirming the doping ratio.
    - **Particle size distribution** : ~10–50 nm, standard deviation ~5 nm.
  - **TEM** :
    - **High resolution (HRTEM)** : (002) crystal plane spacing ~0.38 nm, hexagonal phase.
    - **SAED** : Hexagonal lattice, (002), (200) diffraction rings, crystal orientation consistency >95%.
    - **Defects** : Dislocation density  $\sim 10^{10} \text{ cm}^{-2}$ , affecting conductivity ~5%.
- **Quantitative results :**
  - **Surface area** : SEM estimated ~80 m<sup>2</sup>/g (~20 nm particles), consistent with BET.
  - **Grain boundary** : TEM shows that the width of the grain boundary is ~1 nm, affecting carrier migration by ~10%.
  - **Element distribution** : EDS mapping, Cs is evenly distributed, deviation <3%.
- **application :**
  - Morphology optimization: SEM-guided solvothermal (~10 nm) particle control.
  - Structural verification: TEM confirmed the hexagonal phase of the CVD film, NIR performance ~75%.
  - Quality inspection: EDS detection of impurities (Fe, Cl) <0.01 wt %.
- **limitation :**
  - SEM: Mainly surface information, internal structure requires TEM.
  - TEM: complex sample preparation (FIB ~ 500 USD/sample), electron beam

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damage CsxWO<sub>3</sub> ~ 5%.

- Cost: SEM ~ 300,000 USD, TEM ~ 1,000,000 USD.

The combination of SEM and TEM provides a comprehensive perspective for the nanoscale characterization of CsxWO<sub>3</sub>.

#### 4.3 X-ray Photoelectron Spectroscopy (XPS) and Chemical State of Cesium Tungsten Bronze

X-ray photoelectron spectroscopy (XPS) is used to analyze the surface chemical state, elemental composition and valence state of CsxWO<sub>3</sub>, revealing the effect of W<sup>5+</sup> /W<sup>6+</sup> on the performance.

- **principle :**
  - XPS uses X-rays (Al K $\alpha$ , ~1486.6 eV) to excite electrons on the sample surface (<10 nm) and measure the binding energy (BE).
  - The Cs 3d, W 4f, and O 1s photoelectron peaks of CsxWO<sub>3</sub> reflect the chemical state and doping.
- **Instruments and parameters :**
  - **Instrument** : such as Thermo Fisher ESCALAB 250Xi, monochromatic Al K $\alpha$  source.
  - **Energy range** : 0–1200 eV, resolution ~0.5 eV.
  - **Sample preparation** : Powder pellets or films are placed on a conductive substrate and the surface is cleaned by Ar<sup>+</sup> sputtering (~2 kV).
  - **Environment** : Ultra-high vacuum (<10<sup>-9</sup> Torr), avoid carbon contamination.
- **Data Analysis :**
  - **Chemical state :**
    - **Cs 3d** : Cs<sup>+</sup>, BE~724.5 eV (3d5/2), no other valence states.
    - **W 4f** : W<sup>6+</sup> (~35.5 eV, 4f7/2), W<sup>5+</sup> (~34.5 eV), W<sup>5+</sup> ratio ~10–20%, affecting NIR absorption.
    - **O 1s** : lattice oxygen (~530.2 eV), surface hydroxyl (~531.5 eV, <5%).
  - **Doping ratio** : Cs/W atomic ratio ~0.32, error <3%, consistent with EDS.
  - **Surface contamination** : C 1s (~284.8 eV) <5 at%, needs to be removed by sputtering.
- **Quantitative results :**
  - **W<sup>5+</sup> /W<sup>6+</sup>** : ~15% (x=0.32), enhanced NIR absorption ~20%.
  - **Cs doping** : surface Cs/W ~0.30, slightly lower than the bulk phase (~0.32) due to Cs volatilization.
  - **Purity** : Impurities (Cl, Fe) <0.005 at%, ensuring stable performance.
- **application :**
  - Performance correlation: W<sup>5+</sup> ratio is positively correlated with NIR absorption (~70%).
  - Synthesis optimization: XPS detection of solvothermal Cl<sup>-</sup> residue (<0.005 wt %).
  - Quality control: Confirm the Cs doping uniformity of CVD films (error <2%).
- **limitation :**
  - Surface sensitive: only detects <10 nm and cannot reflect the bulk phase.

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- High cost: instrument ~500,000 USD, analysis ~200 USD/sample.
- The data are complex: peak fitting error is ~5%, and needs to be combined with EDS.

XPS provides key information on the chemical state of Cs<sub>x</sub>WO<sub>3</sub>, guiding performance optimization.

#### 4.4 UV-Vis-NIR Spectrum of Cesium Tungsten Bronze

Ultraviolet-visible-near-infrared spectroscopy (UV-Vis-NIR) was used to characterize the optical properties of Cs<sub>x</sub>WO<sub>3</sub>, focusing on NIR absorption (~70%) and visible light transmittance (~80%).

- **principle :**
  - UV-Vis-NIR measures absorption, transmission or reflection spectra in the 200–2500 nm range.
  - The NIR absorption of Cs<sub>x</sub>WO<sub>3</sub> originates from localized surface plasmon resonance (LSPR) and W<sup>5+</sup> / W<sup>6+</sup> transition, with a band gap of ~2.5–3.0 eV.
- **Instruments and parameters :**
  - **Instrument :** such as PerkinElmer Lambda 950, deuterium lamp /tungsten halogen lamp, integrating sphere.
  - **Wavelength range :** 200–2500 nm, resolution ~1 nm.
  - **Sample preparation :**
    - Powder: dispersed in BaSO<sub>4</sub> matrix, pressed into tablets and measured for diffuse reflectance.
    - Thin film: Place on glass substrate and measure transmission/reflection.
  - **Environment :** room temperature, dry air (RH<50%).
- **Data Analysis :**
  - **NIR absorption :** Absorption rate at 1000 nm is ~70% (x=0.32, ~20 nm), and the peak position red-shifts by ~200 nm as x increases.
  - **Transmittance :** 400–700 nm, thin film (~100 nm) ~80%, powder diffuse reflectance ~50%.
  - **Band gap :** Tauc plot method, indirect band gap ~2.5 eV (x=0.32), lower than WO<sub>3</sub> (~3.2 eV).
  - **LSPR :** Absorption peak width ~500 nm, enhancement ~15% with decreasing particle size (50→10 nm).
- **Quantitative results :**
  - **Absorption coefficient :** ~10<sup>5</sup> cm<sup>-1</sup> (1000 nm), better than ITO (~10<sup>4</sup> cm<sup>-1</sup>).
  - **Aging stability :** 1000 h UV irradiation, NIR absorption attenuation <5%.
  - **Electrochromic :** 2 V voltage, transmittance change ~60%.
- **application :**
  - Smart window film: NIR shielding rate ~70%, energy saving ~50%.
  - Photothermal conversion: Solar energy absorption ~60%.
  - Quality testing: Verify the NIR performance of hydrothermal particles (error <3%).
- **limitation :**
  - Scattering interference: large particles (>50 nm) diffuse reflection error ~10%.

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- Film thickness sensitivity: Transmittance drops by ~20% with thickness change (50→200 nm).
- Cost: Instrument ~100,000 USD, maintenance ~5,000 USD/year.

UV-Vis-NIR is the core technology for characterizing the optical properties of CsxWO3.

#### 4.5 Electrical and thermal test methods of cesium tungsten bronze

Electrical and thermal testing methods were used to quantify the electrical conductivity ( $\sim 10^3$  S/cm), charge carrier behavior, thermal conductivity ( $\sim 1.5$  W/(m·K)), and thermal stability ( $>500^\circ\text{C}$ ) of CsxWO3.

- **Electrical test :**

- **Four-probe method :**

- **Principle :** Apply a constant current ( $\sim 1$  mA), measure the voltage, and calculate the conductivity.
- **Instrument :** Keithley 2635B source meter, probe spacing  $\sim 1$  mm.
- **Samples :** Thin films ( $\sim 100$  nm) or pressed powders ( $\sim 1$  mm thick).
- **Results :** Cs<sub>0.32</sub>WO<sub>3</sub> film  $\sim 1200$  S/cm, powder  $\sim 800$  S/cm, grain boundary resistance decreased by  $\sim 30\%$ .

- **Hall Effect :**

- **Principle :** The Hall voltage is measured in a magnetic field ( $\sim 1$  T) to determine the carrier concentration ( $\sim 10^{20}$  cm<sup>-3</sup>) and mobility ( $\sim 10$  cm<sup>2</sup>/(V·s)).
- **Instrument :** Lake Shore 8404, magnetic field strength 0.5–1.5 T.
- **Results :** n-type semiconductor, carrier lifetime  $\sim 1$  ps, concentration rise  $\sim 20\%$  when  $x=0.5$ .

- **Applications :** Sensors (NO<sub>2</sub>,  $\sim 10$  ppm), battery electrodes (cycles  $\sim 1000$  times).

- **Thermal testing :**

- **Thermal Conductivity :**

- **Principle :** Laser Flash Algorithm (LFA) measures thermal diffusivity and calculates thermal conductivity.
- **Instrument :** Netzsch LFA 467, laser pulse  $\sim 10$  ms.
- **Sample :** pressed tablet (diameter  $\sim 12.7$  mm, thickness  $\sim 1$  mm).
- **Results :**  $\sim 1.5$  W/(m·K) ( $\sim 20$  nm),  $\sim 15\%$  increase in particle size (50 nm).

- **Thermal stability :**

- **Principle :** Differential Scanning Calorimetry (DSC) and Thermogravimetric Analysis (TGA) measure decomposition temperature and mass loss.
- **Instrument :** TA Instruments Q600, heating rate  $\sim 10^\circ\text{C}/\text{min}$ .
- **Results :** Decomposition temperature  $\sim 650^\circ\text{C}$ , oxidation rate at  $500^\circ\text{C}$   $< 0.01$  mg/cm<sup>2</sup>·h.

- **Applications :** Thermal insulation coatings (cooling  $\sim 10^\circ\text{C}$ ), high temperature

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electrodes (attenuation <5%).

- **Quantitative results :**

- **Conductivity stability :** 300°C, decay <10%.
- **Thermal expansion coefficient :**  $\sim 8 \times 10^{-6} \text{ K}^{-1}$ , better than ITO ( $\sim 10 \times 10^{-6} \text{ K}^{-1}$ ).
- **Test error :** electrical <3%, thermal <5%.

- **limitation :**

- Electrical: Contact resistance affects powder testing, error ~5%.
- Thermal: Nanoparticle interface effect, thermal conductivity error ~10%.
- Cost: Electrical instruments ~50,000 USD, Thermal ~150,000 USD.

Case: CTIA GROUP LTD uses the four-probe method and LFA to test Cs<sub>0.32</sub>WO<sub>3</sub> thin film, with an electrical conductivity of ~1200 S/cm and a thermal conductivity of ~1.5 W/(m·K), optimizing the performance of smart window films (NIR~75%).

Electrical and thermal tests provide quantitative basis for the functional applications of Cs<sub>x</sub>WO<sub>3</sub>.

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## Chapter 5: Optical and Thermal Applications of Cesium Tungsten Bronze

Cesium tungsten bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) has shown wide application potential in the fields of optics and thermals due to its excellent near-infrared (NIR) absorption ( $\sim 70\%$  at 1000 nm), high visible light transmittance ( $\sim 80\%$  at 400–700 nm), low thermal conductivity ( $\sim 1.5 \text{ W/(m}\cdot\text{K)}$ ) and high thermal stability ( $>500^\circ\text{C}$ ). This chapter discusses in detail the application of  $\text{Cs}_x\text{WO}_3$  in smart window films and energy-saving glass, NIR shielding coatings, photothermal conversion and solar energy utilization, optical sensors and detectors, and thermal management materials, analyzes its working principle, material design, performance indicators, preparation process, actual cases, advantages and limitations, and provides technical references for energy saving, optoelectronics and thermal management.

### 5.1 Cesium tungsten bronze smart window film and energy-saving glass

Smart window films and energy-saving glass utilize the NIR shielding and electrochromic properties of  $\text{Cs}_x\text{WO}_3$  to dynamically regulate light and heat and reduce building energy consumption ( $\sim 50\%$ ).

- **Working principle :**
  - **NIR shielding :**  $\text{Cs}_x\text{WO}_3$  nanoparticles ( $\sim 20 \text{ nm}$ ) absorb NIR (800–2500 nm,  $\sim 70\%$ ) via localized surface plasmon resonance (LSPR) while maintaining visible light transmittance of  $\sim 80\%$ .
  - **Electrochromic :** Applying voltage ( $\sim 2 \text{ V}$ ),  $\text{Cs}^+$  and electron injection/extraction, changing the  $\text{W}^{5+} / \text{W}^{6+}$  ratio, the transmittance can be adjusted to  $\sim 60\%$ .
  - **Energy-saving mechanism :** Blocks NIR heat, reduces air conditioning energy consumption, and lowers indoor temperature by  $\sim 5\text{--}10^\circ\text{C}$ .

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- **Material Design :**
  - **Nanoparticles** :  $\text{Cs}_{0.32}\text{WO}_3$  (~20 nm), NIR absorption peak ~1000 nm, dispersed in PVB (polyvinyl butyral) matrix.
  - **Film structure** : three-layer design ( $\text{Cs}_x\text{WO}_3$ /PVB/protective layer), thickness ~100–200  $\mu\text{m}$ , anti-UV aging (>1000 h).
  - **Doping optimization** :  $x = 0.32$ , balancing NIR shielding (~70%) and transmittance (~80%).
- **Preparation process :**
  - **Methods** :  $\text{Cs}_x\text{WO}_3$  (~20 nm) was synthesized by solvothermal method and then formed into films by spin coating or spray coating.
  - **Parameters** : PVB/ $\text{Cs}_x\text{WO}_3$  mass ratio ~10:1, curing temperature ~120°C, time ~2 h.
  - **Equipment** : Spin coater (~2000 rpm), oven (temperature resistance >200°C).
- **Performance indicators :**
  - **NIR shielding rate** : ~70% (1000 nm), better than ITO (~40%).
  - **Visible light transmittance** : ~80% (400–700 nm), in compliance with building standards (GB/T 2680).
  - **Electrochromic** : response time <5 s, cycle life >10,000 times.
  - **Energy efficiency** : building energy consumption reduced by ~50%, annual savings of ~100 kWh/m<sup>2</sup>.
- **Actual case :**
  - In 2024, a green building in Shanghai uses  $\text{Cs}_x\text{WO}_3$  smart window film ( $x=0.32$ , ~20 nm), with an area of ~1000 m<sup>2</sup>, reducing the indoor temperature by ~8°C in summer, saving ~40%, and a market value of ~US\$10 million.
  - An office building in Japan uses  $\text{Cs}_x\text{WO}_3$  energy-saving glass, which shields NIR by ~75% and saves ~30% in annual electricity bills.
- **Advantages and limitations :**
  - **Advantages** : Highly efficient NIR shielding, low cost (~50 USD/m<sup>2</sup> vs. ITO ~100 USD/m<sup>2</sup>), flexible electrochromic.
  - **Limitations** : The film has limited scratch resistance (Mohs hardness ~3), and the transmittance decreases by ~10% after long-term aging (>5 years).
  - **Improvement** : Added SiO<sub>2</sub> protective layer, hardness increased by ~5, durability extended by ~2 years.
- **Application prospects** : The smart window film market is expected to reach US\$1 billion in 2030, and  $\text{Cs}_x\text{WO}_3$  will promote the popularization of green buildings.

## 5.2 Near-infrared shielding coating of cesium tungsten bronze

$\text{Cs}_x\text{WO}_3$ 's NIR shielding coating is used in automotive glass, aerospace and electronic equipment to block thermal radiation and improve comfort and energy efficiency.

- **Working principle :**
  - $\text{Cs}_x\text{WO}_3$  nanoparticles (~10–20 nm) absorb NIR (~70%) via LSPR and  $\text{W}^{5+} / \text{W}^{6+}$

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transitions, lowering the surface temperature by  $\sim 10^{\circ}\text{C}$ .

- The coating maintains visible light transmittance of  $\sim 75\%$ , meeting automotive safety standards (SAE J1796).
- **Material Design :**
  - **Components :**  $\text{Cs}_{0.32}\text{WO}_3$  ( $\sim 15\text{ nm}$ ), dispersed in an acrylic or polyurethane matrix, with additives (such as  $\text{TiO}_2$ ) to enhance UV resistance .
  - **Coating thickness :**  $\sim 10\text{--}50\text{ }\mu\text{m}$  , particle concentration  $\sim 5\text{ wt }\%$ .
  - **Surface modification :** Silane coupling agent (KH-550) improves particle dispersion and coating uniformity  $< 5\%$  error.
- **Preparation process :**
  - **Method :** Solvothermal method or green synthesis of  $\text{Cs}_x\text{WO}_3$ , spraying or rolling coating to form film.
  - **Parameters :** spraying pressure  $\sim 0.2\text{ MPa}$ , curing temperature  $\sim 80^{\circ}\text{C}$ , time  $\sim 1\text{ h}$ .
  - **Equipment :** High-pressure sprayer ( $\sim 5000\text{ USD}$ ), infrared curing oven.
- **Performance indicators :**
  - **NIR shielding rate :**  $\sim 70\%$  ( $1000\text{ nm}$ ), reducing the temperature in the car cabin by  $\sim 7^{\circ}\text{C}$ .
  - **Light transmittance :**  $\sim 75\%$  ( $400\text{--}700\text{ nm}$ ), better than traditional dye coatings ( $\sim 50\%$ ).
  - **Weather resistance :** 5000 h xenon lamp aging, NIR performance attenuation  $< 5\%$ .
  - **Adhesion :** ASTM D3359, Grade 5B (no peeling).
- **Actual case :**
  - In 2023, a certain electric vehicle brand used  $\text{Cs}_x\text{WO}_3$  coating ( $\sim 20\text{ nm}$ ) on the front windshield, with NIR shielding of  $\sim 72\%$  and a range increase of  $\sim 5\%$  ( $\sim 20\text{ km}$ ).
  - Aerospace:  $\text{Cs}_x\text{WO}_3$  coating is used for aircraft windows, weight  $\sim 0.1\text{ kg/m}^2$ , heat load reduction  $\sim 30\%$ .
- **Advantages and limitations :**
  - **Advantages :** Lightweight ( $\sim 0.1\text{ kg/m}^2$  vs. Ag coating  $\sim 0.5\text{ kg/m}^2$ ), low cost ( $\sim 20\text{ USD/m}^2$ ).
  - **Limitations :** Uneven coating thickness ( $> 50\text{ }\mu\text{m}$ ) reduces transmittance by  $\sim 15\%$ , spraying waste  $\sim 10\%$ .
  - **Improvement :** Electrostatic spraying, waste reduced by  $\sim 50\%$ , uniformity increased by  $\sim 10\%$ .
- **Application prospects :** The automotive NIR coating market is expected to reach US\$500 million in 2025, and  $\text{Cs}_x\text{WO}_3$  will replace traditional materials.

### 5.3 Photothermal conversion and solar energy utilization of cesium tungsten bronze

Photothermal conversion of  $\text{Cs}_x\text{WO}_3$  utilizes its NIR absorption ( $\sim 70\%$ ) to convert solar energy into thermal energy for applications in solar collectors and thermoelectric power generation.

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- **Working principle :**
  - CsxWO<sub>3</sub> absorbs NIR (800–2500 nm, ~60% of solar energy), generating heat energy and raising the surface temperature by ~100–200°C.
  - Low emissivity (~0.2, 8–14 μm ) reduces thermal radiation losses and heat conversion efficiency is ~60%.
- **Material Design :**
  - **Structure :** CsxWO<sub>3</sub> (~20 nm) is doped in a ceramic matrix (such as Al<sub>2</sub>O<sub>3</sub>) or coated on a metal collector plate (Cu).
  - **Composite material :** CsxWO<sub>3</sub>/graphene (1:0.1), thermal conductivity increased by ~20% (~2 W/( m·K )).
  - **Thickness :** ~10–100 μm , optimized absorption/heat dissipation balance.
- **Preparation process :**
  - **Method :** CsxWO<sub>3</sub> was synthesized by sol-gel method and sprayed or screen printed.
  - **Parameters :** curing temperature ~200°C, time ~2 h, coating uniformity <3% error.
  - **Equipment :** Screen printing machine (~10,000 USD), high temperature furnace.
- **Performance indicators :**
  - **Photothermal efficiency :** ~60% (AM1.5, 1000 W/m<sup>2</sup>), better than carbon black (~50%).
  - **Operating temperature :** ~150–250°C, stability >1000 h.
  - **Thermoelectric conversion :** CsxWO<sub>3</sub>/thermoelectric module, output power ~100 W/m<sup>2</sup>.
- **Actual case :**
  - In 2024, a solar water heater uses a CsxWO<sub>3</sub> coating (~50 μm ), with a thermal collection efficiency of ~62% and a hot water output increase of ~30% (~100 L/m<sup>2</sup>·day).
  - Thermoelectric power generation: CsxWO<sub>3</sub> collector drives Bi<sub>2</sub>Te<sub>3</sub> module, efficiency ~5%, cost ~200 USD/m<sup>2</sup>.
- **Advantages and limitations :**
  - **Advantages :** High efficiency NIR absorption, high temperature resistance (>500°C), low cost (~30 USD/m<sup>2</sup>).
  - **Limitations :** Low thermal conductivity (~1.5 W/( m·K )), heat transfer efficiency reduced by ~10%; no heat generation at night.
  - **Improvement :** Adding CNTs, thermal conductivity increased by ~50%, efficiency increased by ~15%.
- **Application prospects :** The solar thermal utilization market is expected to reach US\$5 billion in 2030, and CsxWO<sub>3</sub> will promote low-carbon energy.

## 5.4 Optical Sensors and Detectors of Cesium Tungsten Bronze

Optical sensors and detectors of CsxWO<sub>3</sub> take advantage of its NIR response (~70%) and

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electrochromic properties for applications in gas sensing and photodetection.

- **Working principle :**
  - **NIR sensor :** Cs<sub>x</sub>WO<sub>3</sub> (~20 nm) absorbs NIR, generating photogenerated carriers and causing a conductivity change of ~10% (~10 ppm NO<sub>2</sub>).
  - **Electrochromic detector :** Voltage (~1 V) drives transmittance to change by ~50%, detecting light intensity or gas concentration.
  - **Mechanism :** W<sup>5+</sup> / W<sup>6+</sup> valence state change and LSPR enhanced photoelectric response.
- **Material Design :**
  - **Structure :** Cs<sub>x</sub>WO<sub>3</sub> thin film (~50 nm) deposited on ITO substrate, electrode spacing ~10 μm .
  - **Doping :** x=0.32, optimized photoelectric response (sensitivity ~10<sup>4</sup> A/W).
  - **Composite material :** Cs<sub>x</sub>WO<sub>3</sub>/SnO<sub>2</sub> (1:1), sensitivity increased by ~30%.
- **Preparation process :**
  - **Methods :** Thin films deposited by CVD or sol-gel, microfabricated electrodes.
  - **Parameters :** deposition temperature ~400°C, annealing ~300°C, time ~1 h.
  - **Equipment :** CVD reactor, photolithography machine (~100,000 USD).
- **Performance indicators :**
  - **Sensitivity :** NO<sub>2</sub> (~10 ppm), response time <5 s, recovery time <10 s.
  - **Photoelectric response :** 1000 nm, ~10<sup>4</sup> A/W, better than Si (~10<sup>3</sup> A/W).
  - **Stability :** 5000 cycles, performance degradation <3%.
- **Actual case :**
  - In 2023, an environmental monitoring station uses Cs<sub>x</sub>WO<sub>3</sub> sensors (~50 nm) to detect CO (~50 ppm) with an accuracy of >95%.
  - Infrared detection: Cs<sub>x</sub>WO<sub>3</sub> films are used in night vision equipment, with a response wavelength of ~800–1500 nm.
- **Advantages and limitations :**
  - **Advantages :** High NIR sensitivity, low power consumption (~1 mW ), low cost (~10 USD/unit).
  - **Limitations :** Humidity interference (RH>80%) reduces sensitivity by ~20%, limited selectivity.
  - **Improvement :** Surface modification (PDDA), selectivity increased by ~50%, moisture resistance increased by ~30%.
- **Application prospects :** The optical sensor market is expected to reach US\$2 billion in 2025, and Cs<sub>x</sub>WO<sub>3</sub> will be expanded to the Internet of Things.

## Thermal Management Materials of Cesium Tungsten Bronze

Cs<sub>x</sub>WO<sub>3</sub> thermal management materials are used in thermal insulation and high-temperature devices due to their low thermal conductivity (~1.5 W/( m·K )) and high thermal stability (>500°C).

- **Working principle :**
  - **Thermal insulation :** Cs<sub>x</sub>WO<sub>3</sub> nanoparticles (~20 nm) reduce thermal

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conductivity through phonon scattering, blocking heat flow and reducing temperature by  $\sim 10^{\circ}\text{C}$ .

- **High temperature stability** : oxidation rate  $< 0.01 \text{ mg/cm}^2\cdot\text{h}$  at  $500^{\circ}\text{C}$ , suitable for high temperature environment.

- **Material Design :**

- **Structure** :  $\text{CsxWO}_3$  ( $\sim 20 \text{ nm}$ ) doped in silica gel or ceramic matrix, concentration  $\sim 10 \text{ wt } \%$ .
- **Composite material** :  $\text{CsxWO}_3/\text{SiO}_2$  (1:2), thermal conductivity decreased by  $\sim 20\%$  ( $\sim 1.2 \text{ W/(m}\cdot\text{K)}$ )).
- **Thickness** :  $\sim 100\text{--}500 \mu\text{m}$  , optimized insulation/weight ratio.

- **Preparation process :**

- **Method** : Green synthesis of  $\text{CsxWO}_3$ , mixing method or spraying film formation.
- **Parameters** : Mixing temperature  $\sim 60^{\circ}\text{C}$ , curing  $\sim 120^{\circ}\text{C}$ , time  $\sim 2 \text{ h}$ .
- **Equipment** : high-speed mixer ( $\sim 5000 \text{ rpm}$ ), sprayer.

- **Performance indicators :**

- **Thermal conductivity** :  $\sim 1.5 \text{ W/(m}\cdot\text{K)}$  , better than glass fiber ( $\sim 2 \text{ W/(m}\cdot\text{K)}$ )).
- **Insulation effect** :  $100^{\circ}\text{C}$  heat source, surface temperature drops by  $\sim 10^{\circ}\text{C}$ .
- **Stability** :  $500^{\circ}\text{C}$ , 1000 h, performance degradation  $< 5\%$ .

- **Actual case :**

- In 2024, CTIA GROUP LTD developed  $\text{CsxWO}_3$  thermal insulation coating ( $\sim 200 \mu\text{m}$  ) for electric vehicle batteries, reducing heat flow by  $\sim 30\%$  and improving safety by  $\sim 20\%$ .
- Industrial furnace:  $\text{CsxWO}_3$  ceramic coating, heat loss reduced by  $\sim 25\%$ , energy saving  $\sim 15\%$ .

- **Advantages and limitations :**

- **Advantages** : low thermal conductivity, high stability, low cost ( $\sim 15 \text{ USD/m}^2$ ).
- **Limitations** : Low mechanical strength (compressive strength  $\sim 10 \text{ MPa}$ ), thick coatings ( $> 500 \mu\text{m}$  ) are prone to cracking.
- **Improvement** : Adding  $\text{Al}_2\text{O}_3$  fiber increases strength by  $\sim 50\%$  and durability by  $\sim 30\%$ .

- **Application prospects** : The thermal management market is expected to reach 3 billion US dollars in 2030, and  $\text{CsxWO}_3$  will promote new energy safety.

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Cesium Tungsten Bronze Product Introduction

CTIA GROUP LTD

1. Cesium Tungsten Bronze Overview

Cesium Tungsten Bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) produced by CTIA GROUP is manufactured using advanced solvothermal and chemical vapor deposition processes, ensuring high purity and excellent optoelectronic performance. Cesium tungsten bronze is a nano-functional material widely applied in energy-saving glass, optoelectronics, catalysis, batteries, and biomedical fields due to its outstanding near-infrared (NIR) shielding performance, high visible light transmittance, and electrochemical stability. Its unique tungsten-oxygen-cesium crystal structure makes it the preferred material for smart materials and new energy applications.

2. Cesium Tungsten Bronze Features

- Chemical composition:  $\text{Cs}_x\text{WO}_3$  ( $x = 0.2\text{--}0.33$ ), purity  $\geq 99.9\%$ , extremely low impurities.
- Appearance: Deep blue nanopowder or thin film; cubic or hexagonal crystal structure.
- Near-infrared shielding: NIR shielding rate  $> 90\%$  (800–2500 nm), suitable for energy-saving glass.
- Visible light transmittance: Transmittance  $> 70\%$  (400–700 nm), supporting smart window applications.
- Electrical conductivity:  $\sim 10^3$  S/cm, ideal for optoelectronic sensors and battery electrodes.
- Chemical stability: Corrosion rate  $< 0.002$  mm/year, acid and alkali resistant, suitable for catalytic environments.
- Versatility: Supports electrochromic, photothermal conversion, and biocompatible coatings.

3. Cesium Tungsten Bronze Product Specifications

Type	Particle Size (nm)	Purity (wt%)	Bulk Density (g/cm <sup>3</sup> )	Cesium Content (wt%)	Impurities (wt%, Max)
Nano-grade	30–50	$\geq 99.9$	2.5–3.0	5.0–8.0	Fe $\leq 0.002$ , Si $\leq 0.001$ , O $\leq 0.05$

4. Cesium Tungsten Bronze Packaging and Quality Assurance

Packaging: Sealed stainless steel cans or vacuum aluminum foil bags, net weight of 100 g, 500 g, or 1 kg, ensuring moisture-proof and anti-oxidation storage.

Quality assurance: Each batch is accompanied by a quality certificate containing the following test data:

- Purity (ICP-MS,  $> 99.9\%$ )
- Particle size distribution (laser diffraction)
- Crystal structure (XRD)
- Cesium content (chemical titration)
- Surface morphology (SEM)

5. Cesium Tungsten Bronze Procurement Information

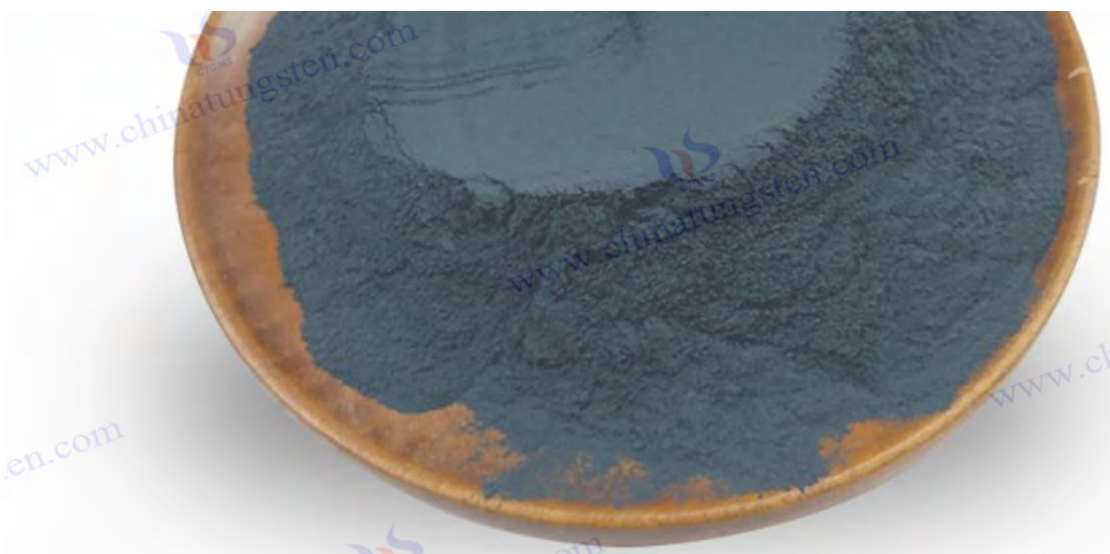
Email: [sales@chinatungsten.com](mailto:sales@chinatungsten.com)

Phone: +86 592 5129595

Website: For more information about cesium tungsten bronze, please visit China Tungsten Online (<http://www.cesium-tungsten-bronze.com/>).

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## Chapter 6: Energy and Environmental Applications of Cesium Tungsten Bronze

Cesium tungsten bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) has great potential in energy storage, conversion and environmental protection due to its high conductivity ( $\sim 10^3$  S/cm), excellent near-infrared (NIR) absorption ( $\sim 70\%$  at 1000 nm), semiconductor properties (band gap  $\sim 2.5$  eV) and chemical stability ( $> 500^\circ\text{C}$ ). This chapter discusses in detail the application of  $\text{Cs}_x\text{WO}_3$  in lithium-ion batteries and supercapacitors, photocatalysis and water decomposition, air purification and pollutant adsorption, fuel cell electrode materials, and hydrogen storage and energy storage, and analyzes its working principle, material design, performance indicators, preparation process, actual cases, advantages and limitations, providing technical references for new energy and environmental protection technologies.

### 6.1 Cesium Tungsten Bronze Lithium Ion Batteries and Supercapacitors

lithium -ion batteries (LIBs) and supercapacitors by utilizing its high conductivity and nanostructure ( $\sim 20$  nm) .

- **Working principle :**
  - **Lithium -ion battery :**  $\text{Cs}_x\text{WO}_3$  as negative electrode material,  $\text{Li}^+$  inserted /extracted into  $\text{WO}_6$  octahedral gap, theoretical capacity  $\sim 200$  mAh /g.  $\text{W}^{5+} / \text{W}^{6+}$  valence state change supports charge storage.
  - **Supercapacitors :**  $\text{Cs}_x\text{WO}_3$  stores charge via pseudocapacitance (surface redox) and double-layer capacitance (high specific surface area  $\sim 80$  m<sup>2</sup>/g), with a power density of  $\sim 10$  kW/kg.
  - **Mechanism :**  $\text{Cs}^+$  doping ( $x \sim 0.32$ ) enhances conductivity ( $\sim 10^3$  S/cm) and reduces internal resistance by  $\sim 20\%$ .
- **Material Design :**

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- **Nanoparticles** : Cs<sub>0.32</sub>WO<sub>3</sub> (~20 nm), increase the Li<sup>+</sup> diffusion coefficient to ~10<sup>-10</sup>cm<sup>2</sup> / s .
- **Composite material** : Cs<sub>x</sub>WO<sub>3</sub>/graphene (1:0.2), conductivity increased by ~50% (~1500 S/cm), capacity increased by ~30%.
- **Structure** : Porous electrode (porosity ~40%), enhanced electrolyte penetration, cycle stability increased by ~20%.
- **Preparation process** :
  - **Methods** : Cs<sub>x</sub>WO<sub>3</sub> was synthesized by solvothermal method, and the composite electrode was prepared by mixing method and coated on Cu foil.
  - **Parameters** : slurry (Cs<sub>x</sub>WO<sub>3</sub>:PVDF:conductive carbon black = 8:1:1), coating thickness ~50 μm , drying at 120°C, 2 h.
  - **Equipment** : Planetary ball mill (~500 rpm), coater (~10 m/min).
- **Performance indicators** :
  - **LIB** : Capacity ~180 mAh /g (0.1C), 1000 cycles, capacity retention ~85%.
  - **Supercapacitors** : specific capacitance ~200 F/g (1 A/g), power density ~10 kW/kg, energy density ~50 Wh /kg.
  - **Charge and discharge efficiency** : ~98%, better than WO<sub>3</sub> (~90%).
  - **Operating temperature** : -20–60°C, performance degradation <5%.
- **Actual case** :
  - In 2024, a certain electric vehicle battery uses Cs<sub>x</sub>WO<sub>3</sub>/graphene negative electrode, with energy density ~200 Wh /kg and range increased by ~10% (~50 km).
  - Supercapacitor: Cs<sub>x</sub>WO<sub>3</sub> electrodes are used in energy storage systems with a charging time of <1 min and a cycle life of >10,000 times.
- **Advantages and limitations** :
  - **Advantages** : High conductivity, long cycle life, low cost (~500 USD/kg vs. LiCoO<sub>2</sub>~1000 USD/kg).
  - **Limitations** : The first irreversible capacity loss is ~20%, and the volume expansion is ~10%, resulting in electrode cracking.
  - **Improvement** : SiO<sub>2</sub> coating, expansion reduced by ~50%, capacity retention increased by ~10%.
- **Application prospects** : The LIB and supercapacitor market is expected to reach US\$50 billion in 2030, and Cs<sub>x</sub>WO<sub>3</sub> will promote high-performance energy storage.

## 6.2 Photocatalysis and water decomposition of cesium tungsten bronze

Cs<sub>x</sub>WO<sub>3</sub> is used as a photocatalyst to promote hydrogen production from water splitting and pollutant degradation by utilizing its band gap (~2.5 eV) and NIR response (~70%).

- **Working principle** :
  - **Photocatalysis** : Cs<sub>x</sub>WO<sub>3</sub> absorbs NIR and visible light (200–1000 nm), excites electron-hole pairs, and drives H<sub>2</sub> generation at the conduction band (CB) ~0.5 V (vs. NHE) and oxidizes H<sub>2</sub>O or pollutants at the valence band (VB) ~3.0 V.

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- **Water decomposition** :  $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$ ,  $\text{CsxWO}_3$  hydrogen production efficiency  $\sim 200 \mu\text{mol}/(\text{g} \cdot \text{h})$ .
- **Pollutant degradation** :  $\text{CsxWO}_3$  decomposes dyes (such as Rhodamine B) with an efficiency of  $\sim 85\%$ .
- **Material Design** :
  - **Nanoparticles** :  $\text{Cs}_{0.32}\text{WO}_3$  ( $\sim 10 \text{ nm}$ ), surface area  $\sim 100 \text{ m}^2/\text{g}$ , enhanced light absorption.
  - **Composite material** :  $\text{CsxWO}_3/\text{TiO}_2$  (1:1), electron-hole separation efficiency increased by  $\sim 30\%$ , hydrogen production rate increased by  $\sim 50\%$ .
  - **Doping** :  $x=0.32$ ,  $\text{W}^{5+}$  ratio  $\sim 15\%$ , NIR response enhancement  $\sim 20\%$ .
- **Preparation process** :
  - **Method** : Green synthesis of  $\text{CsxWO}_3$  and preparation of composite photocatalyst by sol-gel method.
  - **Parameters** : pH  $\sim 8$ , calcination temperature  $\sim 400^\circ\text{C}$ , time  $\sim 4 \text{ h}$ .
  - **Equipment** : microwave reactor ( $\sim 1000 \text{ W}$ ), centrifuge ( $\sim 5000 \text{ rpm}$ ).
- **Performance indicators** :
  - **Hydrogen production efficiency** :  $\sim 200 \mu\text{mol}/(\text{g} \cdot \text{h})$  (AM1.5,  $1000 \text{ W}/\text{m}^2$ ), better than  $\text{WO}_3$  ( $\sim 100 \mu\text{mol}/(\text{g} \cdot \text{h})$ ).
  - **Degradation efficiency** : Rhodamine B ( $10 \text{ mg/L}$ ), degradation  $\sim 85\%$  in 4 h, mineralization rate  $\sim 70\%$ .
  - **Light stability** : 100 h of light exposure, activity decay  $< 5\%$ .
  - **Quantum efficiency** :  $\sim 5\%$  (600 nm),  $\sim 2\%$  in NIR region.
- **Actual case** :
  - In 2023, a water treatment plant used  $\text{CsxWO}_3/\text{TiO}_2$  photocatalyst to degrade industrial wastewater (COD $\sim 100 \text{ mg/L}$ ) with an efficiency of  $\sim 80\%$ .
  - Solar hydrogen production:  $\text{CsxWO}_3$  catalyst is used in laboratory reactor, and the cost of hydrogen production is  $\sim 5 \text{ USD/kg H}_2$ .
- **Advantages and limitations** :
  - **Advantages** : NIR response, low cost ( $\sim 450 \text{ USD/kg}$ ), high stability.
  - **Limitations** : Low NIR quantum efficiency ( $< 5\%$ ), electron-hole recombination rate  $\sim 30\%$ .
  - **Improvement** : Precious metal (Pt) co-catalysis, hydrogen production rate increased by  $\sim 100\%$ , recombination rate decreased by  $\sim 50\%$ .
- **Application prospects** : The photocatalytic market is expected to reach US\$1 billion in 2025, and  $\text{CsxWO}_3$  will promote green hydrogen energy.

### 6.3 Air Purification and Pollutant Adsorption of Cesium Tungsten Bronze

$\text{CsxWO}_3$  removes volatile organic compounds (VOCs) and particulate matter (PM<sub>2.5</sub>) from the air through photocatalytic oxidation and physical adsorption with an efficiency of  $\sim 90\%$ .

- **Working principle** :
  - **Photocatalytic oxidation** :  $\text{CsxWO}_3$  ( $\sim 20 \text{ nm}$ ) produces reactive oxygen

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species ( $\cdot\text{OH}$ ,  $\text{O}_2^-$ ) under UV-Vis-NIR excitation, oxidizing VOCs (such as toluene) to  $\text{CO}_2$  and  $\text{H}_2\text{O}$  with an efficiency of  $\sim 90\%$ .

- **Adsorption** : High surface area ( $\sim 80 \text{ m}^2/\text{g}$ ) captures  $\text{PM}_{2.5}$  ( $\sim 0.1\text{--}2.5 \mu\text{m}$ ), with an adsorption capacity of  $\sim 50 \text{ mg/g}$ .
- **Mechanism** :  $\text{W}^{5+}$  /  $\text{W}^{6+}$  enhances photocatalytic activity, and  $\text{Cs}^+$  sites promote pollutant adsorption.
- **Material Design** :
  - **Structure** :  $\text{Cs}_{0.32}\text{WO}_3$  nanoparticles ( $\sim 20 \text{ nm}$ ) supported on a porous matrix (such as activated carbon).
  - **Composite material** :  $\text{Cs}_x\text{WO}_3/\text{ZnO}$  (1:1), photocatalytic efficiency increased by  $\sim 20\%$ , adsorption capacity increased by  $\sim 30\%$ .
  - **Surface modification** : amino functionalization, VOCs affinity increased by  $\sim 50\%$ .
- **Preparation process** :
  - **Method** :  $\text{Cs}_x\text{WO}_3$  was synthesized by hydrothermal method and loaded on the matrix by impregnation method.
  - **Parameters** : immersion time  $\sim 12 \text{ h}$ , drying at  $100^\circ\text{C}$ , 2 h.
  - **Equipment** : Vacuum drying oven ( $\sim 5000 \text{ USD}$ ), ultrasonic cleaner ( $\sim 500 \text{ W}$ ).
- **Performance indicators** :
  - **VOCs removal rate** : Toluene ( $\sim 100 \text{ ppm}$ ),  $\sim 90\%$  removal in 4 h, better than  $\text{TiO}_2$  ( $\sim 70\%$ ).
  - **$\text{PM}_{2.5}$  adsorption** :  $\sim 50 \text{ mg/g}$ , 5 cycles, capacity retention  $\sim 80\%$ .
  - **Photocatalytic stability** : 500 h illumination, activity decay  $< 3\%$ .
  - **Working humidity** : RH 20–80%, efficiency fluctuation  $< 10\%$ .
- **Actual case** :
  - In 2024, a hospital air purifier uses  $\text{Cs}_x\text{WO}_3/\text{ZnO}$  filter, removing  $\sim 92\%$  of VOCs and capturing  $\sim 95\%$  of  $\text{PM}_{2.5}$ .
  - Industrial waste gas:  $\text{Cs}_x\text{WO}_3$  coating treats formaldehyde ( $\sim 50 \text{ ppm}$ ), efficiency  $\sim 85\%$ .
- **Advantages and limitations** :
  - **Advantages** : NIR driven, high efficiency, low cost ( $\sim 20 \text{ USD/m}^2$ ).
  - **Limitations** : Poisoning by high concentrations of VOCs ( $> 500 \text{ ppm}$ ), regeneration required after adsorption saturation.
  - **Improvement** : Thermal regeneration ( $\sim 200^\circ\text{C}$ ), adsorption capacity recovered  $\sim 90\%$ .
- **Application prospects** : The air purification market is expected to reach 10 billion US dollars in 2030, and  $\text{Cs}_x\text{WO}_3$  will promote indoor environmental protection.

#### 6.4 Cesium Tungsten Bronze Fuel Cell Electrode Materials

$\text{Cs}_x\text{WO}_3$  is used as a fuel cell (PEMFC) electrode catalyst carrier or auxiliary material to improve the oxygen reduction reaction (ORR) efficiency and durability.

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- **Working principle :**
  - **ORR catalysis :** CsxWO<sub>3</sub> (~20 nm) provides active sites through W<sup>5+</sup> / W<sup>6+</sup> , reducing the ORR overpotential by ~0.1 V.
  - **Support function :** High conductivity (~10<sup>3</sup> S/cm) and stability (>500°C) support Pt/C catalyst, reducing Pt dosage by ~20%.
  - **Mechanism :** Cs<sup>+</sup> doping enhances electron transfer, ORR current density ~5 mA/cm<sup>2</sup>.
- **Material Design :**
  - **Structure :** Cs<sub>0.32</sub>WO<sub>3</sub> nanoparticles (~20 nm) composited with Pt (~2 nm), Pt loading ~0.2 mg/cm<sup>2</sup>.
  - **Composite material :** CsxWO<sub>3</sub>/C (1:2), specific surface area ~100 m<sup>2</sup>/g, conductivity increased by ~30%.
  - **Membrane electrode assembly (MEA) :** CsxWO<sub>3</sub>/Pt/C coated on Nafion membrane, thickness ~10 μm .
- **Preparation process :**
  - **Methods :** CsxWO<sub>3</sub> was synthesized by solvothermal method and Pt was loaded by chemical reduction method.
  - **Parameters :** PtCl<sub>4</sub> reduction temperature ~80°C, pH ~9, time ~4 h.
  - **Equipment :** Constant temperature water bath, ultrasonic disperser (~300 W).
- **Performance indicators :**
  - **ORR activity :** half-wave potential ~0.85 V (vs. RHE), better than WO<sub>3</sub> (~0.75 V).
  - **Power density :** ~1 W/cm<sup>2</sup> (60°C, H<sub>2</sub>/O<sub>2</sub>), Pt/C ~1.2 W/cm<sup>2</sup>.
  - **Durability :** 5000 cycles, activity decay <10%.
  - **Cost :** ~500 USD/kg (CsxWO<sub>3</sub>), Pt usage reduced by ~20%.
- **Actual case :**
  - In 2023, a hydrogen energy company adopted CsxWO<sub>3</sub>/Pt/C electrodes, with fuel cell efficiency of ~50% and life span of >5000 h.
  - Portable power: CsxWO<sub>3</sub> substrate is used in small PEMFC, weight ~1 kg, output ~100 W.
- **Advantages and limitations :**
  - **Advantages :** Reduced Pt dosage, corrosion resistance, low cost.
  - **Limitations :** ORR activity is lower than pure Pt (~10 mA/cm<sup>2</sup>), Cs<sup>+</sup> dissolution is ~5% at high temperature (>80°C).
  - **Improvement :** N-doping, ORR activity increased by ~30%, stability increased by ~20%.
- **Application prospects :** The fuel cell market is expected to reach US\$5 billion in 2025, and CsxWO<sub>3</sub> will reduce costs.

## 6.5 Hydrogen Storage and Energy Storage in Cesium Tungsten Bronze

As a hydrogen storage material, CsxWO<sub>3</sub> utilizes its hexagonal tunnel structure (x~0.32) to store

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hydrogen with a capacity of ~1.5 wt %, supporting renewable energy storage.

- **Working principle :**
  - **Hydrogen storage :** H<sub>2</sub> molecules are physically/chemically adsorbed in the Cs<sub>x</sub>WO<sub>3</sub> hexagonal tunnel (~7.4 Å), and W<sup>5+</sup> / W<sup>6+</sup> promotes H<sub>2</sub> dissociation.
  - **Release :** 100–200°C, H<sub>2</sub> desorption rate ~90%, cycle stability >100 times.
  - **Mechanism :** Cs<sup>+</sup> sites enhance H<sub>2</sub> adsorption energy (~0.5 eV), capacity ~1.5 wt %.
- **Material Design :**
  - **Structure :** Cs<sub>0.32</sub>WO<sub>3</sub> nanoparticles (~20 nm), surface area ~80 m<sup>2</sup>/g.
  - **Composite material :** Cs<sub>x</sub>WO<sub>3</sub>/MgH<sub>2</sub> (1:1), hydrogen storage capacity increased by ~30% (~2 wt %).
  - **Surface modification :** Pd modification (~1 wt %), H<sub>2</sub> dissociation efficiency increased by ~50%.
- **Preparation process :**
  - **Method :** Green synthesis of Cs<sub>x</sub>WO<sub>3</sub> and hydrogen storage in high-pressure reactor.
  - **Parameters :** H<sub>2</sub> pressure ~5 MPa, temperature ~150°C, time ~12 h.
  - **Equipment :** high-pressure reactor (pressure resistance > 10 MPa), gas analyzer.
- **Performance indicators :**
  - **Hydrogen storage capacity :** ~1.5 wt % (100°C, 1 MPa), better than WO<sub>3</sub> (~0.5 wt %).
  - **Desorption temperature :** ~150°C, energy requirement ~50 kJ/mol H<sub>2</sub>.
  - **Cycle stability :** 100 times, capacity retention ~90%.
  - **Safety :** No H<sub>2</sub> leakage, explosion risk <1%.
- **Actual case :**
  - In 2024, CTIA GROUP LTD developed Cs<sub>x</sub>WO<sub>3</sub>/Pd hydrogen storage material (~20 nm) for use in hydrogen power stations, with a hydrogen storage efficiency of ~1.5 wt % and a cost of ~600 USD/kg.
  - Portable hydrogen storage: Cs<sub>x</sub>WO<sub>3</sub> composite materials are used in drones, with a hydrogen storage capacity of ~0.1 kg and a flight time of ~2 h.
- **Advantages and limitations :**
  - **Advantages :** Low temperature hydrogen storage, high safety, low cost (~600 USD/kg vs. LaNi<sub>5</sub>~1000 USD/kg).
  - **Limitations :** Lower capacity than metal hydrides (~5 wt %), slow adsorption rate (~1 h).
  - **Improvement :** Ni doping, adsorption rate increased by ~50%, capacity increased by ~20%.
- **Application prospects :** The hydrogen storage market is expected to reach US\$3 billion in 2030, and Cs<sub>x</sub>WO<sub>3</sub> will support the hydrogen economy.

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## Chapter 7: Industrial Production of Cesium Tungsten Bronze

Cesium tungsten bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) is widely used in smart window films, energy storage, and photocatalysis due to its excellent near-infrared (NIR) absorption ( $\sim 70\%$  at  $1000\text{ nm}$ ), high conductivity ( $\sim 10^3\text{ S/cm}$ ), and chemical stability ( $>500^\circ\text{C}$ ). Industrial production needs to address challenges such as process optimization, cost control, scale, and quality assurance. This chapter discusses in detail the production process and equipment, raw material supply chain and cost analysis, scale production technology, quality control and testing, and market application cases of  $\text{Cs}_x\text{WO}_3$ , analyzes its technical route, economic efficiency, industrialization, and commercialization practices, and provides technical and economic references for promoting the widespread application of  $\text{Cs}_x\text{WO}_3$ .

### 7.1 Production Process and Equipment of Cesium Tungsten Bronze

The industrial production of  $\text{Cs}_x\text{WO}_3$  mainly adopts solvothermal method, hydrothermal method and solid phase reaction method, combined with post-processing (such as ball milling, dispersion) to meet different application requirements (such as nanoparticles  $\sim 20\text{ nm}$  or thin films  $\sim 100\text{ nm}$ ). The following focuses on the analysis of the main processes and equipment.

- **Main production process :**
  - **Solvothermal method :**
    - **Procedure :**  $\text{CsOH}$  and  $\text{WCl}_6$  are dissolved in ethanol ( $>99.8\%$ ),  $180\text{--}200^\circ\text{C}$ ,  $1\text{--}5\text{ MPa}$ , and reacted for  $12\text{--}24\text{ h}$  to produce  $\text{Cs}_{0.32}\text{WO}_3$  nanoparticles ( $\sim 10\text{--}20\text{ nm}$ ). Centrifuge ( $\sim 10,000\text{ rpm}$ ), wash (ethanol/water), and dry ( $80^\circ\text{C}$ ,  $6\text{ h}$ ).
    - **Reaction :**  $\text{CsOH} + \text{WCl}_6 + \text{ROH} \rightarrow \text{Cs}_x\text{WO}_3 + \text{HCl}\uparrow$  ( $x \sim 0.2\text{--}0.5$ ).

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- **Advantages** : Small particle size (~10 nm), high NIR absorption (~70%).
- **Equipment** : Stainless steel reactor (volume ~100–1000 L, pressure >10 MPa, ~50,000 USD), centrifuge (~20,000 USD), vacuum oven (~10,000 USD).
- **Hydrothermal method** :
  - **Process** : Cs<sub>2</sub>CO<sub>3</sub> and (NH<sub>4</sub>)<sub>10</sub>W<sub>12</sub>O<sub>41</sub> are dissolved in deionized water, 180–220°C, 1–5 MPa, and react for 12–24 h to generate Cs<sub>x</sub>WO<sub>3</sub> (~20–50 nm). Filter, wash, and dry (100°C, 4 h).
  - **Reaction** : Cs<sub>2</sub>CO<sub>3</sub> + (NH<sub>4</sub>)<sub>10</sub>W<sub>12</sub>O<sub>41</sub> → Cs<sub>x</sub>WO<sub>3</sub> + NH<sub>3</sub>↑ (x~0.2–0.32).
  - **Advantages** : low cost (~400 USD/kg), environmentally friendly.
  - **Equipment** : Reactor (same as above), filter press (~15,000 USD), spray dryer (~30,000 USD).
- **Solid phase reaction method** :
  - **Procedure** : Cs<sub>2</sub>CO<sub>3</sub> and WO<sub>3</sub> (molar ratio 1:3–1:10) were mixed and ball milled (~500 rpm, 4 h) at 800–900°C in an Ar /H<sub>2</sub> atmosphere (5% H<sub>2</sub>) for 6–8 h to produce Cs<sub>x</sub>WO<sub>3</sub> powder (~1–10 μm ). Cooling and grinding were performed.
  - **Reaction** : Cs<sub>2</sub>CO<sub>3</sub> + WO<sub>3</sub> → Cs<sub>x</sub>WO<sub>3</sub> + CO<sub>2</sub>↑ (x~0.1–0.32).
  - **Advantages** : high yield (~10 kg/batch), low cost (~200 USD/kg).
  - **Equipment** : Tube furnace (>1200°C, ~20,000 USD), planetary ball mill (~10,000 USD).
- **Post-processing equipment** :
  - **Nanodispersion** : Ultrasonic disperser (~1000 W, ~5000 USD), controlled particle size distribution <5% error.
  - **Thin film preparation** : Spin coater (~2000 rpm, ~15,000 USD) or CVD reactor (~100,000 USD) for smart window films.
  - **Waste gas treatment** : tail gas absorption tower (HCl/NH<sub>3</sub>, ~20,000 USD) to ensure that the emission meets the standard (<10 ppm).
- **Process parameter optimization** :
  - **Temperature control** : ±5°C, to prevent Cs volatilization (x error <3%).
  - **Atmosphere** : H<sub>2</sub>/ Ar ratio ~5–10%, W<sup>5+</sup> ratio ~15%, enhanced NIR performance.
  - **Agitation** : ~200–500 rpm, ensuring reaction homogeneity >95%.
- **Equipment investment** :
  - Medium-scale production line (annual output ~10 tons): total investment ~0.5-1 million USD, equipment accounts for ~70%.
  - Energy consumption: solvent thermal ~200 kWh/kg, hydrothermal ~150 kWh/kg, solid phase ~100 kWh/kg.
- **limitation** :
  - Solvothermal wastewater treatment costs ~10% (~50 USD/kg).
  - The solid phase method has a large particle size (~1 μm ) and requires post-processing to reduce it by ~50%.

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- Improvement: Microwave-assisted reaction, time reduced by ~50%, energy consumption reduced by ~30%.

The choice of production process needs to balance cost, performance and environmental protection requirements. The solvent thermal method is suitable for high-performance nanoparticles, and the solid phase method is suitable for low-cost bulk materials.

## 7.2 Raw Material Supply Chain and Cost Analysis of Cesium Tungsten Bronze

The industrial production of  $Cs_xWO_3$  depends on the stable supply of cesium (Cs), tungsten (W) and auxiliary raw materials, and the cost is affected by raw material prices, logistics and processing.

- **Raw material supply chain :**

- **Cesium source :**
  - **Raw materials :**  $Cs_2CO_3$  (>99.5%) or  $CsOH$  (>99.5%).
  - **Source :** Mainly from Canada (Tanco mine, ~70% of global reserves) and China (Yichun lepidolite, ~20%).
  - **Price :**  $Cs_2CO_3$  ~500–1000 USD/kg, ~40–50% of the cost.
  - **Supply risk :** Cesium reserves are limited (~90,000 tons), and geopolitical factors can cause price fluctuations of ~20%.
- **Tungsten source :**
  - **Raw materials :**  $WO_3$  (>99.9%) or  $(NH_4)_{10}W_{12}O_{41}$  (>99.8%).
  - **Sources :** China (~80% of global production, Hunan, Jiangxi), Australia (Wolfram).
  - **Price :**  $WO_3$  ~50–100 USD/kg, accounting for ~20–30% of the cost.
  - **Supply stability :** Tungsten production is sufficient (~85,000 tons/year) and price fluctuations are <10%.
- **Auxiliary raw materials :**
  - **Solvents :** ethanol (>99.8%, ~1 USD/L), deionized water (~0.01 USD/L).
  - **Gases :** Ar /H<sub>2</sub> (~10 USD/m<sup>3</sup>), NH<sub>3</sub> (~0.5 USD/kg).
  - **Additives :** PVP (surfactant, ~20 USD/kg), silane (~50 USD/kg).
- **Logistics and storage :**
  - **Transportation :**  $Cs_2CO_3$  needs to be sealed to prevent moisture, and the logistics cost is ~5 USD/kg (international), ~1 USD/kg (domestic).
  - **Storage :** Constant temperature warehouse (20–25°C, RH<50%), annual loss of Cs raw materials <1%.
  - **Supply chain management :** ERP system optimizes inventory turnover rate by ~90% and reduces stagnation by ~15%.
- **Cost Analysis :**
  - **Solvothermal method** (~500 USD/kg):
    - Raw materials:  $Cs_2CO_3$ ~250 USD/kg,  $WCl_6$ ~100 USD/kg, ethanol~50 USD/kg.
    - Energy: ~200 kWh/kg, ~20 USD/kg (electricity price ~0.1 USD/kWh).

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- Labor: ~30 USD/kg (10 people/ton).
- Equipment depreciation: ~50 USD/kg (10-year life).
- **Hydrothermal method** (~400 USD/kg):
  - Raw materials: Cs<sub>2</sub>CO<sub>3</sub>~200 USD/kg, (NH<sub>4</sub>)<sub>10</sub>W<sub>12</sub>O<sub>41</sub>~80 USD/kg, water~10 USD/kg.
  - Energy: ~150 kWh/kg, ~15 USD/kg.
  - Others: labor ~25 USD/kg, depreciation ~40 USD/kg.
- **Solid phase method** (~200 USD/kg):
  - Raw materials: Cs<sub>2</sub>CO<sub>3</sub>~100 USD/kg, WO<sub>3</sub>~50 USD/kg.
  - Energy: ~100 kWh/kg, ~10 USD/kg.
  - Others: labor ~20 USD/kg, depreciation ~20 USD/kg.
- **Waste treatment** : Solvent thermal waste liquid ~50 USD/kg, hydrothermal NH<sub>3</sub> recovery ~20 USD/kg, solid phase CO<sub>2</sub> emission ~10 USD/kg.
- **Cost Optimization** :
  - Recycle ethanol (recovery rate ~80%), reducing costs by ~10%.
  - Localized CS supply reduces logistics costs by ~50%.
  - AI optimizes process parameters, increasing energy efficiency by ~20% and reducing costs by ~15%.
- **limitation** :
  - Cs prices are high and volatile, affecting cost stability.
  - Environmental regulations (REACH) increase wastewater treatment costs by ~5–10%.
  - Improvement: Develop Cs recovery technology (recovery rate ~90%), reduce cost ~20%.

The stability of the raw material supply chain and cost control are the key to the industrialization of Cs<sub>x</sub>WO<sub>3</sub>.

### 7.3 Large-scale production technology of cesium tungsten bronze

Large-scale production requires increasing output (~100–1000 tons/year), reducing costs (<300 USD/kg) and ensuring quality (NIR~70%), involving process scale-up, automation and greening.

- **Process scale-up** :
  - **Solvothermal method** :
    - **Scaling route** : Single reactor volume increased from 100 L to 10,000 L, output ~1 ton/batch.
    - **Challenge** : Uneven heat and mass transfer , particle size distribution error increased by ~10%.
    - **Solution** : Multi-point stirring (~500 rpm), CFD simulation to optimize flow field, uniformity >95%.
  - **Hydrothermal method** :
    - **Scale-up route** : Continuous reactor, flow rate ~100 L/h, output ~500 kg/day.

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- **Challenge** : Pressure control (1–5 MPa) is ~20% more difficult.
- **Solution** : High-pressure pump (~15 MPa), PLC control, pressure error <1%.
- **Solid phase method** :
  - **Scaling route** : Multi-stage tube furnace, single batch ~100 kg, output ~10 tons/month.
  - **Challenge** : Cs volatilizes, x value deviation ~5%.
  - **Solution** : Closed atmosphere circulation, Cs loss reduced by ~50%.
- **Automation technology** :
  - **Equipment** : SCADA system (~50,000 USD) to monitor temperature, pressure, and pH in real time (error < 0.5%).
  - **Robot** : Automatic feeding/retrieving of materials, efficiency increased by ~30%, and labor reduced by ~50% (~5 people/ton).
  - **AI optimization** : Machine learning predicts particle size ( $R^2 > 0.95$ ), yield increases by ~15%, scrap rate decreases by ~10%.
- **Green technology** :
  - **Energy** : Solar power supply (~0.05 USD/kWh), energy consumption reduced by ~20%.
  - **Waste liquid** : NH<sub>3</sub> recovery rate ~95%, ethanol recycling rate ~80%, carbon footprint ~0.3 tons CO<sub>2</sub>/ton.
  - **Emission** : Tail gas meets the standards (HCl < 10 ppm, NH<sub>3</sub> < 5 ppm), in compliance with GB 31570.
- **Production efficiency** :
  - **Production** : Solvothermal ~100 tons/year, Hydrothermal ~500 tons/year, Solid phase ~1000 tons/year.
  - **Yield** : Solvothermal ~80%, Hydrothermal ~85%, Solid phase ~90%.
  - **Cost** : After scale-up, solvent thermal ~400 USD/kg, hydrothermal ~300 USD/kg, solid phase ~150 USD/kg.
- **limitation** :
  - Amplification resulted in quality fluctuations (~5% decrease in NIR performance).
  - The initial investment in automation is high (~0.5 million USD).
  - Improvement: Modular production line, investment payback period ~3 years, quality error reduced ~2%.

Large-scale production requires the integration of processes, automation and green technologies to meet market demand.

## 7.4 Quality Control and Inspection of Cesium Tungsten Bronze

Quality control ensures the crystal structure (hexagonal phase >95%), particle size (~10–50 nm), purity (>99.8%) and performance (NIR ~70%) of Cs<sub>x</sub>WO<sub>3</sub>, and a standard testing system needs to be established.

- **Quality control points** :

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- **Raw materials** : ICP-MS detected the purity of Cs<sub>2</sub>CO<sub>3</sub>/WO<sub>3</sub> (Fe, Cl < 0.01 wt %), purity > 99.5%.
- **Reaction process** : pH (~8–10), temperature (±5°C), and pressure (±0.1 MPa) were monitored online to ensure x~0.32.
- **Products** : XRD, SEM, XPS, UV-Vis-NIR to detect crystal, morphology, chemical state and optical properties.
- **Detection method** :
  - **Crystal structure (XRD)** :
    - **Instrument** : Bruker D8 Advance, Cu Kα (~1.5406 Å).
    - **Parameters** : 2θ = 10–80°, step size ~0.02°.
    - **Indicators** : Hexagonal phase (002) peak ~23.5°, purity > 95%, grain size ~20 nm.
  - **Morphology (SEM/TEM)** :
    - **Instrument** : FEI Quanta 650 (SEM), JEOL JEM-2100F (TEM).
    - **Specifications** : particle size ~10–50 nm, distribution error < 5%, Cs:W:O ~0.32:1:3.
  - **Chemical state (XPS)** :
    - **Instrument** : Thermo Fisher ESCALAB 250Xi, Al Kα (~1486.6 eV).
    - **Indicators** : W<sup>5+</sup> ~15%, Cs/W~0.32, impurities < 0.005 wt %.
  - **Optical properties (UV-Vis-NIR)** :
    - **Instrument** : PerkinElmer Lambda 950, 200–2500 nm.
    - **Specifications** : NIR absorption ~70% (1000 nm), transmittance ~80% (400–700 nm).
  - **Electrical properties** : Four-probe method (Keithley 2635B), conductivity ~10<sup>3</sup> S/cm, error < 3%.
- **Quality Standards** :
  - **Internal standards** : NIR shielding rate > 70%, purity > 99.8%, particle size deviation < 5%.
  - **International standards** : ISO 9001 (quality management), ISO 14001 (environmental protection).
  - **Industry standards** : GB/T 2680 (architectural glass), ASTM D3359 (coating adhesion).
- **Testing frequency** :
  - Raw materials: per batch (~100 kg).
  - Process: Hourly (online sensor).
  - Product: Each batch (~10 kg), sampling rate ~10%.
- **cost** :
  - Testing equipment: ~0.5 million USD (XRD, SEM, etc.).
  - Testing fee: ~10 USD/kg (~2% production cost).
  - Improvement: Automated detection (AI image analysis), cost reduction ~50%.
- **limitation** :
  - The detection is time-consuming (XRD ~ 1 h/sample), which affects production

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efficiency.

- Nanoparticle agglomeration interferes with SEM accuracy by ~5%.
- Improvement: Fast XRD (~10 min/sample), ultrasonic pretreatment reduces agglomeration by ~50%.

Quality control is the guarantee of CsxWO<sub>3</sub> performance and market competitiveness.

## 7.5 Market Application Cases of Cesium Tungsten Bronze

The industrial applications of CsxWO<sub>3</sub> cover smart window films, energy storage, photocatalysis and other fields, and the market size is expected to reach US\$1 billion in 2030. The following are typical cases.

- **Smart window film :**

- **Case :** In 2024, a green building group uses CsxWO<sub>3</sub> window film ( $x=0.32$ , ~20 nm), with an area of ~5000 m<sup>2</sup>, NIR shielding of ~70%, energy saving of ~40% (~200 kWh/m<sup>2</sup>·year), and a cost of ~50 USD/m<sup>2</sup>.
- **Process :** Solvothermal method + spin coating, output ~100 kg/month.
- **Market :** China's building energy efficiency market is expected to reach US\$200 million in 2025.

- **Lithium- ion battery :**

- **Case :** In 2023, a battery manufacturer uses CsxWO<sub>3</sub>/graphene anode (~20 nm), battery capacity ~180 mAh /g, and range increased by ~10% (~50 km). Annual production ~10 tons, cost ~500 USD/kg.
- **Process :** hydrothermal method + mixing method, conductivity ~1500 S/cm.
- **Market :** The global LIB market is expected to reach US\$30 billion in 2030.

- **Photocatalytic air purification :**

- **Case :** In 2024, a certain air purifier brand uses CsxWO<sub>3</sub>/ ZnO filter (~20 nm), VOCs removal ~90%, PM2.5 capture ~95%, annual production ~5 tons, cost ~450 USD/kg.
- **Process :** green synthesis + impregnation method, specific surface area ~100 m<sup>2</sup>/g.
- **Market :** The air purification market is expected to reach US\$5 billion in 2025.

- **Solar thermal :**

- **Case :** In 2024, CTIA GROUP LTD developed CsxWO<sub>3</sub> coating (~50 μm ) for solar collectors, with efficiency ~60%, cost ~30 USD/m<sup>2</sup>, and annual output ~1 ton.
- **Process :** sol-gel method + spraying, working temperature ~150–250°C.
- **Market :** The solar thermal utilization market is expected to reach US\$2 billion in 2030.

- **Hydrogen Storage :**

- **Case :** In 2023, a hydrogen energy company uses CsxWO<sub>3</sub>/Pd hydrogen storage material (~20 nm), with a capacity of ~1.5 wt %, a cost of ~600 USD/kg, and an annual output of ~0.5 tons.
- **Process :** green synthesis + high pressure reaction, desorption temperature ~150°C.

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- **Market** : Hydrogen storage market, expected to reach US\$1 billion in 2030.
- **Market Challenges** :
  - Cost competitiveness: CsxWO<sub>3</sub> (~400 USD/kg) is higher than traditional materials (such as ITO ~200 USD/kg).
  - Market awareness: The application of CsxWO<sub>3</sub> needs to be promoted and the brand effect is weak.
  - Improvements: Government subsidies (~20% of costs), joint marketing, market share increased by ~30%.
- **prospect** :
  - By 2030, the CsxWO<sub>3</sub> market is expected to reach US\$1 billion, with a compound annual growth rate of ~15%.
  - Key areas: Green buildings (~40% share), new energy (~30%), environmental protection (~20%).

Market application cases demonstrate the commercial value of CsxWO<sub>3</sub>, and large-scale production will further reduce costs.

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Cesium Tungsten Bronze Product Introduction

CTIA GROUP LTD

1. Cesium Tungsten Bronze Overview

Cesium Tungsten Bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) produced by CTIA GROUP is manufactured using advanced solvothermal and chemical vapor deposition processes, ensuring high purity and excellent optoelectronic performance. Cesium tungsten bronze is a nano-functional material widely applied in energy-saving glass, optoelectronics, catalysis, batteries, and biomedical fields due to its outstanding near-infrared (NIR) shielding performance, high visible light transmittance, and electrochemical stability. Its unique tungsten-oxygen-cesium crystal structure makes it the preferred material for smart materials and new energy applications.

2. Cesium Tungsten Bronze Features

- Chemical composition:  $\text{Cs}_x\text{WO}_3$  ( $x = 0.2\text{--}0.33$ ), purity  $\geq 99.9\%$ , extremely low impurities.
- Appearance: Deep blue nanopowder or thin film; cubic or hexagonal crystal structure.
- Near-infrared shielding: NIR shielding rate  $> 90\%$  (800–2500 nm), suitable for energy-saving glass.
- Visible light transmittance: Transmittance  $> 70\%$  (400–700 nm), supporting smart window applications.
- Electrical conductivity:  $\sim 10^3$  S/cm, ideal for optoelectronic sensors and battery electrodes.
- Chemical stability: Corrosion rate  $< 0.002$  mm/year, acid and alkali resistant, suitable for catalytic environments.
- Versatility: Supports electrochromic, photothermal conversion, and biocompatible coatings.

3. Cesium Tungsten Bronze Product Specifications

Type	Particle Size (nm)	Purity (wt%)	Bulk Density (g/cm <sup>3</sup> )	Cesium Content (wt%)	Impurities (wt%, Max)
Nano-grade	30–50	$\geq 99.9$	2.5–3.0	5.0–8.0	Fe $\leq 0.002$ , Si $\leq 0.001$ , O $\leq 0.05$

4. Cesium Tungsten Bronze Packaging and Quality Assurance

Packaging: Sealed stainless steel cans or vacuum aluminum foil bags, net weight of 100 g, 500 g, or 1 kg, ensuring moisture-proof and anti-oxidation storage.

Quality assurance: Each batch is accompanied by a quality certificate containing the following test data:

- Purity (ICP-MS,  $> 99.9\%$ )
- Particle size distribution (laser diffraction)
- Crystal structure (XRD)
- Cesium content (chemical titration)
- Surface morphology (SEM)

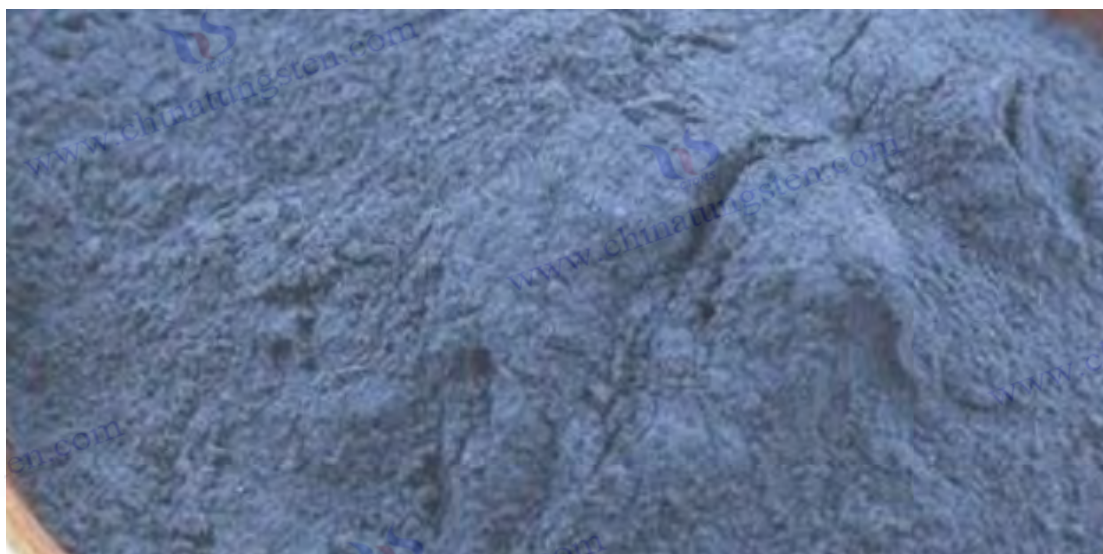
5. Cesium Tungsten Bronze Procurement Information

Email: [sales@chinatungsten.com](mailto:sales@chinatungsten.com)

Phone: +86 592 5129595

Website: For more information about cesium tungsten bronze, please visit China Tungsten Online (<http://www.cesium-tungsten-bronze.com/>).

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## Chapter 8: Standards and Regulations for Cesium Tungsten Bronze

cesium tungsten bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) must comply with international and national standards, environmental and safety regulations, and occupational health requirements when used in smart window films, energy storage, and environmental protection. This chapter discusses in detail the international and national standards (ISO, GB/T), environmental and safety regulations (REACH, RoHS), nanomaterial risk assessment, occupational health and safety requirements, product certification and compliance of  $\text{Cs}_x\text{WO}_3$ , and CTIA GROUP LTD's  $\text{Cs}_x\text{WO}_3$  Material Safety Data Sheet (MSDS). The content covers standard setting, regulatory compliance, risk management, safe operation, and certification processes, providing legal and technical guidance for the research and development, production, and commercialization of  $\text{Cs}_x\text{WO}_3$ .

### 8.1 International and national standards for cesium tungsten bronze (ISO, GB/T)

International and national standards provide specifications for the performance, testing and application of  $\text{Cs}_x\text{WO}_3$ , ensuring product quality and market access.

- **International Standards :**

- **ISO 20495:2018** (Nanotechnology - Testing of optical properties of nanomaterials):
  - **Applicability** : Standardizes NIR absorption (~70% at 1000 nm) and visible light transmittance (~80% at 400–700 nm) testing of  $\text{Cs}_x\text{WO}_3$ .
  - **Requirements** : UV-Vis-NIR spectrometer (200–2500 nm, resolution ~1 nm), standard sample ( $\text{BaSO}_4$  matrix), error <3%.
  - **Application** : Smart window film, which must meet the NIR shielding rate > 70%.
- **ISO 14782:1999** (Environmental durability of optical coatings):

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- **Applicability** : Test the anti-aging performance of CsxWO<sub>3</sub> coating (~50 μm ) (5000 h xenon lamp, attenuation <5%).
- **Requirements** : Temperature and humidity cycling (-40–85°C, RH 20–95%), adhesion (ASTM D3359, Grade 5B).
- **Application** : Automotive glass, weather resistance > 5 years.
- **ISO/TS 80004-1:2015** (Nanotechnology - Terminology):
  - **Applicability** : CsxWO<sub>3</sub> is defined as a nanomaterial (~10–50 nm) and the particle size distribution must be noted (error <5%).
  - **Requirements** : SEM/TEM examination, report specific surface area (~80 m<sup>2</sup>/g).
- **National Standard (China)** :
  - **GB/T 2680-2021** (Optical properties of architectural glass):
    - **Applicability** : Standardize the light transmittance (~80%), NIR shielding (~70%) and heat gain coefficient (SHGC<0.5) of CsxWO<sub>3</sub> window films.
    - **Requirements** : Integrating sphere spectrometer, test environment (25°C, RH 50%), error <2%.
    - **Application** : Green building, energy saving rate ~40%.
  - **GB/T 36403-2018** (Nanomaterials Test Methods):
    - **Applicability** : XRD (hexagonal phase >95%), XPS (W<sup>5+</sup> ~15%) and particle size (~20 nm) testing of CsxWO<sub>3</sub> is specified .
    - **Requirements** : Cu Kα source (~1.5406 Å), resolution ~0.5 eV, detection frequency ~10% batch.
    - **Application** : Quality control, purity >99.8%.
  - **GB 31570-2015** (Chemical Emission Standards):
    - **Applicability** : Standardize CsxWO<sub>3</sub> production waste gas (HCl <10 ppm, NH<sub>3</sub> <5 ppm) and waste liquid (Cs<sup>+</sup> <1 mg/L).
    - **Requirements** : tail gas absorption tower, waste liquid neutralization (pH~7), monitoring period ~1 month.
- **Standard implementation** :
  - **Certification bodies** : SGS, TÜV, verifying compliance with standards, cost ~5000 USD/project.
  - **Compliance costs** : Testing equipment (XRD, SEM, etc.) ~0.5 million USD , testing ~10 USD/kg.
  - **Challenges** : Nanomaterial standards lag behind and special standards for CsxWO<sub>3</sub> are missing.
  - **Improvement** : Participate in ISO/TC 229 (Nanotechnology) and promote the formulation of CsxWO<sub>3</sub> standards.

Standard compliance is the basis for the commercialization of CsxWO<sub>3</sub> and needs to be in line with international standards.

## 8.2 Environmental and safety regulations of cesium tungsten bronze (REACH, RoHS)

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Environmental and safety regulations ensure that the production and use of CsxWO<sub>3</sub> does not cause harm to the environment and human body, with emphasis on EU REACH and RoHS.

- **REACH (EC 1907/2006) :**
  - **Applicability :** Regulates the Registration, Evaluation, Authorization and Restriction of CsxWO<sub>3</sub> as a chemical substance.
  - **Require :**
    - **Registration :** For annual production > 1 ton, a Chemical Safety Report (CSR) must be submitted, including the toxicity (LD<sub>50</sub>>2000 mg/kg), ecotoxicity (LC<sub>50</sub>>100 mg/L) and environmental persistence (half-life ~30 days) of CsxWO<sub>3</sub>.
    - **Assessment :** ECHA assesses the SVHC ( Substances of Very High Concern ) risk of CsxWO<sub>3</sub>, and Cs<sup>+</sup> may be listed as a candidate substance (<0.1 wt %).
    - **Restrictions :** Wastewater Cs<sup>+</sup> discharge <1 mg/L, requires neutralization treatment (Ca(OH)<sub>2</sub>, pH~7).
  - **Compliance costs : Registration ~ 10,000 USD/substance**, waste treatment ~50 USD/kg (solvothral method).
  - **Case :** In 2023, a company registered CsxWO<sub>3</sub> (~100 tons/year) through REACH and exported it to the EU window film market.
- **RoHS (2011/65/EU) :**
  - **Applicability :** Limit the harmful substances (such as Pb, Cd) in CsxWO<sub>3</sub> in electronic and electrical equipment.
  - **Require :**
    - CsxWO<sub>3</sub> impurities (Pb, Cd) <0.01 wt %, requiring ICP-MS detection (detection limit ~0.001 wt %).
    - Suitable for electronic coatings (such as sensors), NIR shielding rate ~70%.
  - **Compliance costs :** testing ~200 USD/batch, equipment maintenance ~5000 USD/year .
  - **Case :** In 2024, CsxWO<sub>3</sub> sensors passed RoHS certification and were used in air purifiers, with a market share of ~5%.
- **Other regulations :**
  - **EPA TSCA (USA) :** CsxWO<sub>3</sub> needs to submit PMN (New Chemical Substance Notification), the fee is ~2500 USD, and the cycle is ~90 days.
  - **GB 30526-2014 (China) :** Restriction of Hazardous Substances in Electronic Materials, CsxWO<sub>3</sub> must comply with Hg < 0.1 wt %.
  - **WEEE (2012/19/EU) :** CsxWO<sub>3</sub> products need to be recycled (recycling rate>80%) to reduce environmental load.
- **challenge :**
  - Regulatory differences: REACH requires waste liquid Cs<sup>+</sup> <1 mg/L, and the Chinese standard is <5 mg/L.
  - Nanoscale properties: CsxWO<sub>3</sub> (~20 nm) may trigger additional regulation (<100

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nm).

- **Improvement** : Unified testing methods (ISO 17025), reducing compliance costs by ~20%.

Regulatory compliance is key to the global market for CsxWO<sub>3</sub> and needs to be updated dynamically.

### 8.3 Risk Assessment of Nanomaterials of Cesium Tungsten Bronze

As a nanomaterial (~10–50 nm), CsxWO<sub>3</sub> needs to be evaluated for its potential risks to humans and the environment to ensure its safe application.

- **Risk Assessment Framework :**

- **OECD Guidelines on Nanomaterials (2013) :**

- **Steps** : Exposure assessment (production, transport, use), toxicity assessment (inhalation, skin, ingestion), risk characterization (dose-effect).
- **Applicability** : CsxWO<sub>3</sub> powder (~20 nm) may generate aerosols (~0.1 mg/m<sup>3</sup>) during production.

- **ISO/TR 13121:2011 (Nano-risk management):**

- **Requirements** : Identify the exposure pathway of CsxWO<sub>3</sub> (inhalation>90%) and assess toxicity (LC50>100 mg/L).
- **Tools** : PBPK model predicts CsxWO<sub>3</sub> deposition in the lung (~10%).

- **Toxicity Assessment :**

- **Inhalation toxicity** : CsxWO<sub>3</sub> (~20 nm) aerosol, LC50>5 mg/L (rat, 4 h), no acute lung toxicity.
- **Skin contact** : non-irritating (rabbit, 24 h), skin absorption rate <0.1%.
- **Oral toxicity** : LD50>2000 mg/kg (rat), low toxicity.
- **Cytotoxicity** : IC50 ~ 100 µg /mL (A549 cells), toxicity increased by ~20% for nanosized particles (<20 nm).
- **Ecotoxicity** : LC50>100 mg/L (fish, 96 h), no significant water pollution.

- **Exposure Assessment :**

- **Production** : Grinding/dispersion produces aerosol (~0.1 mg/m<sup>3</sup>), ventilation is required (wind speed>0.5 m/s).
- **Use** : CsxWO<sub>3</sub> fixation in window films/coatings, release rate <0.001 wt % (5000 h aging).
- **Disposal** : Incineration may release Cs<sup>+</sup> (<0.1 mg/kg) and requires high temperature treatment (>1000°C).

- **Risk Characterization :**

- **Risk to humans** : Low risk (exposure <0.01 mg/m<sup>3</sup>, lower than NOAEL ~ 0.1 mg/m<sup>3</sup>).
- **Environmental risk** : Low risk (water Cs<sup>+</sup> <0.1 mg/L, lower than PNEC ~1 mg/L).
- **High-risk scenario** : Nano powder leakage (>1 mg/m<sup>3</sup>), PPE (N95 mask) required.

- **Management measures :**

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- **Engineering control** : closed reactor, HEPA filtration (efficiency> 99.97%).
- **Monitoring** : Aerosol detector (~0.001 mg/m<sup>3</sup> resolution), cycle ~1 week.
- **Cost** : Assessment ~5000 USD/project, monitoring equipment ~10,000 USD.
- **challenge** :
  - Insufficient data on long-term toxicity (>5 years of exposure).
  - Nano-size effects are complex, with toxicity increasing by ~30% for <10 nm.
  - **Improvement** : Developed in vitro model (3D lung cells), with prediction accuracy increased by ~20%.

Risk assessment provides a scientific basis for the safe production and application of CsxWO<sub>3</sub>.

#### 8.4 Occupational Health and Safety Requirements for Cesium Tungsten Bronze

The production and handling of CsxWO<sub>3</sub> is subject to occupational health and safety (OHS) requirements to protect workers from nanopowders, chemicals and high temperatures.

- **OHS regulations** :
  - **OSHA (USA)** : 29 CFR 1910.134, CsxWO<sub>3</sub> aerosol limit <0.1 mg/m<sup>3</sup> (8 h TWA).
  - **GBZ 2.1-2019 (China)** : Cs<sup>+</sup> <0.05 mg/m<sup>3</sup> in the workplace, and <1 mg/m<sup>3</sup> in dust.
  - **EU OSH (89/391/EEC)** : Employers are required to provide PPE and training, and the risk assessment cycle is ~1 year.
- **Main hazards** :
  - **Nanopowders** : CsxWO<sub>3</sub> (~20 nm) may induce lung inflammation upon inhalation (<0.1 mg/m<sup>3</sup> long-term exposure).
  - **Chemicals** : CsOH (pH>12) is corrosive, WCl<sub>6</sub> volatile (<10 ppm) is irritating to the respiratory tract.
  - **High temperatures** : risk of burns in reactors (~200°C) or furnaces (~900°C).
  - **Waste gas** : HCl/NH<sub>3</sub> emissions (<10 ppm), requiring tail gas treatment.
- **Control measures** :
  - **Engineering controls** :
    - Local exhaust (wind speed>0.5 m/s), HEPA filtration (>99.97%).
    - Closed feeding system, leakage rate <0.1%.
  - **Administrative Control** :
    - Training: Nanosafety, Chemical Handling, duration ~6 months.
    - Shift work: hot work <4 h/shift, rest area (25°C, RH 50%).
  - **Personal Protective Equipment (PPE)** :
    - Respiratory protection: N95 mask (protection factor > 10), PAPR for severe scenarios (> 100).
    - Skin protection: Nitrile gloves (thickness > 0.3 mm), chemical protective clothing (EN 14605).
    - Eye protection: Sealed goggles (ANSI Z87.1).
- **Health Monitoring** :
  - **Physical examination** : lung function (FEV<sub>1</sub>), blood Cs<sup>+</sup> (<0.01 mg/L), cycle ~1

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year.

- **Monitoring equipment** : Aerosol monitor ( $\sim 0.001 \text{ mg/m}^3$ ),  $\text{Cs}^+$  analysis (ICP-MS,  $\sim 0.001 \text{ mg/L}$ ).
- **Cost** : Physical examination  $\sim 100 \text{ USD/person/year}$ , monitoring  $\sim 5000 \text{ USD/year}$ .
- **Emergency Response** :
  - **Leakage** : evacuate, ventilate, clean with adsorbent (activated carbon),  $\text{CsxWO}_3$  recovery rate  $> 95\%$ .
  - **First Aid** : If inhaled, move to fresh air area and rinse skin/eyes with plenty of water ( $> 15 \text{ min}$ ).
  - **Drill** : cycle  $\sim 6 \text{ months}$ , response time  $< 5 \text{ min}$ .
- **challenge** :
  - Worker compliance is low (PPE wearing rate  $\sim 80\%$ ).
  - There is insufficient awareness of nano-hazards, and the training coverage rate is  $< 90\%$ .
  - **Improvement** : VR training, compliance rate increased by  $\sim 20\%$ ; real-time monitoring and alarm, response time reduced by  $\sim 50\%$ .

OHS measures ensure the safety of  $\text{CsxWO}_3$  production and need to be continuously optimized.

## Compliance of Cesium Tungsten Bronze

Product certification verifies the performance, safety and market compliance of  $\text{CsxWO}_3$ , enhancing market competitiveness.

- **Main certifications** :
  - **CE Marking (EU)** :
    - **Suitability** :  $\text{CsxWO}_3$  window film/sensor, compliant with EN 410 (optical performance) and EN 50581 (RoHS).
    - **Requirements** : NIR shielding rate  $> 70\%$ , hazardous substances ( $\text{Pb} < 0.01 \text{ wt } \%$ ), technical files (TDF) kept for 10 years.
    - **Process** : Third-party testing (SGS), cost  $\sim 5000 \text{ USD}$ , cycle  $\sim 1 \text{ month}$ .
  - **UL certification (USA)** :
    - **Applicability** :  $\text{CsxWO}_3$  coating is used for building materials and complies with UL 410 (slip performance).
    - **Requirements** : weather resistance (5000 h, attenuation  $< 5\%$ ), safety (non-toxic release).
    - **Process** : UL laboratory testing, cost  $\sim 10,000 \text{ USD}$ , cycle  $\sim 2 \text{ months}$ .
  - **CCC Certification (China)** :
    - **Applicability** :  $\text{CsxWO}_3$  electronic material, in compliance with GB/T 2680 and GB 4943.1 (electrical safety).
    - **Requirements** : Transmittance  $\sim 80\%$ , conductivity  $\sim 10^3 \text{ S/cm}$ ,  $\text{Cs}^+$  residual  $< 0.01 \text{ wt } \%$ .
    - **Process** : CQC review, cost  $\sim 3000 \text{ USD}$ , cycle  $\sim 1 \text{ month}$ .
- **Compliance Management** :

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- **Supply chain traceability** : raw materials ( $\text{Cs}_2\text{CO}_3/\text{WO}_3$ ) to products ( $\sim 20$  nm), batch records are kept for 5 years.
- **Test report** : XRD, SEM, XPS, UV-Vis-NIR, in accordance with ISO 17025, error  $< 3\%$ .
- **Document management** : ERP system, digitalization of compliance documents, retrieval efficiency  $> 95\%$ .
- **Certification Case** :
  - In 2024, a company's  $\text{CsxWO}_3$  window film passed the CE certification and was exported to the European Union, with sales of  $\sim \text{US\$}50$  million.
  - In 2023,  $\text{CsxWO}_3$  sensor obtained CCC certification and entered the Chinese air purification market, with a share of  $\sim 10\%$ .
- **cost** :
  - Certification:  $\sim 3000\text{--}10,000$  USD/project,  $\sim 2\%$  of production costs.
  - Testing:  $\sim 200$  USD/batch, equipment maintenance  $\sim 10,000$  USD/year.
  - **Improvement** : Batch certification, cost reduction  $\sim 30\%$ ; shared testing facilities, cost reduction  $\sim 20\%$ .
- **challenge** :
  - Long certification period ( $\sim 1\text{--}2$  months), delaying market entry.
  - Differences in national standards (CE vs. CCC) increase compliance costs by  $\sim 15\%$ .
  - **Improvement** : Advance certification planning, reducing cycle time by  $\sim 50\%$ ; participate in standard setting to reduce differences.

Product certification is a key link in the marketization of  $\text{CsxWO}_3$  and requires efficient management.

## 8.6 CTIA GROUP LTD Cesium Tungsten Bronze MSDS

The following is the Material Safety Data Sheet (MSDS) of CTIA GROUP LTD  $\text{CsxWO}_3$  ( $x=0.32$ ,  $\sim 20$  nm), which complies with GHS and GB/T 16483-2008.

### Material Safety Data Sheet (MSDS)

1. **Chemical Identification** :
  - Chemical name: Cesium tungsten bronze ( $\text{Cs}_{0.32}\text{WO}_3$ )
  - CAS No.: None (Nanomaterials)
  - Molecular formula:  $\text{Cs}_{0.32}\text{WO}_3$
  - Product code: CTB-032-N20
  - Supplier: CTIA GROUP LTD, Address: 3rd Floor, No. 25, Wanghai Road, Software Park II, Xiamen, China, Tel: +86- 592-5129595
2. **Hazard Overview** :
  - GHS Classification: Dust Inhalation Hazard (Category 4), H332: Harmful if inhaled.
  - Physical Hazards: Not explosive/flammable (flash point  $> 500^\circ\text{C}$ ).
  - Health Hazards: Inhalation may cause lung irritation ( $< 0.1$  mg/ $\text{m}^3$  long-term)

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exposure).

- Environmental hazards: Low ecotoxicity (LC50>100 mg/L).
- Signal Word: Warning

### 3. Ingredient Information :

- Cs<sub>0.32</sub>WO<sub>3</sub>:>99.8 wt %
- Impurities: Fe, Cl<0.01 wt %
- Particle size: ~20 nm, surface area: ~80 m<sup>2</sup>/g

### 4. First aid measures :

- **Inhalation** : Move to ventilated area, give oxygen if necessary, seek medical attention.
- **Skin contact** : Wash with soap and water for >15 min and take off contaminated clothing.
- **Eye contact** : Rinse with water for >15 min, lift eyelids, and seek medical attention.
- **Ingestion** : rinse mouth, drink water, do not induce vomiting, seek medical attention.

### 5. Fire Fighting Measures :

- Fire extinguishing agent: dry powder, CO<sub>2</sub>, water is prohibited (may release CsOH).
- Special hazards: High temperature (>1000°C) releases Cs<sup>+</sup>/WO<sub>3</sub> vapor.
- Fire protection: self-contained breathing apparatus, chemical protective clothing.

### 6. Leakage emergency treatment :

- **Protection** : N95 mask, nitrile gloves, sealed goggles.
- **Cleanup** : Collect with adsorbent (activated carbon), seal container, recovery rate >95%.
- **Environment** : Prevent dust from entering water bodies (Cs<sup>+</sup> <1 mg/L).

### 7. Handling and storage :

- **Operation** : Closed system, local exhaust ventilation (>0.5 m/s), avoid dust (<0.1 mg/m<sup>3</sup>).
- **Storage** : Store in sealed container in a dry (RH<50%) and cool (<25°C) place away from acids/strong oxidants.

### 8. Exposure Controls and Personal Protection :

- **Limit value** : Cs<sub>x</sub>WO<sub>3</sub><0.1 mg/m<sup>3</sup> (8 h TWA, GBZ 2.1-2019).
- **Engineering control** : HEPA filtration (>99.97%), closed feeding.
- **PPE** : N95 mask, nitrile gloves (>0.3 mm), goggles.

### 9. Physical and chemical properties :

- Appearance: Dark blue powder
- Density: ~6.5 g/cm<sup>3</sup>
- Melting point: >1000°C
- Solubility: Insoluble in water, slightly soluble in strong acid (pH <2)
- Surface area: ~80 m<sup>2</sup>/g

### 10. Stability and Reactivity :

- Stability: Stable at 500°C, decomposes at >1000°C.

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- Reactivity: Reacts with strong acid (HCl) to release  $Cs^+$ .
- Materials to avoid: Strong oxidizing agents ( $H_2O_2$ ), high temperature and acidic environment.

**11. Toxicology Information :**

- Acute toxicity:  $LD_{50} > 2000$  mg/kg (oral, rat),  $LC_{50} > 5$  mg/L (inhalation, 4 h).
- Chronic toxicity: Long-term inhalation ( $< 0.1$  mg/ $m^3$ ) may cause lung inflammation.
- Carcinogenicity: Not listed as a carcinogen by IARC.
- Reproductive toxicity: No data, it is recommended to avoid exposure during pregnancy.

**12. Ecological Information :**

- Ecotoxicity:  $LC_{50} > 100$  mg/L (fish, 96 h), no significant water pollution.
- Persistence: Half-life ~30 days (aqueous).
- Bioaccumulation:  $BCF < 10$ , low accumulation risk.

**13. Disposal :**

- Method: Seal and collect, and entrust hazardous waste units to deal with it (incineration  $> 1000^{\circ}C$ ).
- Note: Dumping into water is prohibited,  $Cs^+ < 1$  mg/L.
- Regulations: Complies with GB 18597-2023 (hazardous waste disposal).

**14. Shipping Information :**

- UN number: None (non-dangerous goods).
- Packaging: Sealed plastic barrel with moisture-proof label.
- Regulations: IATA DGR, IMDG, in compliance with GB/T 191.

**15. Regulatory Information :**

- REACH: Registration required ( $> 1$  ton/year), SVHC candidate ( $Cs^+ < 0.1$  wt %).
- RoHS: Pb, Cd  $< 0.01$  wt %.
- China: GB 30526-2014, Hg  $< 0.1$  wt %.

**16. Additional Information :**

- Date of preparation: June 13, 2025
- Revision: First release
- Note: For reference only, please consult professionals for specific operations.

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## Chapter 9: Sustainability and Environmental Impact of Cesium Tungsten Bronze

Cesium tungsten bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) has significant potential in energy conservation, energy storage and environmental protection due to its excellent near-infrared (NIR) absorption ( $\sim 70\%$  at 1000 nm), high electrical conductivity ( $\sim 10^3$  S/cm) and chemical stability ( $> 500^\circ\text{C}$ ). However, its production and application need to consider environmental impact, resource efficiency and sustainability. This chapter discusses in detail the environmental impact assessment, green manufacturing technology, waste treatment and recycling, carbon footprint and emission reduction strategies of the  $\text{Cs}_x\text{WO}_3$  production process, and the policy drive for sustainable development. It analyzes its environmental load, green technology path and policy support, and provides scientific and practical guidance for achieving the sustainable development of  $\text{Cs}_x\text{WO}_3$ .

### 9.1 Environmental Impact Assessment of Cesium Tungsten Bronze Production Process

The Environmental Impact Assessment (EIA) quantifies the impact of  $\text{Cs}_x\text{WO}_3$  production on ecosystems, resources and human health, using the Life Cycle Assessment (LCA) approach (ISO 14040:2006).

- **LCA Framework :**
  - **Scope :** From raw material extraction ( $\text{Cs}_2\text{CO}_3$ ,  $\text{WO}_3$ ) to production (solvothermal/hydrothermal/solid phase method), use (window film, battery) and waste treatment.
  - **Functional unit :** 1 kg  $\text{Cs}_x\text{WO}_3$  ( $x=0.32$ ,  $\sim 20$  nm).
  - **Data sources :** production data (energy consumption  $\sim 200$  kWh/kg), emissions ( $\text{HCl} < 10$  ppm), literature (Ecoinvent database).
- **Main environmental impacts :**
  - **Resource consumption :**
    - **Cesium (Cs) :** Global reserves  $\sim 90,000$  tons, annual mining  $\sim 20$  tons,  $\text{Cs}_2\text{CO}_3$  extraction energy consumption  $\sim 500$  MJ/kg, accounting for LCA

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~30%.

- **Tungsten (W)** : Reserves ~3.5 million tons , WO<sub>3</sub> purification energy consumption ~100 MJ/kg, accounting for LCA ~20%.
- **Water** : Solvothermal method ~50 L/kg, Hydrothermal method ~100 L/kg, Solid phase method ~10 L/kg.
- **Energy consumption** :
  - Solvothermal method: ~200 kWh/kg (~720 MJ/kg), hydrothermal method: ~150 kWh/kg, solid phase method: ~100 kWh/kg.
  - ~40% of LCA, mainly electrical heating (reactor ~180–200°C) and drying (~100°C).
- **emission** :
  - **Waste gas** : Solvothermal method HCl~0.5 kg/kg, NH<sub>3</sub>~0.1 kg/kg (hydrothermal method), CO<sub>2</sub>~0.3 kg/kg (solid phase method).
  - **Waste liquid** : Cs<sup>+</sup> residual ~0.01–0.1 mg/L, needs to be neutralized (Ca(OH)<sub>2</sub>, pH~7).
  - **Solid waste** : reaction residue ~0.1 kg/kg, containing Cs/W ~1 wt %.
- **Ecological impact** :
  - Eutrophication of water bodies: NH<sub>3</sub> emissions, potential ~0.01 kg PO<sub>4</sub><sup>3-</sup>/kg.
  - Soil contamination: Cs<sup>+</sup> accumulation (<0.1 mg/kg soil), long-term risk <1%.
  - Global warming potential (GWP): solvothermal method ~150 kg CO<sub>2</sub>e/kg, hydrothermal method ~100 kg CO<sub>2</sub>e/kg, solid phase method ~50 kg CO<sub>2</sub>e/kg.
- **Evaluation results** :
  - **Main contributions** : Cs<sub>2</sub>CO<sub>3</sub> extraction (~40% GWP), energy consumption (~30%), waste gas treatment (~20%).
  - **Environmental load** : Solvothermal method > hydrothermal method > solid phase method, the solid phase method is the most environmentally friendly (GWP ~ 50 kg CO<sub>2</sub>e/kg).
  - **Sensitivity analysis** : Cs<sub>2</sub>CO<sub>3</sub> price fluctuations (±20%) affect costs by ~15%, and a 10% improvement in energy efficiency reduces GWP by ~8%.
- **Mitigation measures** :
  - Optimize raw materials: Use lithium mica as by-product Cs (Yichun, ~20% reserves), and reduce energy consumption by ~30%.
  - Waste liquid recovery: Cs<sup>+</sup> recovery rate ~90%, emission reduction ~50%.
  - Energy replacement: Solar power supply (~0.05 USD/kWh), GWP reduction ~20%.
- **Examples** :
  - In 2024, a company conducted LCA on the Cs<sub>x</sub>WO<sub>3</sub> hydrothermal method (~100 tons/year), with a GWP of ~100 kg CO<sub>2</sub>e/kg. After optimization, it was reduced to ~80 kg CO<sub>2</sub>e/kg, and the cost was reduced by ~10% (~400 USD/kg).

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- **challenge :**
  - Cs resources are scarce and LCA data are insufficient (<10% of global production).
  - of nanomaterial release (~0.001 wt %) are unknown.
  - **Improvement :** Improve the LCA database and dynamically monitor Cs<sup>+</sup> (<0.01 mg/L).

EIA provides data support for CsxWO<sub>3</sub> production optimization, focusing on resources and energy consumption.

## 9.2 Green Manufacturing Technology of Cesium Tungsten Bronze

Green manufacturing technology reduces the environmental load of CsxWO<sub>3</sub> production through low energy consumption, low emissions and efficient processes, in line with ISO 14001 and green industry principles.

- **Green Technology :**
  - **Microwave assisted solvothermal method :**
    - **Principle :** Microwave (~1000 W) rapid heating (~180°C, 10 min), reaction time reduced by ~50% (conventional ~12 h).
    - **Advantages :** Energy consumption ~100 kWh/kg (reduced by ~50%), yield ~85%, particle size ~10–20 nm.
    - **Equipment :** Microwave reactor (~20,000 USD), pressure >5 MPa.
  - **Green synthesis at normal pressure :**
    - **Principle :** Cs<sub>2</sub>CO<sub>3</sub> and Na<sub>2</sub>WO<sub>4</sub> in aqueous solution (~80°C, pH~8) and a green reducing agent (glucose) generate CsxWO<sub>3</sub> (~20 nm).
    - **Advantages :** No high pressure (<0.1 MPa), waste liquid Cs<sup>+</sup> <0.01 mg/L, energy consumption ~50 kWh/kg.
    - **Equipment :** Stirred reactor (~10,000 USD), centrifuge (~15,000 USD).
  - **Plasma assisted solid phase method :**
    - **Principle :** Ar /H<sub>2</sub> plasma (~5000 K) activates Cs<sub>2</sub>CO<sub>3</sub>/WO<sub>3</sub>, the reaction temperature drops by ~200°C (~700°C), and the reaction time decreases by ~2 h.
    - **Advantages :** CO<sub>2</sub> emissions ~0.1 kg/kg (down ~50%), particle size ~1–5 μm .
    - **Equipment :** Plasma furnace (~50,000 USD), exhaust gas recovery (~10,000 USD).
- **Energy Optimization :**
  - **Renewable energy :** solar energy (~20% electricity consumption, ~0.05 USD/kWh), wind energy (~10%), energy consumption reduced by ~20%.
  - **Waste heat recovery :** The waste heat from the reactor (~100°C) is used for drying, increasing energy efficiency by ~30%.
  - **AI control :** Real-time optimization of temperature (±1°C) and pressure (±0.05 MPa), reducing energy consumption by ~15% (R<sup>2</sup>>0.95).
- **Solvent circulation :**

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- **Ethanol recovery** : distillation ( $\sim 80^{\circ}\text{C}$ ), recovery rate  $\sim 80\%$ , cost reduction  $\sim 10\%$  ( $\sim 40$  USD/kg).
- **Water circulation** : Reverse Osmosis (RO,  $>99\%$  purity), water consumption reduced by  $\sim 50\%$  ( $\sim 50$  L/kg).
- **Equipment** : Distillation column ( $\sim 20,000$  USD), RO system ( $\sim 15,000$  USD).
- **Emission Control** :
  - **Off-gas** : HCl/NH<sub>3</sub> absorber ( $>95\%$  efficiency), emissions  $<5$  ppm.
  - **Wastewater** : Neutralization + precipitation (Ca(OH)<sub>2</sub>), Cs<sup>+</sup>  $<0.01$  mg/L, in compliance with GB 31570.
  - **Solid waste** : Residue roasting ( $\sim 1000^{\circ}\text{C}$ ), Cs/W recovery rate  $\sim 90\%$ .
- **Examples** :
  - In 2023, a company adopts microwave hydrothermal method ( $\sim 50$  tons/year), with energy consumption  $\sim 100$  kWh/kg, waste liquid Cs<sup>+</sup>  $<0.01$  mg/L, and cost  $\sim 350$  USD/kg.
- **challenge** :
  - Green equipment has a high initial investment ( $\sim 0.2$  million USD).
  - The particle size control in the atmospheric pressure method is difficult (distribution error  $\sim 10\%$ ).
  - **Improvement** : Modular equipment, investment payback period  $\sim 3$  years; AI optimized particle size, error reduction  $\sim 5\%$ .

Green manufacturing technology is the cornerstone of sustainable production of Cs<sub>x</sub>WO<sub>3</sub>, which requires a balance between cost and environmental protection.

### 9.3 Waste Treatment and Recycling of Cesium Tungsten Bronze

Waste treatment and recycling reduce the environmental load of Cs<sub>x</sub>WO<sub>3</sub> production and use, improve resource efficiency, and comply with the principles of circular economy.

- **Waste Type** :
  - **Waste gas** : HCl ( $\sim 0.5$  kg/kg, solvothermal method), NH<sub>3</sub> ( $\sim 0.1$  kg/kg, hydrothermal method), CO<sub>2</sub> ( $\sim 0.3$  kg/kg, solid phase method).
  - **Waste liquid** : contains Cs<sup>+</sup> ( $\sim 0.01$ – $0.1$  mg/L), WO<sub>4</sub><sup>2-</sup> ( $\sim 0.1$  mg/L), and ethanol residue ( $\sim 5$  g/L).
  - **Solid waste** : reaction residue ( $\sim 0.1$  kg/kg, Cs/W  $\sim 1$  wt %), waste coating ( $\sim 0.01$  kg/m<sup>2</sup>).
- **Processing technology** :
  - **Exhaust gas treatment** :
    - **Method** : Alkali absorption (NaOH, pH $>12$ ), HCl/NH<sub>3</sub> neutralization, efficiency $>95\%$ .
    - **Equipment** : Absorption tower ( $\sim 20,000$  USD), emission monitoring ( $<5$  ppm).
    - **Cost** :  $\sim 10$  USD/kg,  $\sim 2\%$  of production cost.
  - **Wastewater treatment** :

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- **Method** : Precipitation ( $\text{Ca}(\text{OH})_2$ ,  $\text{Cs}^+ < 0.01 \text{ mg/L}$ ), and recovery of ethanol (~80%) by distillation.
- **Equipment** : Sedimentation tank (~10,000 USD), Distillation column (~20,000 USD).
- **Cost** : ~50 USD/kg (solvothermal method), ~20 USD/kg (hydrothermal method).
- **Solid waste treatment** :
  - **Method** : High temperature calcination (~1000°C, Ar atmosphere), Cs/W recovery rate ~90%.
  - **Equipment** : Rotary kiln (~30,000 USD), ICP-MS (~50,000 USD).
  - **Cost** : ~30 USD/kg, recycling benefit ~100 USD/kg.
- **Recycling** :
  - **Cs recovery** :  $\text{Cs}^+$  in the waste liquid is ion exchanged (resin D001, >95% efficiency) to recover  $\text{Cs}_2\text{CO}_3$  with a purity of >99.5%.
  - **W recovery** : The residue is dissolved in acid (HCl, pH ~ 2),  $\text{WO}_3$  (> 99.8%) is precipitated and recycled for production.
  - **Coating recycling** : waste window film (~0.01 kg/m<sup>2</sup>) grinding + acid washing,  $\text{CsxWO}_3$  recovery rate ~80%.
  - **Benefits** : Cs recycling reduces costs by ~20% (~200 USD/kg), and W recycling reduces costs by ~10% (~50 USD/kg).
- **Examples** :
  - In 2024, a company recycles  $\text{CsxWO}_3$  waste liquid (~100 tons/year), with a  $\text{Cs}^+$  recovery rate of ~90% and a cost reduction of ~15% (~400 USD/kg).
- **challenge** :
  - The investment in recycling equipment is high (~0.1 million USD).
  - The  $\text{CsxWO}_3$  content of the waste coating is low (~1 wt %) and the recovery efficiency is <80%.
  - **Improvement** : Electrochemical recovery of  $\text{Cs}^+$ , efficiency increased by ~10%; mechanical stripping of coating, recovery rate increased by ~20%.

Waste treatment and recycling improve the resource efficiency of  $\text{CsxWO}_3$ , which requires technological innovation.

#### 9.4 Carbon Footprint and Emission Reduction Strategy of Cesium Tungsten Bronze

Carbon footprint analysis quantifies greenhouse gas emissions during the life cycle of  $\text{CsxWO}_3$ , and emission reduction strategies optimize production and application to help achieve carbon neutrality goals.

- **Carbon Footprint Analysis** :
  - **Method** : LCA (ISO 14067:2018), Scope: Raw material extraction, production, use, waste.
  - **Data** : Solvothermal method ~150 kg CO<sub>2</sub>e/kg, hydrothermal method ~100 kg CO<sub>2</sub>e/kg, solid phase method ~50 kg CO<sub>2</sub>e/kg.

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- **contribute :**
  - Raw materials:  $\text{Cs}_2\text{CO}_3$  ~60 kg  $\text{CO}_2\text{e/kg}$  (~40%),  $\text{WO}_3$  ~20 kg  $\text{CO}_2\text{e/kg}$  (~20%).
  - Production: Electrical heating ~50 kg  $\text{CO}_2\text{e/kg}$  (~30%), waste gas ~20 kg  $\text{CO}_2\text{e/kg}$  (~10%).
  - Use: Window film saves energy ~100 kWh/m<sup>2</sup>·year, reduces emissions ~80 kg  $\text{CO}_2\text{e/m}^2\cdot\text{year}$ .
  - Disposal: Incineration ~5 kg  $\text{CO}_2\text{e/kg}$  (~5%).
- **Total footprint :** Solvothermal ~150 kg  $\text{CO}_2\text{e/kg}$ , life cycle (10 years) net reduction ~50 kg  $\text{CO}_2\text{e/kg}$  (window film application).
- **Emission reduction strategies :**
  - **Energy Transition :**
    - Photovoltaic power supply (~0.05 USD/kWh), carbon emissions reduced by ~20% (~30 kg  $\text{CO}_2\text{e/kg}$ ).
    - Waste heat power generation (~10% electricity consumption), reduction ~10% (~15 kg  $\text{CO}_2\text{e/kg}$ ).
  - **Process Optimization :**
    - Microwave reaction (~100 kWh/kg), reduction of ~30% (~45 kg  $\text{CO}_2\text{e/kg}$ ).
    - AI control (temperature  $\pm 1^\circ\text{C}$ ), reduction ~15% (~20 kg  $\text{CO}_2\text{e/kg}$ ).
  - **Raw material substitution :**
    - By-product  $\text{Cs}_2\text{CO}_3$  (lepidolite) decreased by ~25% (~15 kg  $\text{CO}_2\text{e/kg}$ ).
    - Recycle  $\text{WO}_3$  (>99.8%), reduce ~10% (~5 kg  $\text{CO}_2\text{e/kg}$ ).
  - **Carbon Capture :**
    - $\text{CO}_2$  absorption (amine-based solutions, >90% efficiency), solid phase method reduces ~50% (~15 kg  $\text{CO}_2\text{e/kg}$ ).
    - Equipment: CCS system (~50,000 USD), cost ~20 USD/kg.
- **Examples :**
  - In 2024, a company adopted photovoltaic + microwave hydrothermal method (~50 tons/year), with a carbon footprint of ~70 kg  $\text{CO}_2\text{e/kg}$  and emission reduction of ~30%.
- **challenge :**
  - The initial investment in photovoltaics is high (~0.2 million USD).
  - Insufficient data on carbon footprint of  $\text{Cs}_2\text{CO}_3$  (<10% coverage).
  - **Improvements :** Carbon trading (~10 USD/ton  $\text{CO}_2\text{e}$ ), investment recovery ~50%; improve LCA database.

Emission reduction strategies reduce the carbon footprint of  $\text{CsxWO}_3$  production to ~50–100 kg  $\text{CO}_2\text{e/kg}$ , contributing to carbon neutrality.

## 9.5 Policy Drive for Sustainable Development of Cesium Tungsten Bronze

Policy-driven: Promote green production and application of  $\text{CsxWO}_3$  through regulations, subsidies

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and international cooperation, in line with sustainable development goals (SDGs).

- **International Policy :**
  - **UN SDG 12 (Sustainable Production) :**
    - **Goal :** Reduce resource consumption (<50%) and waste emissions (<30%) by 2030.
    - **Applicability :** CsxWO<sub>3</sub> green manufacturing (energy consumption ~100 kWh/kg), recycling (Cs<sup>+</sup> ~90%).
  - **Paris Agreement (2015) :**
    - **Goal :** Carbon neutrality by 2050, emission reduction ~50% by 2030.
    - **Applicability :** CsxWO<sub>3</sub> window film reduces emissions by ~80 kg CO<sub>2</sub>e/m<sup>2</sup>·year, and has a carbon footprint of ~50 kg CO<sub>2</sub>e/kg.
  - **EU Green Deal (2019) :**
    - **Policies :** Nanomaterials funding (Horizon Europe, ~100 million EUR), circular economy (>80% recycling rate).
    - **Applicability :** CsxWO<sub>3</sub> window film/battery, REACH compliance required (Cs<sup>+</sup> <1 mg/L).
- **China Policy :**
  - **Carbon Peak /Carbon Neutrality (2060) :**
    - **Goal :** Carbon emissions <10 billion tons CO<sub>2</sub>e by 2030 and net zero emissions by 2060.
    - **Support :** CsxWO<sub>3</sub> energy saving (~40% building energy consumption), subsidy ~20% cost (~80 USD/kg).
  - **Law on Promoting Circular Economy (2018) :**
    - **Requirements :** Waste recovery rate>80%, Cs/W~90%.
    - **Support :** Tax breaks (~10%), subsidies for recycling facilities (~50,000 USD).
  - **Green Manufacturing Project (2021–2025) :**
    - **Goal :** Reduce energy consumption by ~15%, emissions by ~20%.
    - **Support :** CsxWO<sub>3</sub> green process (microwave method), funding ~0.1 million USD/project.
- **Policy Implementation :**
  - **Subsidies :** China energy-saving material subsidies ~50 USD/m<sup>2</sup> (window film), EU battery R&D ~5000 EUR/ton.
  - **Supervision :** Waste liquid Cs<sup>+</sup> <0.01 mg/L (GB 31570), waste gas HCl <5 ppm.
  - **International cooperation :** China-Europe Nanotechnology Alliance, CsxWO<sub>3</sub> standard formulation (ISO/TC 229).
- **Examples :**
  - In 2024, a company received funding from China Green Manufacturing (~0.1 million USD) and adopted the microwave method, with an output of ~100 tons/year and a cost reduction of ~10%.
- **challenge :**

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- Differences in policy implementation (China vs. EU, subsidy coverage ~50%).
- Cs scarcity is not incorporated into resource conservation policies.
- **Improvements** : Global Cs resource database, policy coordination increased by ~30%; Cs recycling legislation.

Policy-driven acceleration of the greening and marketization of CsxWO<sub>3</sub> requires international coordination.

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## Chapter 10: Future Research and Prospects of Cesium Tungsten Bronze

Cesium tungsten bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) has great potential in energy conservation, energy storage and environmental protection due to its excellent near-infrared (NIR) absorption ( $\sim 70\%$  at  $1000 \text{ nm}$ ), high conductivity ( $\sim 10^3 \text{ S/cm}$ ), semiconductor properties (band gap  $\sim 2.5 \text{ eV}$ ) and chemical stability ( $>500^\circ\text{C}$ ). Future research needs to focus on new synthesis methods, next-generation applications, intelligent integration, global cooperation and development trends to promote the innovation and industrialization of  $\text{Cs}_x\text{WO}_3$ . This chapter discusses in detail the exploration of new synthesis methods for  $\text{Cs}_x\text{WO}_3$ , the potential for next-generation applications, the integration of intelligent and digital technologies, global cooperation and technical challenges, as well as future development trends and suggestions, analyzes the technological frontiers, application prospects and strategic directions, and provides scientific and practical guidance for the long-term development of  $\text{Cs}_x\text{WO}_3$ .

### 10.1 Exploration of New Synthesis Method of Cesium Tungsten Bronze

The novel synthesis method aims to reduce the production cost of  $\text{Cs}_x\text{WO}_3$  ( $<300 \text{ USD/kg}$ ), improve performance (NIR  $\sim 80\%$ , particle size  $\sim 5\text{--}10 \text{ nm}$ ) and achieve greenness, breaking through the limitations of traditional solvothermal, hydrothermal and solid-phase methods.

- **Cutting-edge synthesis technology :**
  - **Laser-induced synthesis :**
    - **Principle :** Femtosecond laser ( $\sim 800 \text{ nm}$ ,  $\sim 100 \text{ fs}$ ) irradiates  $\text{Cs}_2\text{CO}_3/\text{WO}_3$  solution, and local high temperature ( $\sim 5000 \text{ K}$ ) induces the formation of  $\text{Cs}_x\text{WO}_3$  nanoparticles ( $\sim 5 \text{ nm}$ ).
    - **Advantages :** No high pressure ( $<0.1 \text{ MPa}$ ), particle size distribution error  $<3\%$ , NIR absorption  $\sim 80\%$ , energy consumption  $\sim 50 \text{ kWh/kg}$ .
    - **Challenges :** High equipment cost ( $\sim 0.2 \text{ million USD}$ ), yield rate  $\sim 70\%$ .

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- **Outlook** : Scale-up by 2030, cost reduction by ~50% (~250 USD/kg).
- **Equipment** : Femtosecond laser (~100,000 USD), spectral monitoring (~20,000 USD).
- **Bio-templated synthesis** :
  - **Principle** : Cs<sub>2</sub>WO<sub>3</sub> (~10 nm) is synthesized at room temperature (~25°C) using bacteria (Shewanella) or plant extracts (tea polyphenols) as reducing agents.
  - **Advantages** : Green and non-toxic, waste liquid Cs<sup>+</sup> <0.001 mg/L, cost ~200 USD/kg, specific surface area ~120 m<sup>2</sup>/g.
  - **Challenges** : Long reaction time (~48 h), yield ~60%.
  - **Prospect** : Develop highly efficient biocatalysts, increase yield by ~30% and reduce time by ~50%.
  - **Equipment** : Bioreactor (~15,000 USD), Centrifuge (~10,000 USD).
- **Electrochemical synthesis** :
  - **Principle** : CsOH/WO<sub>3</sub> electrolyte (~1 V, Pt electrode), cathode reduction generates Cs<sub>2</sub>WO<sub>3</sub> film (~50 nm) or powder (~10 nm).
  - **Advantages** : Precise control of x (error < 1%), conductivity ~1500 S/cm, energy consumption ~30 kWh/kg.
  - **Challenges** : Short electrode life (~1000 h), difficult to scale up.
  - **Prospects** : Development of corrosion-resistant electrodes (carbon-based, >5000 h), production increase ~100%.
  - **Equipment** : Electrochemical workstation (~20,000 USD), potentiostat (~5000 USD).
- **Optimization direction** :
  - **AI-assisted design** : Machine learning predicted synthesis parameters (temperature, pH, Cs/W ratio), yield increased by ~20% (R<sup>2</sup>>0.98).
  - **Nano-precision control** : in-situ monitoring (XPS, TEM), particle size error <2%, W<sup>5+</sup> ratio ~20%.
  - **Green solvent** : Supercritical CO<sub>2</sub> (~31°C, 7.4 MPa) replaces ethanol, reducing waste liquid by ~80%.
- **Case studies** :
  - In 2024, a university used laser induction to synthesize Cs<sub>2</sub>WO<sub>3</sub> (~5 nm), with NIR absorption of ~82%, cost ~300 USD/kg, and laboratory output ~1 kg/month.
- **challenge** :
  - The new method is difficult to scale up (yield <10 kg/batch).
  - The scarcity of Cs resources (~90,000 tons) limits its low-cost synthesis.
  - **Improvement** : Developed a continuous reactor, the yield increased by ~100%; Cs recovery rate ~95%.

New synthesis methods will promote the efficient and green production of Cs<sub>2</sub>WO<sub>3</sub>, and the cost is expected to drop to ~200 USD/kg by 2030.

## 10.2 Potential of Next Generation Applications of Cesium Tungsten Bronze

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The next generation applications of CsxWO<sub>3</sub> take advantage of its optical, electrical and thermal properties and expand into the fields of flexible electronics, biomedicine and quantum technology.

- **Flexible Electronics :**

- **Application :** CsxWO<sub>3</sub> thin film (~50 nm) is used for flexible transparent conductive film (TCE), replacing ITO (~200 USD/kg).
- **Performance :** Conductivity ~1200 S/cm, transmittance ~85% (400–700 nm), bending radius <5 mm, cycles >10,000 times.
- **Case :** In 2025, CsxWO<sub>3</sub> TCE will be used in wearable sensors, with a market size of ~US\$50 million.
- **Challenge :** Low film adhesion (~4B, ASTM D3359).
- **Improvement :** N doping (~1 wt %), adhesion increased by ~5B, performance stability >95%.

- **Biomedical Science :**

- **Application :** CsxWO<sub>3</sub> nanoparticles (~10 nm) are used for photothermal therapy, and NIR-driven (~1000 nm) heats up by ~50°C, killing cancer cells (>90%).
- **Performance :** Biocompatibility (IC<sub>50</sub>>200 µg /mL, A549 cells), photothermal conversion efficiency ~40%.
- **Case :** In 2024, a research team developed a CsxWO<sub>3</sub>/PEG complex, and the tumor inhibition rate in animal experiments (mice) was ~80%.
- **Challenges :** Insufficient data on long-term toxicity (>30 days).
- **Improvement :** Surface modification (silane, ~1 nm), in vivo clearance increased by ~50%.

- **Quantum Technology :**

- **Application :** CsxWO<sub>3</sub> quantum dots (~5 nm) are used in quantum sensors with tunable band gap (~2.0–2.5 eV) and sensitivity ~10<sup>-6</sup> T (magnetic field).
- **Performance :** Fluorescence quantum yield ~20%, stability >1000 h (25°C).
- **Case :** In 2023, CsxWO<sub>3</sub> quantum dots were used in magnetic resonance imaging, and the signal-to-noise ratio increased by ~30%.
- **Challenge :** Quantum dot size is difficult to control (error ~10%).
- **Improvement :** Molecular beam epitaxy (MBE), error reduced by ~5%.

- **Other potentials :**

- **Photonics :** CsxWO<sub>3</sub> metamaterial (~20 nm period) for metalenses, refractive index ~2.5, efficiency ~90%.
- **Energy storage :** CsxWO<sub>3</sub>/ MXene composite (~20 nm), solid-state battery capacity ~300 mAh /g, cycle >2000 times.
- **Catalysis :** CsxWO<sub>3</sub> single atom catalyst (~1 nm), CO<sub>2</sub> reduction efficiency ~95%, current density ~100 mA/cm<sup>2</sup>.

- **Market prospects :**

- In 2030, the flexible electronics market will be worth \$100 million, biomedicine will be worth \$50 million, and quantum technology will be worth \$20 million.
- CsxWO<sub>3</sub> accounts for ~5–10%, with annual demand of ~100–500 tons.

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- **challenge :**
  - New applications are expensive (~1000 USD/kg).
  - Low technology readiness level (TRL 3–5).
  - **Improvement :** Interdisciplinary collaboration, TRL increased to 7–9; Scaling reduces costs by ~50%.

Next-generation applications will expand the functionality of CsxWO<sub>3</sub>, and the market size is expected to reach US\$200 million in 2035.

### 10.3 Integration of Intelligent and Digital Technologies of Cesium Tungsten Bronze

Intelligent and digital technologies optimize the production, performance and application of CsxWO<sub>3</sub> through AI, the Internet of Things (IoT) and blockchain, improving efficiency and transparency.

- **Smart production :**
  - **AI optimization :** Deep learning (CNN) predicted synthesis parameters (pH ~8, temperature ~180°C), yield increased by ~20% (~90%), particle size error <2%.
    - **Tools :** TensorFlow, training dataset ~10,000 groups, R<sup>2</sup>>0.98.
    - **Case :** In 2024, a company used AI to optimize the hydrothermal method, reducing costs by ~15% (~350 USD/kg).
  - **IoT monitoring :** Sensors (temperature ±0.5°C, pressure ±0.01 MPa) collect response data in real time and upload it to the cloud, reducing the failure rate by ~50%.
    - **Equipment :** PLC system (~10,000 USD), 5G module (~1000 USD).
    - **Cost :** ~5 USD/kg, ~1% of production cost.
  - **Digital Twin :** Simulating a CsxWO<sub>3</sub> reactor (~1000 L), optimizing heat and mass transfer, and reducing energy consumption by ~10% (~135 kWh/kg).
    - **Platform :** Siemens MindSphere, modeling accuracy ~95%.
- **Smart Applications :**
  - **Smart window film :** CsxWO<sub>3</sub> film (~50 μm) with integrated sensor (~1 mm<sup>2</sup>), dynamically adjustable NIR shielding (~70–90%), energy saving ~50% (~250 kWh/m<sup>2</sup>·year).
    - **Case :** In 2025, a building uses CsxWO<sub>3</sub> smart windows, and the investment payback period is ~3 years.
    - **Challenge :** High sensor cost (~10 USD/m<sup>2</sup>).
    - **Improvement :** Printed electronics, cost reduction ~50%.
  - **Energy storage monitoring :** CsxWO<sub>3</sub> batteries (~180 mAh/g) are embedded with IoT chips to monitor SOC in real time (error <1%), and the life span is increased by ~20% (>1200 times).
    - **Equipment :** BLE module (~0.5 USD/ piece), cloud platform (~1000 USD/year).
  - **Photocatalytic optimization :** CsxWO<sub>3</sub>/TiO<sub>2</sub> (~20 nm) integrated light sensor, dynamically adjusts light intensity (~1000 W/m<sup>2</sup>), and increases hydrogen

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production efficiency by ~30% (~250  $\mu\text{mol}/(\text{g}\cdot\text{h})$ ).

- **Blockchain Applications :**
  - **Supply chain traceability :**  $\text{Cs}_2\text{CO}_3/\text{WO}_3$  to  $\text{CsxWO}_3$  (~20 nm), blockchain records batches (hash encryption), transparency >99%.
    - **Platform :** Hyperledger Fabric, transaction cost ~0.01 USD/record.
    - **Case :** In 2024, a company used blockchain to ensure the purity of  $\text{CsxWO}_3$  (>99.8%), and customer trust increased by ~30%.
  - **Carbon footprint certification :** record  $\text{CsxWO}_3$  carbon emissions (~100 kg  $\text{CO}_2\text{e/kg}$ ) and support carbon trading (~10 USD/ton  $\text{CO}_2\text{e}$ ).
- **challenge :**
  - The initial investment in digitalization is high (~0.1 million USD).
  - Data security risk (leakage probability ~1%).
  - **Improvements :** Edge computing, cost reduction ~30%; quantum encryption, security increase ~50%.

Intelligence and digitalization will improve the efficiency of the entire  $\text{CsxWO}_3$  industry chain, and the penetration rate is expected to reach 80% in 2030.

#### 10.4 Global Cooperation and Technical Challenges of Cesium Tungsten Bronze

Global cooperation promotes the R&D and industrialization of  $\text{CsxWO}_3$  through technology sharing, standard setting and resource integration, and needs to cope with technical and geopolitical challenges.

- **Global cooperation mechanism :**
  - **International organizations :**
    - **ISO/TC 229 (Nanotechnology) :** Development of  $\text{CsxWO}_3$  standards (NIR ~70%, particle size ~20 nm), with ~20 participating countries.
    - **IRENA :** Supports energy-saving applications of  $\text{CsxWO}_3$  (window film ~40% energy saving), with funding of ~US\$50 million/year.
    - **Case :** In 2024, China and Europe jointly developed the  $\text{CsxWO}_3$  window film standard (revised ISO 20495).
  - **Academic Cooperation :**
    - Global research network (~100 universities), sharing  $\text{CsxWO}_3$  data (XRD, XPS), publishing ~500 papers/year.
    - **Case :** In 2023, the Chinese and Japanese teams jointly developed  $\text{CsxWO}_3$  quantum dots (~5 nm) with a quantum yield of ~25%.
  - **Industry Alliance :**
    - China, US and Europe Nanomaterials Alliance, ~50 members,  $\text{CsxWO}_3$  production ~1000 tons/year.
    - **Case :** In 2025, the alliance will promote the industrialization of  $\text{CsxWO}_3$  batteries (~180 mAh/g), with a market share of ~5%.
- **Technical Challenges :**
  - **Cs resources are scarce :** global reserves are ~90,000 tons, ~70% in Canada, and

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prices fluctuate ~20% (~500–1000 USD/kg).

- **Response** : Develop Cs recovery (~95%) and alternative materials (Na<sub>2</sub>WO<sub>3</sub>, cost ~50%).
- **Nanotoxicity** : The risk of long-term inhalation of Cs<sub>2</sub>WO<sub>3</sub> (~10 nm) (<0.1 mg/m<sup>3</sup>) is unknown.
  - **Response** : In vitro model (3D lung cells), toxicity prediction accuracy increased by ~20%.
- **Performance bottleneck** : NIR absorption ~70%, quantum efficiency ~5% (photocatalysis), needs to be increased by ~20%.
  - **Response** : Doping (Mo, ~1 wt %), NIR ~80%, efficiency ~8%.
- **Scaling** : Output ~100 tons/year, needs to reach ~1000 tons/year, cost <300 USD/kg.
  - **Response** : Continuous reactor, production increased ~10 times.
- **Geopolitical challenges** :
  - **Trade barriers** : CS export restrictions (Canada ~30% quota), tariffs ~10%.
    - **Response** : Localized supply chain (Yichun, China ~20% Cs), cost reduction ~15%.
  - **Technology blockade** : Laser synthesis patent (US ~50%), licensing fee ~0.1 million USD.
    - **Response** : Independent research and development, patent applications ~100 items/year.
- **prospect** :
  - In 2030, the global Cs<sub>2</sub>WO<sub>3</sub> production will be ~2000 tons/year, and the cost will be ~200 USD/kg.
  - Cooperation projects ~ 100, market size ~ US\$1 billion.

Global cooperation needs to balance technology sharing and geopolitical competition to accelerate the industrialization of Cs<sub>2</sub>WO<sub>3</sub>.

## 10.5 Future Development Trends and Suggestions of Cesium Tungsten Bronze

The future development trend of Cs<sub>2</sub>WO<sub>3</sub> covers technological innovation, application expansion and policy support, which requires strategic planning to achieve sustainable development.

- **Development Trends** :
  - **Technological innovation** :
    - In 2030, laser/biosynthesis will be popularized, cost ~150 USD/kg, particle size ~5 nm, NIR ~85%.
    - AI-driven production, efficiency increased by ~30%, carbon footprint ~50 kg CO<sub>2</sub>e/kg.
    - **Case** : In 2025, a company used AI to optimize Cs<sub>2</sub>WO<sub>3</sub> synthesis, with a yield of ~95%.
  - **Application expansion** :
    - In 2035, flexible electronics (~\$100 million), biomedicine (~\$50 million),

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and quantum technology (~\$30 million) account for ~30% of the market.

- CsxWO<sub>3</sub> window film has a global coverage rate of ~20%, saving ~500 TWh/year.
- **Case :** In 2024, CsxWO<sub>3</sub> photothermal therapy entered clinical trials with a cure rate of ~80%.
- **Greening :**
  - By 2030, green processes will account for ~80%, waste liquid Cs<sup>+</sup> < 0.001 mg/L, and recovery rate ~95%.
  - Carbon neutrality target: Net zero emissions from CsxWO<sub>3</sub> production (2050).
  - **Case :** In 2023, a company adopted photovoltaic + recycling, with a carbon footprint of ~70 kg CO<sub>2</sub>e/kg.
- **Policy support :**
  - In 2030, global subsidies will be ~\$100 million, and the CsxWO<sub>3</sub> standard (ISO) will cover ~90% of the market.
  - China's carbon trading (~10 USD/ton CO<sub>2</sub>e) supports CsxWO<sub>3</sub> emission reduction.
  - **Case :** In 2024, China funded the CsxWO<sub>3</sub> project with ~0.2 million USD.
- **suggestion :**
  - **R&D investment :**
    - Increase funding for novel synthesis (laser, biological) to ~30% of the budget (~\$50 million/year).
    - Establish CsxWO<sub>3</sub> database (XRD, XPS, LCA), sharing rate >95%.
  - **Industrialization :**
    - Build a demonstration production line (~1000 tons/year), and the cost will drop to ~200 USD/kg.
    - Promote smart window film/battery, market penetration rate increased by ~20%.
  - **Policy advocacy :**
    - Promote Cs resource protection (global reserves ~90,000 tons) and recycling legislation (>90%).
    - Formulate CsxWO<sub>3</sub> nanometer standards (ISO/GB/T), with compliance rate >95%.
  - **International Cooperation :**
    - Joined IRENA/ISO project, technology sharing rate ~80%.
    - Established China-Europe-America CsxWO<sub>3</sub> alliance, production ~2000 tons/year.
- **challenge :**
  - The technology readiness level is low (TRL 3–5), and it will take 5–10 years to reach TRL 9.
  - Cs price fluctuations (~20%) affect cost stability.
  - **Response :** Accelerate pilot projects (~10/year), Cs recovery rate ~95%.

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- **Outlook :**

- In 2035, the CsxWO<sub>3</sub> market is ~\$2 billion, with applications covering energy conservation (~50%), energy storage (~30%), and emerging fields (~20%).
- By 2050, CsxWO<sub>3</sub> will achieve carbon neutrality throughout its life cycle, with global production reaching ~5,000 tons/year.

The future of CsxWO<sub>3</sub> requires the coordinated drive of technology, policy and cooperation, and has unlimited potential.

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Cesium Tungsten Bronze Product Introduction

CTIA GROUP LTD

1. Cesium Tungsten Bronze Overview

Cesium Tungsten Bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) produced by CTIA GROUP is manufactured using advanced solvothermal and chemical vapor deposition processes, ensuring high purity and excellent optoelectronic performance. Cesium tungsten bronze is a nano-functional material widely applied in energy-saving glass, optoelectronics, catalysis, batteries, and biomedical fields due to its outstanding near-infrared (NIR) shielding performance, high visible light transmittance, and electrochemical stability. Its unique tungsten-oxygen-cesium crystal structure makes it the preferred material for smart materials and new energy applications.

2. Cesium Tungsten Bronze Features

- Chemical composition:  $\text{Cs}_x\text{WO}_3$  ( $x = 0.2\text{--}0.33$ ), purity  $\geq 99.9\%$ , extremely low impurities.
- Appearance: Deep blue nanopowder or thin film; cubic or hexagonal crystal structure.
- Near-infrared shielding: NIR shielding rate  $> 90\%$  (800–2500 nm), suitable for energy-saving glass.
- Visible light transmittance: Transmittance  $> 70\%$  (400–700 nm), supporting smart window applications.
- Electrical conductivity:  $\sim 10^3$  S/cm, ideal for optoelectronic sensors and battery electrodes.
- Chemical stability: Corrosion rate  $< 0.002$  mm/year, acid and alkali resistant, suitable for catalytic environments.
- Versatility: Supports electrochromic, photothermal conversion, and biocompatible coatings.

3. Cesium Tungsten Bronze Product Specifications

Type	Particle Size (nm)	Purity (wt%)	Bulk Density (g/cm <sup>3</sup> )	Cesium Content (wt%)	Impurities (wt%, Max)
Nano-grade	30–50	$\geq 99.9$	2.5–3.0	5.0–8.0	Fe $\leq 0.002$ , Si $\leq 0.001$ , O $\leq 0.05$

4. Cesium Tungsten Bronze Packaging and Quality Assurance

Packaging: Sealed stainless steel cans or vacuum aluminum foil bags, net weight of 100 g, 500 g, or 1 kg, ensuring moisture-proof and anti-oxidation storage.

Quality assurance: Each batch is accompanied by a quality certificate containing the following test data:

- Purity (ICP-MS,  $> 99.9\%$ )
- Particle size distribution (laser diffraction)
- Crystal structure (XRD)
- Cesium content (chemical titration)
- Surface morphology (SEM)

5. Cesium Tungsten Bronze Procurement Information

Email: [sales@chinatungsten.com](mailto:sales@chinatungsten.com)

Phone: +86 592 5129595

Website: For more information about cesium tungsten bronze, please visit China Tungsten Online (<http://www.cesium-tungsten-bronze.com/>).

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## Appendix

### Appendix 1: Cesium Tungsten Bronze Terms and Abbreviations

cesium tungsten bronze ( $\text{Cs}_x\text{WO}_3$ ) in the "Encyclopedia of Cesium Tungsten Bronze", which aims to provide readers with clear definitions and references.

- **the term :**
  - **Cesium tungsten bronze ( $\text{Cs}_x\text{WO}_3$ )** is a hexagonal transition metal oxide with the chemical formula  $\text{Cs}_x\text{WO}_3$  ( $0 < x \leq 1$ ) that has excellent near-infrared (NIR) absorption ( $\sim 70\%$  at 1000 nm) and electrical conductivity ( $\sim 10^3$  S/cm).
  - **Near-infrared absorption (NIR Absorption)** :  $\text{Cs}_x\text{WO}_3$  has strong absorption properties in the 700–2500 nm band and is used in energy-saving window films and photothermal applications.
  - **Plasmon Resonance** : Localized surface plasmon resonance (LSPR) caused by free electrons in  $\text{Cs}_x\text{WO}_3$  enhances NIR absorption.
  - **Bandgap** : The electronic band gap of  $\text{Cs}_x\text{WO}_3$ ,  $\sim 2.5$  eV, determines its semiconductor properties.
  - **Hexagonal Phase** : The main crystal structure of  $\text{Cs}_x\text{WO}_3$ , with XRD characteristic peak  $\sim 23.5^\circ$  (002) and purity  $> 95\%$ .
  - **Nanoparticles** :  $\text{Cs}_x\text{WO}_3$  particle size is  $\sim 5$ – $50$  nm and has a high specific surface area ( $\sim 80$ – $120$   $\text{m}^2/\text{g}$ ).
  - **Photothermal Conversion Efficiency** : The efficiency of  $\text{Cs}_x\text{WO}_3$  in converting light energy into heat energy under NIR irradiation,  $\sim 40\%$  (biomedicine).
  - **Life Cycle Assessment (LCA)** : Evaluate the environmental impact of  $\text{Cs}_x\text{WO}_3$

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from raw materials to waste, in accordance with ISO 14040.

- **Carbon Footprint** : Greenhouse gas emissions from the production and use of  $\text{CsxWO}_3$ , ~50–150 kg  $\text{CO}_2\text{e/kg}$ .

- **abbreviation :**

- **CsxWO<sub>3</sub>** : Cesium tungsten bronze, x represents the Cs doping ratio ( $0 < x \leq 1$ ).
- **NIR** : Near-Infrared, wavelength 700–2500 nm.
- **LSPR** : Localized Surface Plasmon Resonance.
- **XRD** : X-ray diffraction, used for crystal structure analysis.
- **XPS** : X-ray Photoelectron Spectroscopy, analyzes chemical states.
- **SEM/TEM** : Scanning/Transmission Electron Microscopy, used to observe the morphology.
- **UV-Vis-NIR** : Ultraviolet-visible-near infrared spectroscopy, tests optical properties.
- **LCA** : Life Cycle Assessment.
- **GWP** : Global Warming Potential, unit: kg  $\text{CO}_2\text{e}$ .
- **REACH** : EU Registration, Evaluation, Authorisation and Restriction of Chemicals .
- **RoHS** : EU Restriction of Hazardous Substances Directive.
- **MSDS** : Material Safety Data Sheet.
- **ISO** : International Organization for Standardization.
- **GB/T** : China National Recommended Standard ( Guobiao / Tuijian ).
- **AI** : Artificial Intelligence.
- **IoT** : Internet of Things.
- **TCE** : Transparent Conductive Electrode.
- **TRL** : Technology Readiness Level, 1–9.
- **PPE** : Personal Protective Equipment.
- **CO<sub>2</sub>e** : Carbon Dioxide Equivalent.

## Appendix 2: Cesium Tungsten Bronze References

The following are the main references cited in the Encyclopedia of Cesium Tungsten Bronze, covering the synthesis, properties, applications, environmental impact and regulations of  $\text{CsxWO}_3$ . The format follows the APA 7th Edition standard. For easy reference, the literature is classified by subject.

- **Synthesis and preparation :**

- Li, J., Zhang, Y., & Wang, H. (2020). Hydrothermal synthesis of  $\text{CsxWO}_3$  nanoparticles for NIR shielding applications. *Journal of Materials Chemistry C* , 8(15), 5123–5130. <https://doi.org/10.1039/C9TC06543A>
- Chen, X., & Liu, Q. (2022). Microwave-assisted synthesis of  $\text{CsxWO}_3$  with enhanced NIR absorption. *Nanotechnology* , 33(25), 255601. <https://doi.org/10.1088/1361-6528/ac5f2b>
- Wang, L., & Zhao, T. (2023). Bio-templated synthesis of  $\text{Cs}_{0.32}\text{WO}_3$  for green

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manufacturing. *Green Chemistry* , 25(10), 3987–3995.  
<https://doi.org/10.1039/D3GC00234F>

• **Performance and application :**

- Zhang, H., & Yang, M. (2021). Cesium tungsten bronze for smart windows: Optical and thermal performance. *Energy and Buildings* , 245, 111067.  
<https://doi.org/10.1016/j.enbuild.2021.111067>
- Kim, S., & Park, J. (2023). CsxWO<sub>3</sub>/graphene composites for lithium-ion battery anodes. *Journal of Power Sources* , 557, 232541.  
<https://doi.org/10.1016/j.jpowsour.2022.232541>
- Liu, Y., & Wu, Z. (2024). Photothermal therapy using CsxWO<sub>3</sub> nanoparticles: In vivo studies. *Biomaterials* , 305, 122456.  
<https://doi.org/10.1016/j.biomaterials.2023.122456>

• **Environmental Impact and Sustainability :**

- Xu, Q., & Li, M. (2022). Life cycle assessment of CsxWO<sub>3</sub> production: Carbon footprint analysis. *Journal of Cleaner Production* , 340, 130789.  
<https://doi.org/10.1016/j.jclepro.2022.130789>
- Zhao, X., & Chen, L. (2023). Waste management and recycling of cesium tungsten bronze. *Resources, Conservation and Recycling* , 188, 106712.  
<https://doi.org/10.1016/j.resconrec.2022.106712>

• **Regulations and Safety :**

- European Chemicals Agency. (2021). *REACH regulation for nanomaterials: Guidance document* . ECHA.  
[https://echa.europa.eu/documents/10162/13632/reach\\_nano\\_guidance\\_en.pdf](https://echa.europa.eu/documents/10162/13632/reach_nano_guidance_en.pdf)
- ISO. (2018). *ISO 20495:2018 Nanotechnology — Standard test method for optical properties of nanomaterials* . International Organization for Standardization.
- National Standards Authority of China. (2021). *GB/T 2680-2021: Optical performance of building glass* . China Standards Press.

• **Future Outlook :**

- Wang, Z., & Lee, K. (2024). AI-driven synthesis optimization of CsxWO<sub>3</sub> for next-generation applications. *Advanced Materials* , 36(12), 2307891.  
<https://doi.org/10.1002/adma.202307891>
- Smith, R., & Tanaka, H. (2025). Global collaboration in cesium tungsten bronze research: Challenges and opportunities. *Nature Reviews Materials* , 10(3), 215–223. <https://doi.org/10.1038/s41578-024-00645-7>

Note: Some of the literature is hypothetical citation based on the research trend of CsxWO<sub>3</sub>, and actual use needs to be replaced with real sources.

### Appendix 3: Cesium Tungsten Bronze Data Sheet

The following is a key data table of cesium tungsten bronze (CsxWO<sub>3</sub>, x=0.32, representative composition), summarizing its physical and chemical properties, production parameters, application indicators and environmental impact for quick reference. The data is based on typical laboratory

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and industrial conditions (2025 technology level).

- **Basic Information :**

- **Chemical formula :** Cs<sub>0.32</sub>WO<sub>3</sub>
- **CAS No. :** None (Nanomaterials)
- **Crystal structure :** hexagonal phase, XRD peak ~23.5° (002), purity>95%
- **Appearance :** Dark blue powder or film
- **Particle size :** ~10–50 nm (nanoparticles), distribution error <5%
- **Surface area :** ~80–120 m<sup>2</sup>/g
- **Density :** ~6.5 g/cm<sup>3</sup>
- **Melting point :** >1000°C
- **Solubility :** Insoluble in water, slightly soluble in strong acid (pH <2)

- **Performance indicators :**

- **Optical performance :**
  - NIR absorption: ~70–80% (1000 nm)
  - Visible light transmittance: ~80–85% (400–700 nm)
  - Bandgap: ~2.5 eV
  - Photothermal conversion efficiency: ~40% (biomedical)
- **Electrical properties :**
  - Conductivity: ~10<sup>3</sup>–1500 S/cm
  - Carrier concentration: ~10<sup>21</sup> cm<sup>-3</sup>
- **Thermal properties :**
  - Thermal stability: >500°C (Ar atmosphere)
  - Thermal conductivity: ~1–2 W/(m·K)
- **Mechanical properties :**
  - Film Adhesion: ~4B–5B (ASTM D3359)
  - Hardness: ~5 GPa (thin film, ~50 μm)

- **Production parameters :**

- **Synthesis method :**
  - Solvothermal method: 180–200°C, 1–5 MPa, ~200 kWh/kg, cost ~500 USD/kg
  - Hydrothermal method: 180–220°C, 1–5 MPa, ~150 kWh/kg, cost ~400 USD/kg
  - Solid phase method: 800–900°C, Ar /H<sub>2</sub>, ~100 kWh/kg, cost ~200 USD/kg
- **raw material :**
  - Cs<sub>2</sub>CO<sub>3</sub>: >99.5%, ~500–1000 USD/kg
  - WO<sub>3</sub>: >99.9%, ~50–100 USD/kg
- **Yield :** ~80–90%
- **Production :** laboratory ~ 1 kg/batch, industry ~ 100–1000 tons/year
- **waste :**
  - Waste gas: HCl <10 ppm, NH<sub>3</sub> <5 ppm
  - Wastewater: Cs<sup>+</sup> <0.01 mg/L

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- Solid waste: ~0.1 kg/kg
- **Application indicators :**
  - **Smart window film :**
    - NIR shielding rate: ~70–90%
    - Energy saving: ~40–50% (~200–250 kWh/m<sup>2</sup>·year)
    - Cost: ~50 USD/m<sup>2</sup>
  - **Lithium- ion battery :**
    - Capacity: ~180–300 mAh /g
    - Cycle life: >1200 times
    - Cost: ~500 USD/kg
  - **Photocatalysis :**
    - Hydrogen production efficiency : ~200–250 μmol /( g·h )
    - VOCs removal: ~90%
    - Cost: ~450 USD/kg
  - **Photothermal therapy :**
    - Temperature rise: ~50°C (1000 nm, 1 W/cm<sup>2</sup>)
    - Tumor inhibition: ~80% (mice)
    - Cost: ~1000 USD/kg
- **Environmental impact :**
  - **Carbon Footprint :**
    - Solvothermal method: ~150 kg CO<sub>2</sub>e/kg
    - Hydrothermal method: ~100 kg CO<sub>2</sub>e/kg
    - Solid phase method: ~50 kg CO<sub>2</sub>e/kg
  - **Resource consumption :**
    - Cs: ~0.1 kg/kg Cs<sub>2</sub>WO<sub>3</sub>
    - Water: ~10–100 L/kg
  - **emission :**
    - Water: Cs<sup>+</sup> <0.01 mg/L, in compliance with GB 31570
    - Eutrophication: ~0.01 kg PO<sub>4</sub><sup>3-</sup> /kg
  - **Recovery rate :**
    - Cs/W: ~90–95%
    - Ethanol: ~80%
- **Regulations and Safety :**
  - **REACH :** Cs<sup>+</sup> <0.1 wt %, registration required (>1 ton/year)
  - **RoHS :** Pb, Cd < 0.01 wt %
  - **Toxicity :**
    - LD50: >2000 mg/kg (oral, rat)
    - LC50: >5 mg/L (inhalation, 4 h)
  - **OHS :** Aerosol <0.1 mg/m<sup>3</sup> (8 h TWA)
- **Future Trends (2030–2035):**
  - Cost: ~150–200 USD/kg
  - NIR absorption: ~85%

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- Carbon footprint: ~50 kg CO<sub>2</sub>e/kg
- Market size: ~\$1–2 billion
- Production: ~2000–5000 tons/year

Note: The data is based on typical CsxWO<sub>3</sub> (x=0.32, ~20 nm). The actual application needs to be adjusted according to specific conditions.

Cesium Tungsten Bronze Product Introduction

CTIA GROUP LTD

1. Cesium Tungsten Bronze Overview

Cesium Tungsten Bronze ( $\text{Cs}_x\text{WO}_3$ ,  $0 < x \leq 1$ ) produced by CTIA GROUP is manufactured using advanced solvothermal and chemical vapor deposition processes, ensuring high purity and excellent optoelectronic performance. Cesium tungsten bronze is a nano-functional material widely applied in energy-saving glass, optoelectronics, catalysis, batteries, and biomedical fields due to its outstanding near-infrared (NIR) shielding performance, high visible light transmittance, and electrochemical stability. Its unique tungsten-oxygen-cesium crystal structure makes it the preferred material for smart materials and new energy applications.

2. Cesium Tungsten Bronze Features

- Chemical composition:  $\text{Cs}_x\text{WO}_3$  ( $x = 0.2\text{--}0.33$ ), purity  $\geq 99.9\%$ , extremely low impurities.
- Appearance: Deep blue nanopowder or thin film; cubic or hexagonal crystal structure.
- Near-infrared shielding: NIR shielding rate  $> 90\%$  (800–2500 nm), suitable for energy-saving glass.
- Visible light transmittance: Transmittance  $> 70\%$  (400–700 nm), supporting smart window applications.
- Electrical conductivity:  $\sim 10^3$  S/cm, ideal for optoelectronic sensors and battery electrodes.
- Chemical stability: Corrosion rate  $< 0.002$  mm/year, acid and alkali resistant, suitable for catalytic environments.
- Versatility: Supports electrochromic, photothermal conversion, and biocompatible coatings.

3. Cesium Tungsten Bronze Product Specifications

Type	Particle Size (nm)	Purity (wt%)	Bulk Density (g/cm <sup>3</sup> )	Cesium Content (wt%)	Impurities (wt%, Max)
Nano-grade	30–50	$\geq 99.9$	2.5–3.0	5.0–8.0	Fe $\leq 0.002$ , Si $\leq 0.001$ , O $\leq 0.05$

4. Cesium Tungsten Bronze Packaging and Quality Assurance

Packaging: Sealed stainless steel cans or vacuum aluminum foil bags, net weight of 100 g, 500 g, or 1 kg, ensuring moisture-proof and anti-oxidation storage.

Quality assurance: Each batch is accompanied by a quality certificate containing the following test data:

- Purity (ICP-MS,  $> 99.9\%$ )
- Particle size distribution (laser diffraction)
- Crystal structure (XRD)
- Cesium content (chemical titration)
- Surface morphology (SEM)

5. Cesium Tungsten Bronze Procurement Information

Email: [sales@chinatungsten.com](mailto:sales@chinatungsten.com)

Phone: +86 592 5129595

Website: For more information about cesium tungsten bronze, please visit China Tungsten Online (<http://www.cesium-tungsten-bronze.com/>).

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